



REVIEW ARTICLE

Coumarin derivatives as promising antibacterial agent(s)



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Abstract Nowadays, bacterial infections epitomize significant health threats globally with an increased morbidity and mortality. Most contemporary antibacterial agents are resisted by pathogenic bacteria - the multidrug resistant (MDR) bacterial strains arising from cross resistances operative in natural bacterial consortia inside human body and in environments. Consequently, the development of newer potential drug candidate(s) is required against the broad spectrum of MDR bacteria. Indeed, the phytochemical coumarin and its derivatives had been reported with broad biological inhibitory properties, including antibacterial activities. In this review, several methods of synthetic strategies of coumarin derivatives as antibacterials were considered with individual schematic compounds by structure-activity relationship (SAR) studies as essential corollaries. Overall, substituents at positions C-3 and C-4 of coumarin are coveted for the development of newer antibacterial agents.

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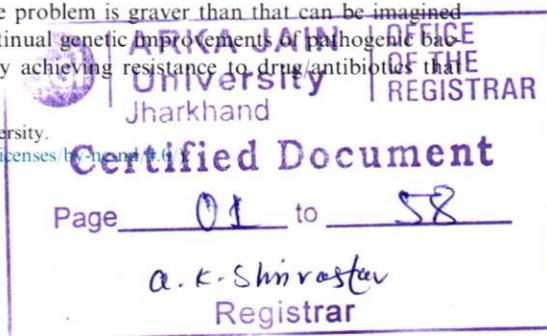
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1. Introduction

Today resistance patterns of pathogenic bacteria to frequently used drugs and antibiotics have become the commonplace incidence, creating annoyances in clinics and havoc in public health worldwide. The problem is graver than that can be imagined because of continual genetic improvements of pathogenic bacterial strains by achieving resistance to drug/antibiotics that



were never treated often; unfortunately, with any new generation of antibiotics or newly introduced drugs newer resistant bacteria evolve persistently. The inherent/intrinsic bacterial genetic exchange mechanisms facilitate bacterial consortia for gaining characters from genetically similar as well as, distant bacteria. Approximately 17 million people are dying each year from infectious diseases, and approximately, 50,000 people in all age group are affected (Holmes et al., 2017; Laxminarayan et al., 2006). Thus, the use of antibiotics rapidly loose effectiveness; thereby a vacuum is created in need/source of coveted antibacterial. Consequently, the development of new antibacterial drug candidate(s) remains as the call of the day. Increasing incidences of microbial infection by the development of microbial resistance of the most antibiotics through either genome of microbial mutations or an evolved the mechanism of resistance of action which is major health problem.

The molecular manipulations of promising two lead compounds with same biological actions are a rational approach to design and develop newer drug(s); it involves an effort to combine two different pharmacophore groups of similar activity into a hybrid lead candidate. Natural products particularly phytochemicals such as, vanillin, thymol, eugenol, menthol, umbelliferon, carvacrol, curcumin, and a few more are being used for several therapeutic purposes namely, antibacterial, antifungal and anticancer agents through main stream medicinal chemistry approaches (Baral et al., 2019; Sahoo et al., 2020a, 2020b). Moreover, coumarin (a phytochemical) is chemically the benz[α]pyrone and freely occurring as constituents or could be condensed with carbohydrate said to be glycosides (Fig. 1). It is a fused ring system between benzene and lactone known as 'pyrone' and structurally resembles to chromone; but the difference in both the positions of carbonyl or ketone system present in individual structures (Jain and Joshi, 2012). The carboxamide coumarin derivative i.e., 'Novobiocin' (Kasperkiewicz et al., 2020) and Chlorobiocin, Aminocoumarin and a few more that are commercially recommended antibiotics. Consequently, these derivatives are being reported as anti-microbial, anti-oxidant, anti-cancer, anti-HIV, anti-diabetic and anti-viral agents (Detsi et al., 2017). Herein, on basis of rationalisation of synthetic strategies of coumarin derivatives and their candidate's potency against several pathogenic bacterial strains were studied with structure-activity relationship (SAR) studies. Moreover, the number of significant antibacterial coumarin candidates had been developed by attaching several substituents as functionality group, incorporation of new scaffolds and coordinate of metal ion complexes in different positions of coumarin. Similarly, norharmane, a natural bioactive compound and nostocine, a phyco-constituent had lent themselves to structural modifications at several positions with hopes to develop newer anti-cancer drug candidates (Sahoo et al., 2019a, 2019b).

This aim of this review is to have a concise account and detailed highlights of structural derivatives of coumarin with individually associated schematic strategies; by the by, to locate candidate(s) with significant antibacterial potency (Fig. 2). This would be the countenance to diverse groups of chemists, biologists and drug developers, to distinguish and to identify promising structures to be judged for further promotion in the development of newer therapeutic or antibacterial agent(s). Indeed, to locate a novel/curious molecule and its derivatives for treating various malicious infectious diseases remain as the obsessive quest in pharmacology. The involve-

ment of principles of medicinal chemistry in modifying a phytochemical would be an eclectic approach for possible use as future antibacterials, in the face of MDR bacteria is a step against the general trend of floccinaucinihilipification against phytocompounds as drug(s) (see Schemes 1-80).

1.1. Sources of coumarin

Coumarin was first isolated from a higher flowering plant, *Dipteryx odorata* (known as tonka beans) in 1820 and seeds of this plant contain aromatic organic bioactive compound coumarin. Concomitantly, other several sources have been located in (Apiaceae, Asteraceae, Apocynaceae, Rutaceae, Calophyllaceae, Fabaceae and a few more- family) (Table 1).

2. Synthesis and antibacterial activities of coumarin derivatives

2.1. Synthesis of 2-quinoxalone-coumarin hydrazone derivatives

A series of quinoxalone coumarin hydrazone derivatives **3a-3d** were prepared by the microwave synthetic procedure and the obtained congeners had been evaluated for antimicrobial actions. These compounds were prepared by the condensation of equimolar concentration of precursor, 3-hydrazonequinoxalone **1** with substituted 3-acetylcoumarins, **2a-2d** in dry DMF, microwave-oven at 400 MW. The developed product was recrystallized by ethanol. The starting hydrazine quinoxalone was prepared by the cyclisation reaction of ortho phenylenediamine (OPD) with oxalic acid in the presence of a mineral acid, which further was reacted with hydrated hydrazine. Overall, SAR studies of these derivatives indicated that the attachment of halogen group to the phenyl ring of coumarin nucleus might have been reported for notable effect on antibacterial actions; the chloro substituted coumarin compound (*E*)-3-(2-(1-(6-chloro-2-oxo-2*H*-chromen-3-yl)ethylidene)hydrazinyl)quinoxalin-2(1*H*)-one **3** had shown inhibition with *E. coli* and *S. aureus* at 30 and 32 mm as zone of inhibition (ZOI) in comparison to Streptomycin (Ajani et al., 2010).

2.2. Synthesis of 4-azidomethyl-7-methyl-coumarin bearing sulfonamide

A series of 4-azidomethyl coumarin bearing sulfonamide derivatives **4a-4e** had been synthesized with four different steps. Initially, the cresol reacted with ethyl acetoacetate in cyclizing agent to produce 7-methyl-4-bromomethyl coumarin **1**, and the obtained compound **1** had undergone chlorosulfonation, yielding the corresponding product, 7-methyl-4-bromomethyl coumarinyl 6-sulfonyl chloride **2**, which was reacted with sodium azide in acetone yielding 4-azidomethyl coumarin **3**; and finally these intermediates in amination with substituted anilines produce 4-(azidomethyl)-7-methyl-2-oxo-*N*-subst.phenyl-2*H*-chromene-6-sulfonamide (**4a-4e**). From SAR studies of these compounds it was clear that the presence of methoxy and nitro-group in benzene sulfonamide residue attached at C-7 position of 4-azidomethyl coumarin, which may produce more significant antibacterial actions. These compounds with methoxy substituents as attached at in either ortho or para of phenyl ring had exhibited control of *Enterococcus faecalis* at the MIC value 1 μ g/mL in comparison to

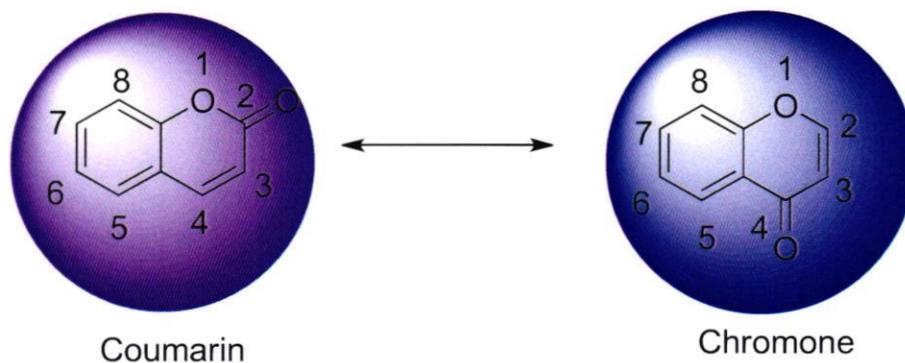


Fig. 1 Structure of chromene nucleus.

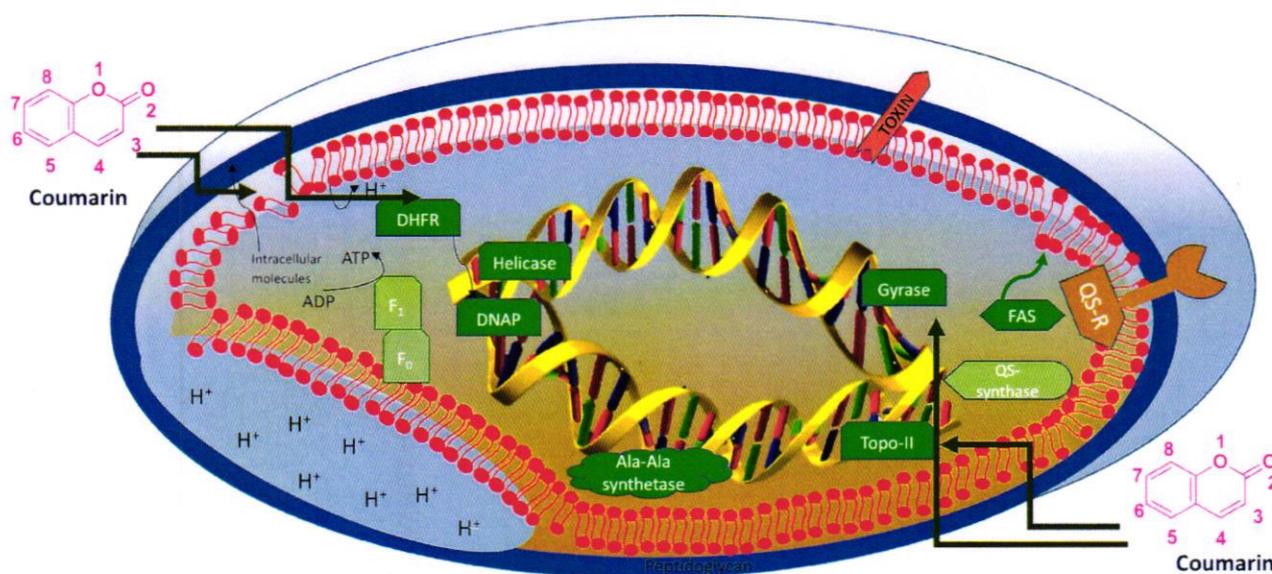
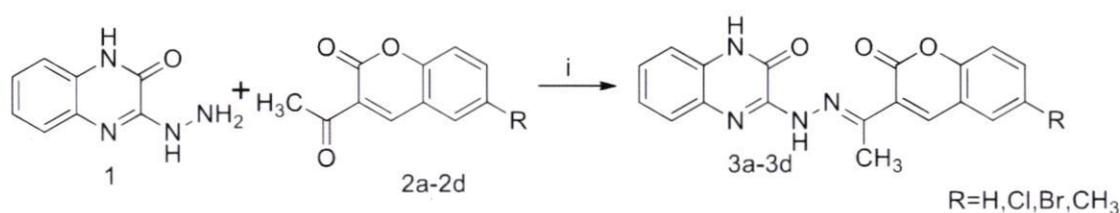


Fig. 2 Schematic representation of bacterial cell inhibitory actions of coumarin derivatives.



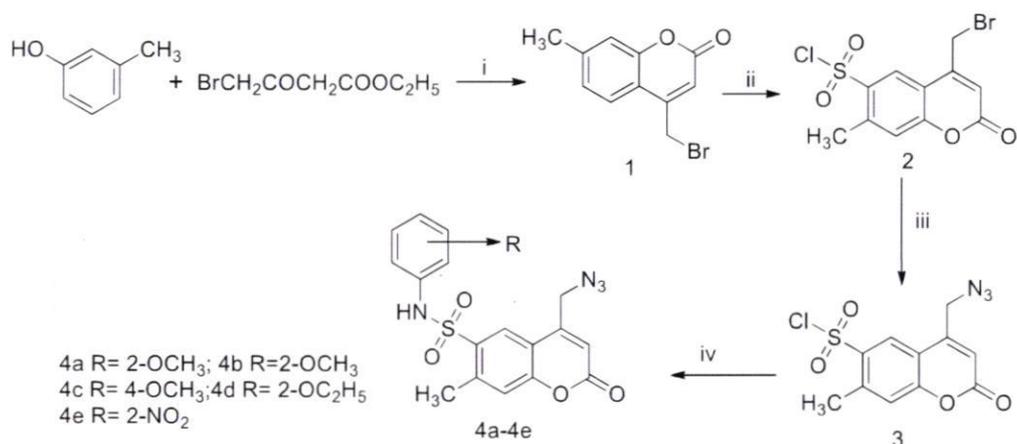
Scheme 1 2-Quinoxalono-coumarin hydrazone derivatives.

Ciprofloxacin. The potent compounds indicated that the compounds with chloro, nitro, methoxy and bromo substituent attached in phenyl ring presence may be in association with increased antibacterial activity (Basanagouda et al., 2010).

2.3. Synthesis of coumarin chalcone derivatives

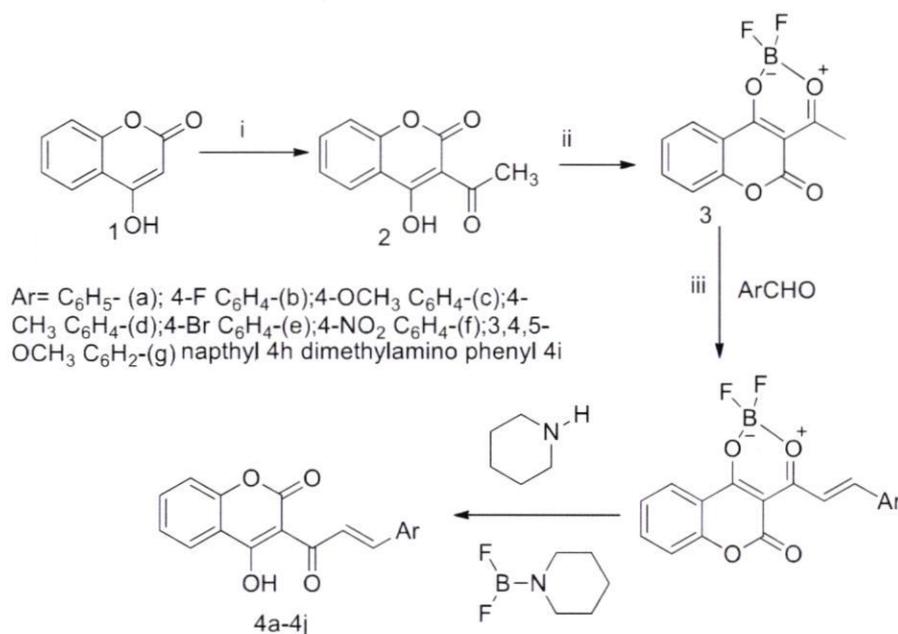
A series of coumarin chalcones had been synthesised and these derivatives **4a-4i** were screened for the antibacterial activities

against several bacterial strains. Initially, acetylated coumarin **2** was obtained by acetylation of 4-hydroxy coumarin **1**, in the presence of acetic acid and phosphorus oxy trichloride, which further on heating toluene solution the compound **2** in boron trifluoride etherate the corresponding boron compound **3** was obtained. Finally, the corresponding chalcones of coumarin derivatives were prepared by the condensation of 3-acetyl-difluoroboronyloxycoumarin **3** with different aryl aldehyde in chloroform solution and a mixed small amount piperidine



Reagents and conditions: i) Conc. H₂SO₄, 0-5°C ii) ClSO₃H, reflux iii) NaN₃, acetone, reflux iii) substituted aniline, benzene

Scheme 2 4-Azidomethyl-7-methyl-coumarin bearing sulfonamide.



Reagents and conditions: i) Acetic acid, POCl₃ ii) Toluene, boron trifluoride iii) CHCl₃, piperidine, 80°C

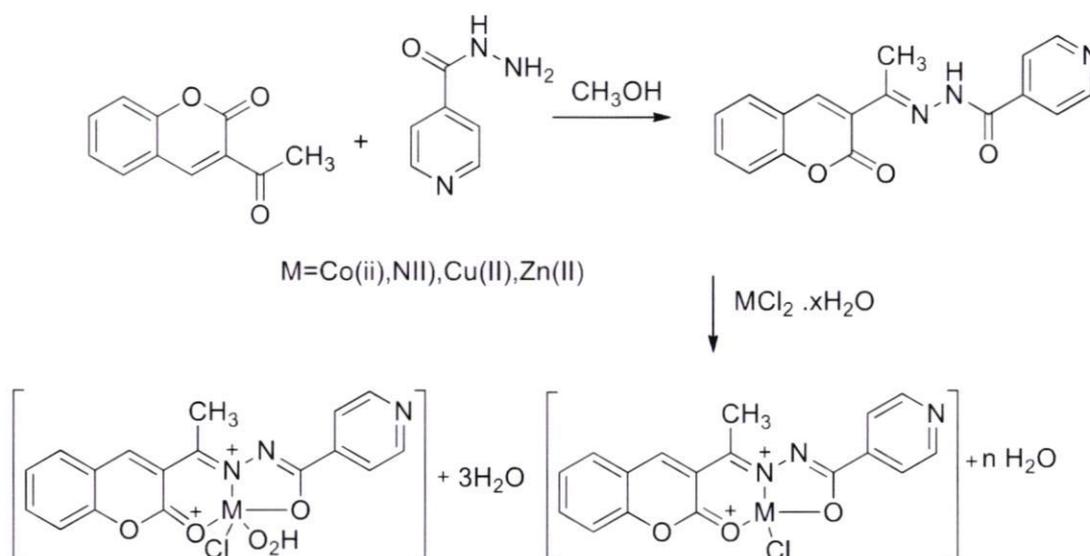
Scheme 3 Coumarin chalcone derivatives.

was added and the mixture was refluxed at 80 °C. The compound coumarin bearing naphthyl containing chalcone analogue, **4h** was obtained and coumarin derived 4-dimethyl amino phenyl chalcone **4i** had shown a good antibacterial activity against *E. coli* with 31 and 32 mm as ZOI, compared to standard Gentamicin (Hamdi et al., 2010).

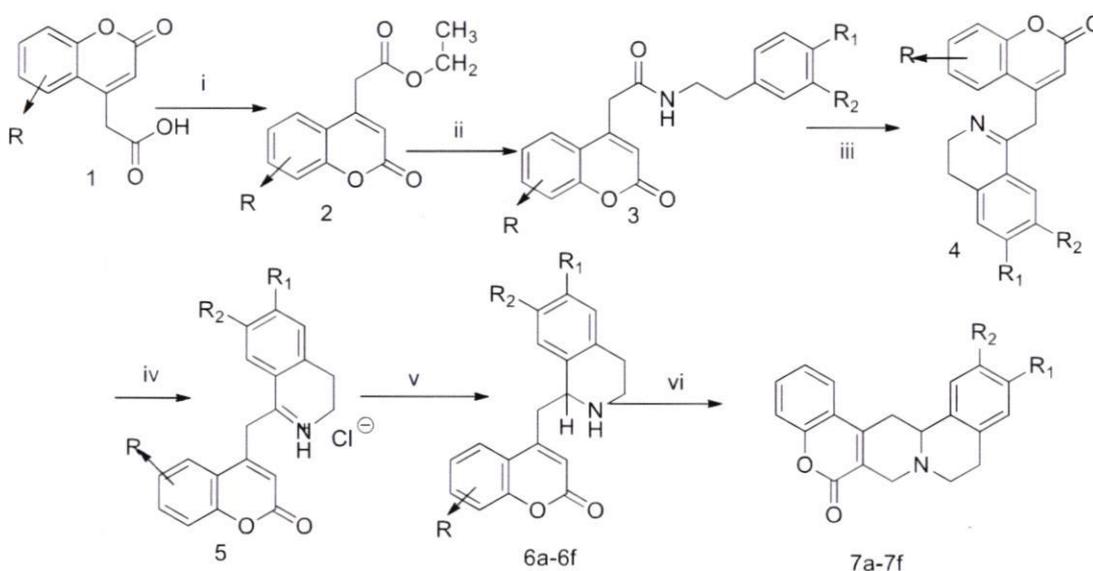
2.4. Synthesis of metal complexes of 3-acetylcoumarin-INH hybrid

Four transitional metal M(II) complexes of Schiff base of isoniazid **4a-4d** were synthesized by reaction of two potent ligands

such as, isoniazid **2** with 3-acetyl coumarin **1** in methanol on mild condition; by the by, the results of antibacterial action against some pathogenic bacterial strains were notable. In this reaction, ligands can chelate to metal ions in the carboimidol form of INH hydrazone via deprotonation, coumarin carbonyl and azomethine also involved the same. Antitubercular activity results showed that compounds schiff base INH-coumarin has been less effective than corresponding metal complexes. Among all the complexes, the Cu(II) complexes had notable antibacterial activity against pathogenic *S. aureus*, *S. faecalis* and *E. coli* at 25, 6.25, 50 µg/mL, whereas Co(II) and Ni(II) had been noted anti-tubercular activity at each MIC value



Scheme 4 Metal complexes of 3-Acetylcoumarin-INH hybrid.



R = 6-Methyl; 7-methyl; 7,8-benzo; 5,6-methylenedioxy $R_1=R_2=H$ and $R_1=R_2=3,4\text{-di-OCH}_3$

Reagents and conditions: i) $\text{C}_2\text{H}_5\text{OH}, \text{H}_2\text{SO}_4, \text{reflux}, 24\text{h}$ ii) phenyl ethyl amine, toluene, reflux, 6h iii) P_2O_5 , dry xylene, reflux, 12h iv) 20% HCl, heat v) $\text{NaBH}_4, \text{CH}_3\text{OH}, \text{rt}, 1\text{h}$ vi) $\text{HCHO}/\text{CH}_3\text{COOH}, \text{reflux}, 12\text{h}$

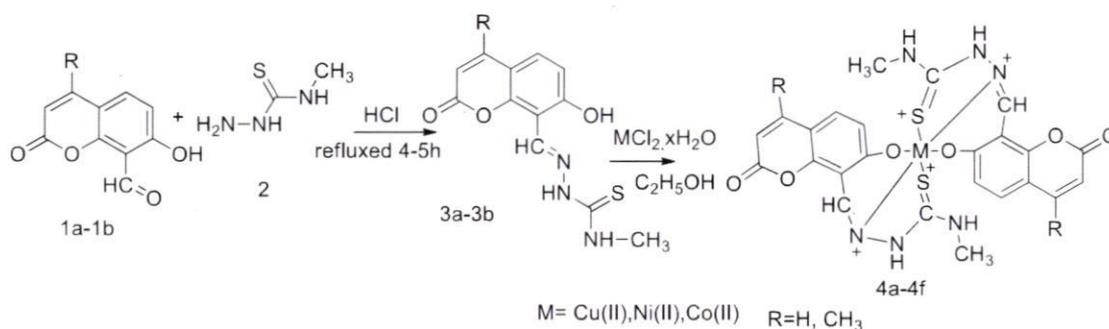
Scheme 5 Coumarin fused with tetrahydroisoquinoline derivatives.

25 $\mu\text{g}/\text{mL}$ in compare to Ciprofloxacin and Isoniazid as the standard, respectively (Hunoor et al., 2010).

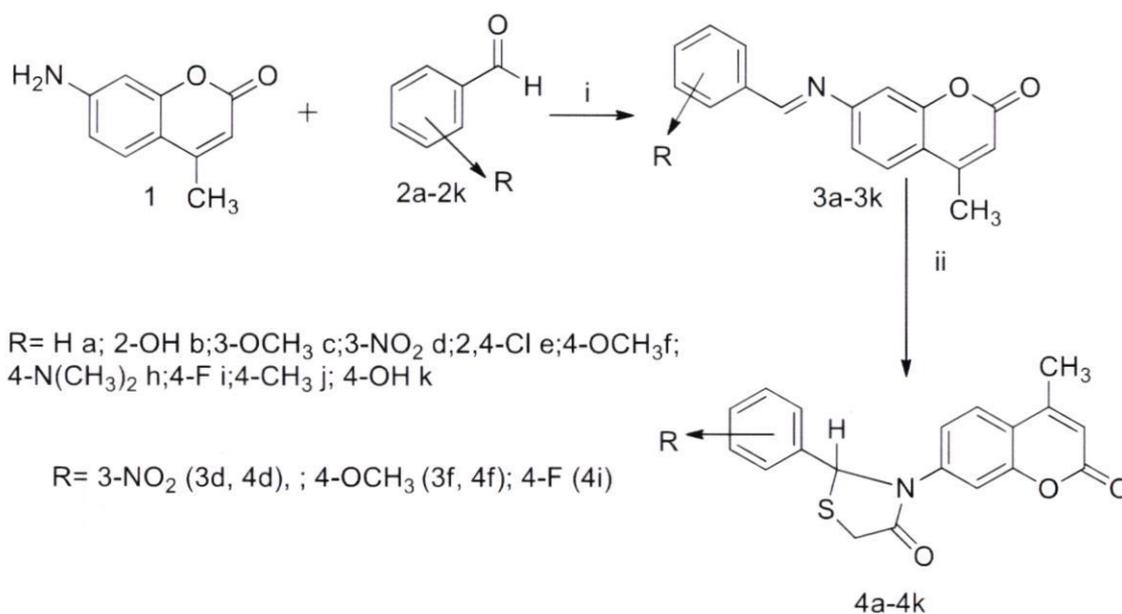
2.5. Synthesis of coumarin fused with tetrahydroisoquinoline derivatives

On the basis of Bischler-napieralsky's protocols, two series of coumarin congeners of tetrahydroisoquinoline **6a-6f** and protoberberine **7a-7f** were synthesised, and the obtained con-

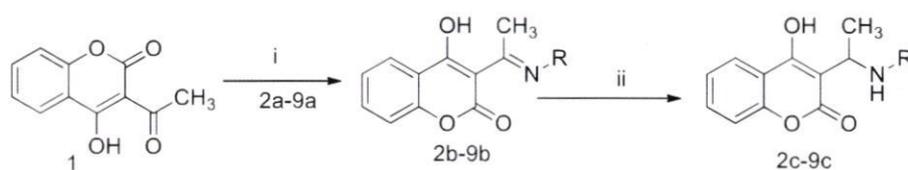
geners were tested for DNA cleaving activity against infection bacterial strains. From the precursor, coumarin 4-carboxylic acid **1** was esterified with ethanol to obtain the corresponding ethyl ester **2**, which further on reaction with disubstituted aryl ethyl amine in toluene solution, yielded 2-(2-oxo-2H-chromen-4-yl)-N-disubstituted phenethyl acetamide **3**. Then, the correspondingly obtained acetamide **3** was readily cyclized in the presence of phosphorous pentoxide to produce the compound **4** containing methylated



Scheme 6 Metal complexes with 4-Methyl 7-hydroxy coumarin thiosemicarbazone.



Scheme 7 4-Methyl coumarin bearing thiazolidinone moiety.



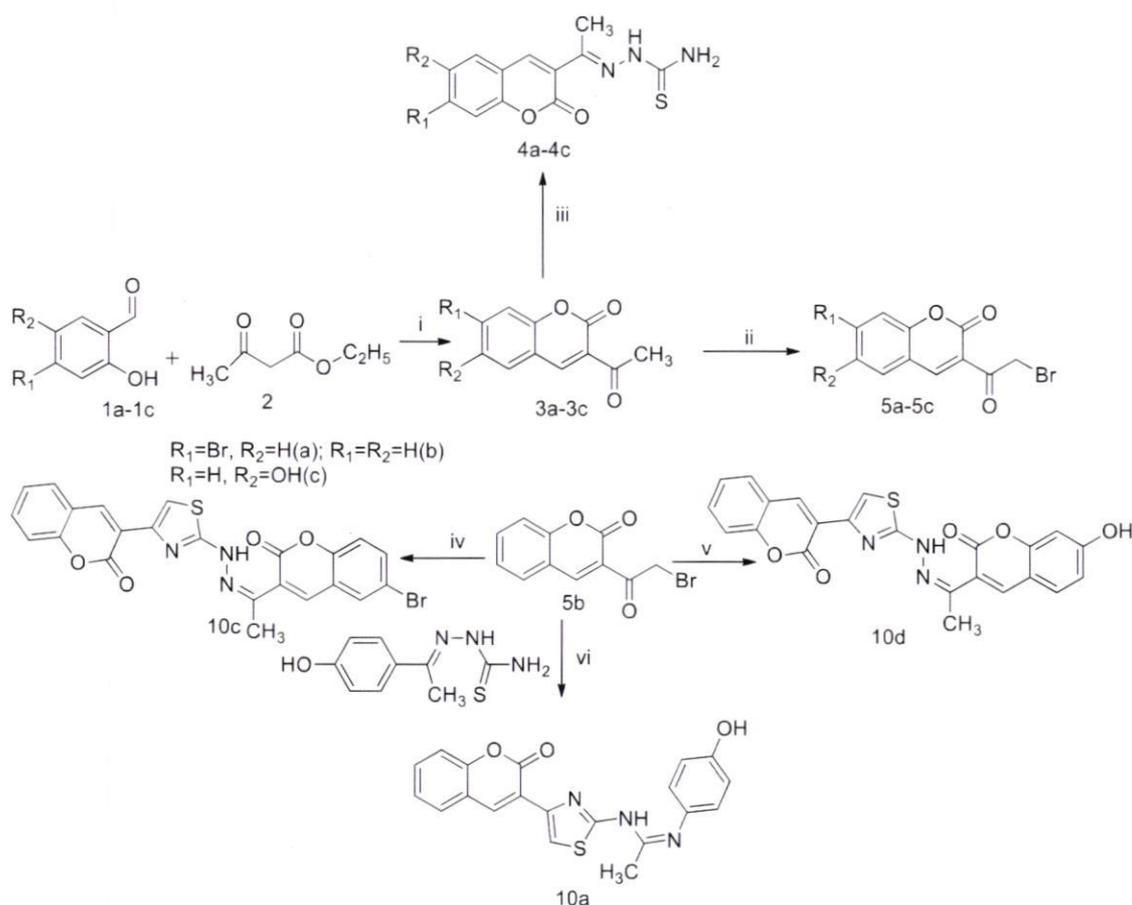
R= Phenyl- 2; 4-tolyl- 3; 3-tolyl- 4; 2-tolyl 5; 4-NO₂-ph 6; 3-NO₂-ph 7; benzyl 8; omegaC₄H₈COOH 9

Reagents and conditions: i) Substituted arylamines, TsOH, toluene, reflux ii) sodium tetraborohydride, THF, MeOH

Scheme 8 3-Aminoalkyl-4-hydroxy coumarin derivatives.

disubstituted, 3,4-dihydroisoquinolyl linked to C-4 position coumarin nucleus. Furthermore, the liberated compound 1,2,3,4-tetrahydroquinoline **5** formed by the reduction of the product **4** in sodium tetra borohydride (NaBH₄), and finally the fragmented coumarin ring was fused with tetrahydroquinoline; the compound **5** was synthesized by reflux condensation of the desired tetrahydroquinoline **6a-6f** with methanol in acetic acid. These intermediate candidates fur-

ther by cyclisation and condensation with formaldehyde in acetic acid produced another series of coumarin congeners **7a-7f**, which were recognised chemically as protoberberine. Among all the desired compounds **6f**, **7e** and **7f**, was fused with 3,4-dimethoxy or 7,8-benzo system of coumarin moieties and those had been reported as good inhibitory actions against bacterial DNA cleavage of *S. aureus* (Jadhav et al., 2010).



Reagents and conditions: i) Piperidine, 0-5°C ii) $\text{Br}_2/\text{CHCl}_3$, 0-5°C iii) $\text{NH}_2\text{NHCSNH}_2$, CH_3COOH , CH_3OH , reflux iv) $\text{C}_{12}\text{H}_{10}\text{BrN}_3\text{O}_2\text{S}$ (4a), $\text{C}_2\text{H}_5\text{OH}$, CHCl_3 , reflux, NH_4OH (5%) v) $\text{C}_{12}\text{H}_{11}\text{N}_3\text{O}_3\text{S}$ (4c) $\text{C}_2\text{H}_5\text{OH}$, CHCl_3 , reflux, NH_4OH (5%) vi) 2-(1-(4-hydroxyphenyl)ethylidene)hydrazine carbothioamide, $\text{C}_2\text{H}_5\text{OH}$, CHCl_3 , reflux

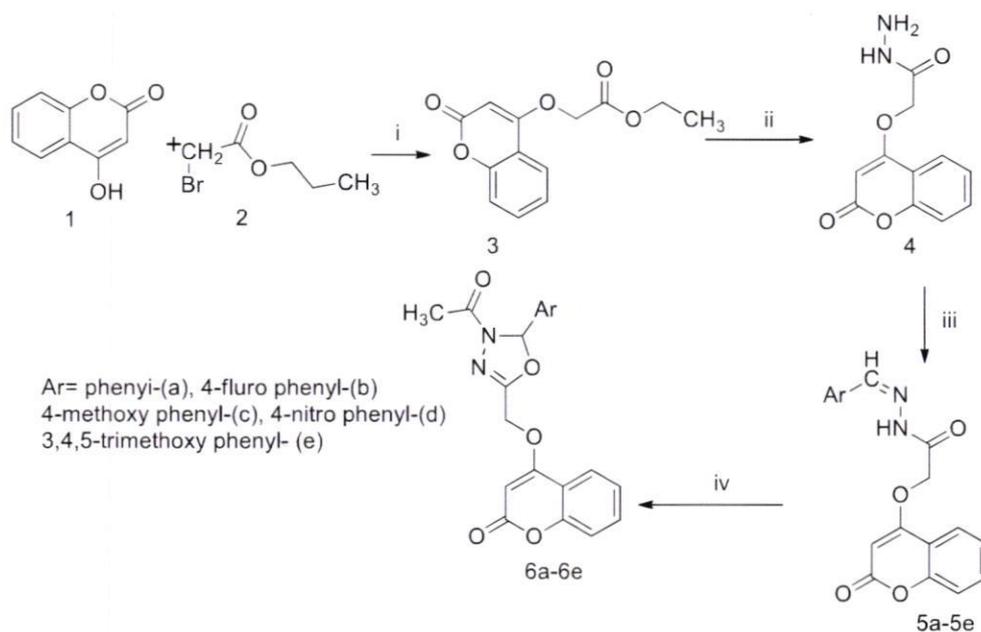
Scheme 9 Hydrazonyl thiazolyl substituted coumarin derivatives.

2.6. Synthesis of metal complexes with 4-methyl 7-hydroxy coumarin thiosemicarbazone

A series of Ni(II), Co(II) and Cu(II) complexes had been synthesized from two coumarin thiosemicarbazone **3a-3b** derivatives, and the obtained analogues and metal complexes **4a-4f** were evaluated for antibacterial activity. An individual compound such as, 7-formyl-6-hydroxy 4-methyl coumarin **1a** and 8-formyl-7-hydroxy coumarin **1b** was reacted with *N*-methyl thiosemicarbazide **2** to produce the corresponding *N*-methyl thiosemicarbazone-coumarin **3a-3b** (ligands). The desired ligands act as tridentate, which coordinated to metal ions through azomethine group, thione of sulfur and phenolic hydroxyl group *via* deprotonation. The Co(II) and Ni(II) complexes of 4-methyl 7-hydroxy coumarin thiosemicarbazone **3a** had been found notable inhibitory activities against *E. coli*, *P. aeruginosa* and *S. typhi* at each MIC value 10 $\mu\text{g}/\text{mL}$ in comparison to Gentamicin (Patil et al., 2010).

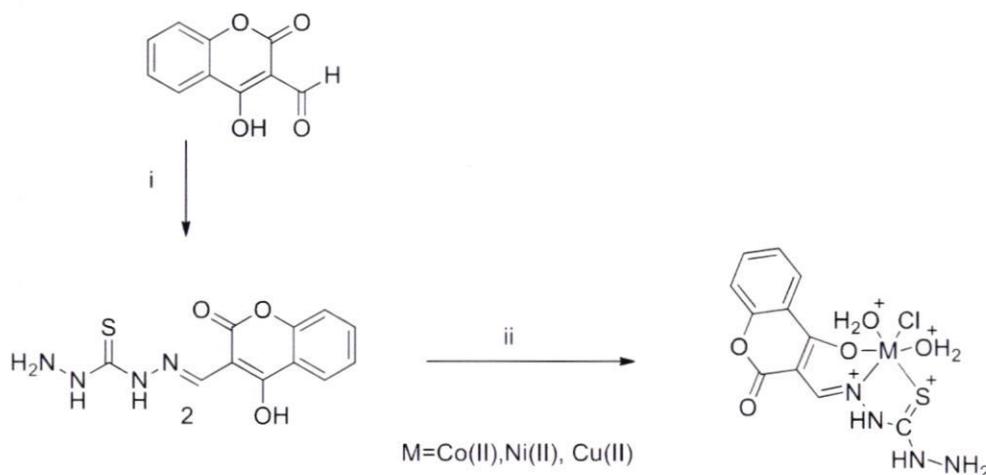
2.7. Synthesis of 4-methyl coumarin bearing thiazolidinone moiety

In this scheme, thiazolidinonyl bearing coumarin analogues had been synthesised by the reaction of 7-amino 4-methyl coumarin **1** with appropriate substituted aryl aldehyde **2a-2k** to produce several Schiff base derivatives **3a-3k**, which on further treatment with thioglycolic acid in the presence of anhydrous zinc chloride yielded cyclised corresponding title analogues **4a-4k**. Furthermore, SAR studies of these derivatives attributed that the compounds with *para*-substituents of phenyl of thiazolidinonyl ring induce their antibacterial potencies. Preliminary, *in-vitro* antibacterial activity of these Schiff base compounds had been less active than its corresponding thiazolidinonyl analogue. The obtained result indicated that the compounds with nitro, methoxy and fluoro substituents in phenyl ring of title analogues such as **3d**, **3f**, **4d**, **4f** and **4i** exhibited good antibacterial agent at MIC doses 100–10 $\mu\text{g}/\text{mL}$ in comparison to Ciprofloxacin (Ronad et al., 2010).



Reagents and conditions: i) Acetone, K_2CO_3 , reflux ii) NH_2NH_2 , C_2H_5OH iii) $ArCHO$, C_2H_5OH iv) Acetic anhydride

Scheme 10 4-Hydroxycoumarin bearing oxadiazolone derivatives.



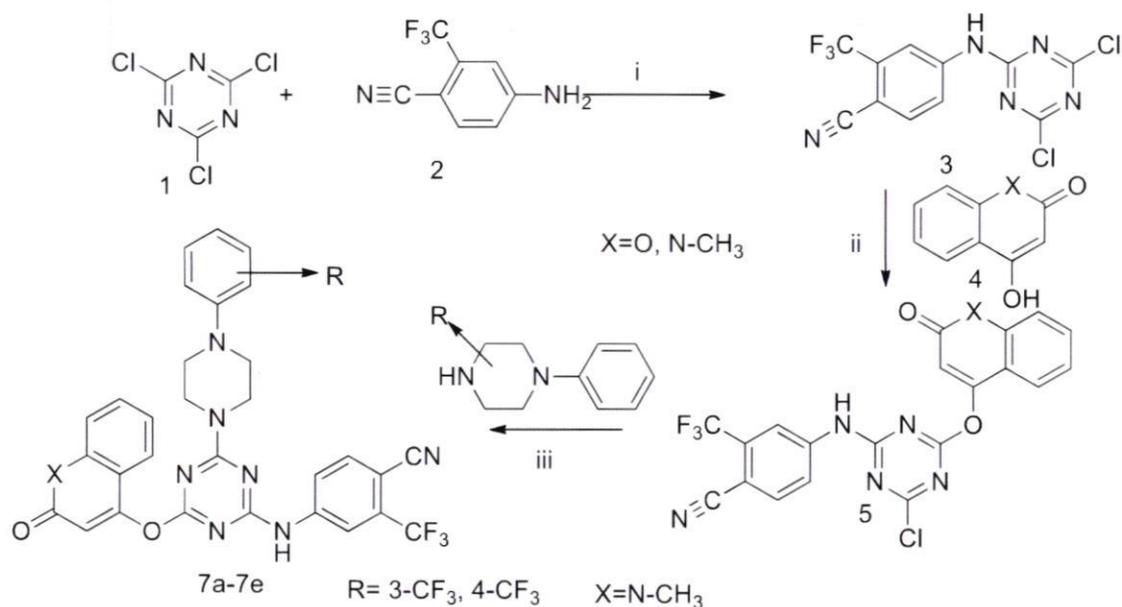
Reagents and conditions: i) $NH_2CSNHNH_2$, CS_2 , DMF ii) $[M(H_2O)Cl_2] \cdot EtOH$

Scheme 11 Metal complexes-semithiocarbazone of 4-Hydroxy coumarin.

2.8. Synthesis of 3-aminoalkyl-4-hydroxy coumarin derivatives

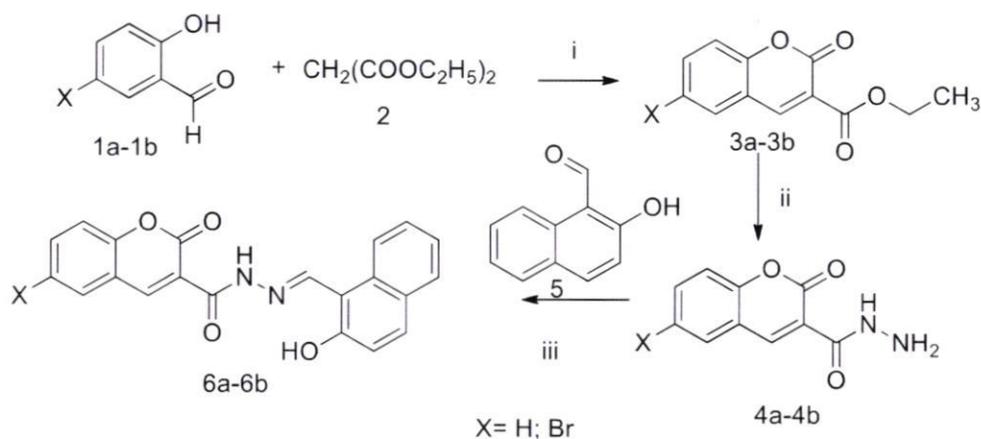
A series 3-amino alkylated-4-hydroxy coumarin derivatives was synthesized by both conventional and microwave assisted methods. Then, the liberated products were examined for antioxidant and antibacterial activities. These analogues had been synthesized by the reaction of 3-acetyl 4-hydroxy coumarin **1** with substituted aryl amines in the presence of catalyst toluene sulfonic acid in anhydrous toluene to produce the respective Schiff based coumarin derivative **2**, which further was reduced by an individual analogue with sodium borohy-

dride in tetrahydrofuran to yield monoalkylated 4-hydroxy coumarin **3**; whereas yields of these analogues had been comparatively more in using microwave assisted methods than the conventional synthetic method, particularly for the analogue namely, nitro substituted anilino derivatives. The antibacterial results of the compounds with unsubstituted phenyl, tolyl residue at anilino of the title compounds namely, **9c**, **2c** and **3c** were of notable inhibition growth against *S. aureus*, *E. coli* and *P. fluorescence* at ranges of zones of inhibition (ZOI) 3.9–15.6 $\mu g/mL$ in comparison to Streptomycin as the standard (Vukovic et al., 2010).



Reagents and conditions: i) THF/(C₂H₅)₃, ii) THF, NaH iii) K₂CO₃, 1,4-Dioxane

Scheme 12 Coumarin bearing triazines derivatives.



Reagents and conditions: i) C₂H₅OH, piperidine, reflux 5h ii) C₂H₅OH, NH₂NH₂·H₂O, reflux 10h
iii) C₂H₅OH, reflux 4-5h

Scheme 13 Metal complexes-Coumarin schiff base derivatives.

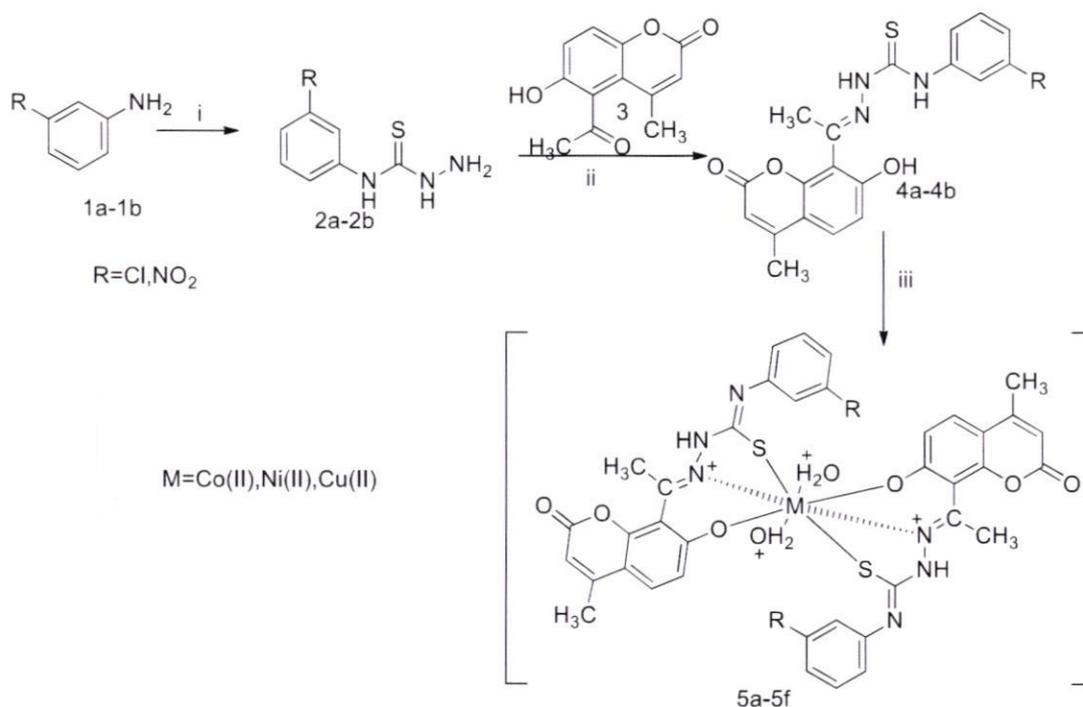
2.9. Synthesis of thiazolyl hydrazoneyl substituted coumarin derivatives

Twelve derivatives bearing hydrazoneyl thiazolyl substituted coumarin were synthesised by the reflux condensation of 3-bromoacetyl coumarin **5b**, and substituted phenyl/substituted 3-acetylcoumarin thiosemicarbazone **4a-4c** in chloroform and ethanol (2:1) yield thiazolyl linked coumarin analogues. By the principle of Hantzsch's reaction, the formation of thiazole ring, which was incorporated in structure between bromoacetyl group and the corresponding thiosemicarbazone congener, in the presence of mixed solvents ethanol and chloroform. The insertion of hydroxyl group and bromo sub-

stituents of benzylidene imine residue had resulted in significant antimycobacterial activity. The compound 6-bromo-3-(1-(2-(4-(2-oxo-2H-chromen-3-yl)thiazol-2-yl)hydrazone)ethyl)-2H-chromen-2-one **10c** was reported as a good antimycobacterial agent with 15 μM as the MIC value in comparison to Isoniazid (INH) (Arshad et al., 2011).

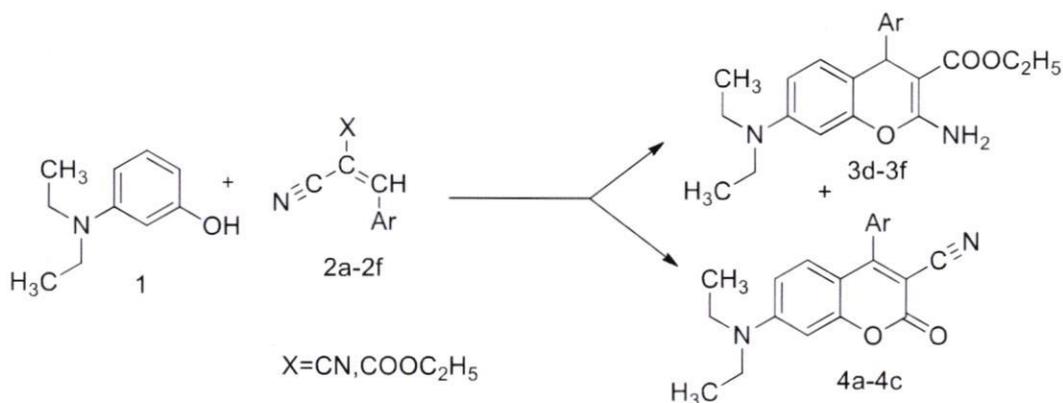
2.10. Synthesis of 4-hydroxy coumarin bearing oxadiazolone derivatives

A novel series of 4-hydroxycoumarin derivatives bearing oxadiazolonyl, was prepared by an intermediate compound ethyl 2-((2-oxo-2H-chromen-4-yl)oxy)acetate **3**; this underwent



Reagents and conditions: i) CS_2 , $\text{NH}_2\text{NH}_2 \cdot \text{H}_2\text{O}$, str., 2-3h, rt ii) $\text{C}_2\text{H}_5\text{OH}$, conc. HCl , reflux 2-3h iii) $\text{MCl}_2 \cdot x\text{H}_2\text{O}$, $\text{C}_2\text{H}_5\text{OH}$, reflux 3h

Scheme 14 Metal complexes of Coumarin-thiosemicarbazones derivatives.



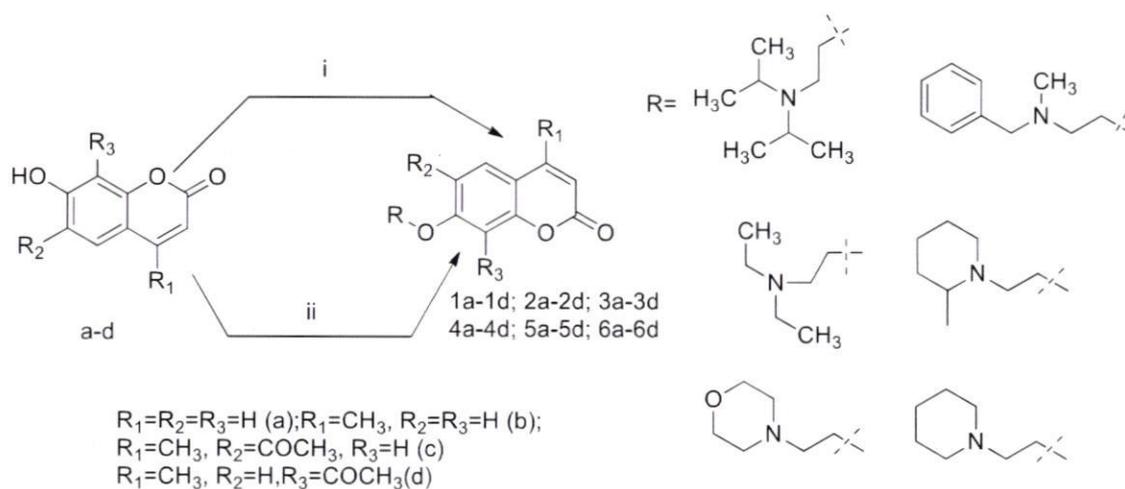
Ar- 4-Cl C_6H_4 , X-CN (a); Ar- 4-Br C_6H_4 , X-CN (b); Ar- 4-FC $_6\text{H}_4$, X-CN (c);
Ar- 4-Cl C_6H_4 , X-COOC $_2\text{H}_5$ (d); Ar- 4-Br C_6H_4 , X-COOC $_2\text{H}_5$ (e) Ar- 4-F C_6H_4 , X-COOC $_2\text{H}_5$ (f)

Reagents and conditions: i) $\text{C}_2\text{H}_5\text{OH}$ ii) piperidine, reflux

Scheme 15 3,4,7-Tri substituted coumarin derivatives.

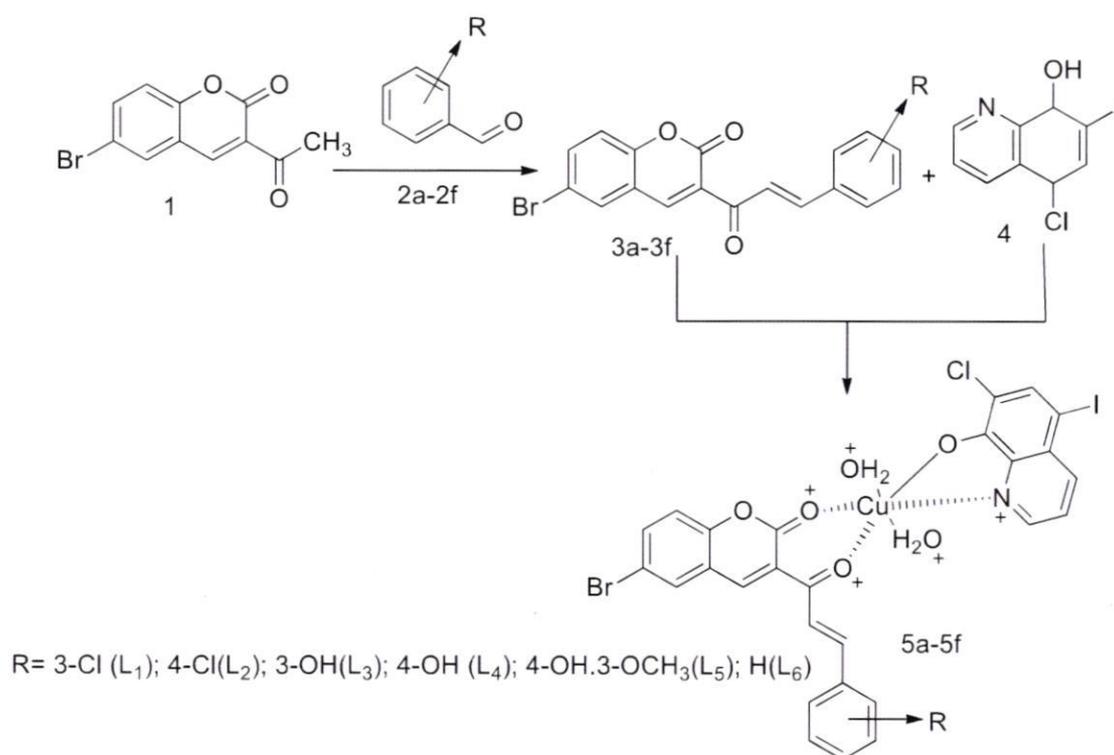
hydrazinolysis in hydrated hydrazine to produce the corresponding hydrazide **4**, which further was used to synthesize by the corresponding Schiffbase with the condensation with different aryl aldehyde in glacial acetic acid for getting *N*-substituted benzylidene-2-((2-oxo-2*H*-chromen-4-yl)oxy)

aceto-hydrazide **5**. Finally these obtained compounds individually reacted and cyclised with acetic anhydride to produce the corresponding 4-((4-acetyl-5-subst. phenyl-4,5-dihydro-1,3,4-oxadiazol-2-yl)methoxy)-2*H*-chromen-2-one **6**. Introduction of 4-fluoro phenyl substituted in the oxadiazolanyl nucleus in



Reagents and conditions: i) Substituted amino alkyl chloride, dry acetone, reflux ii) substituted amino alkyl chloride, MW

Scheme 16 2-Amino alkylated coumarin derivatives.

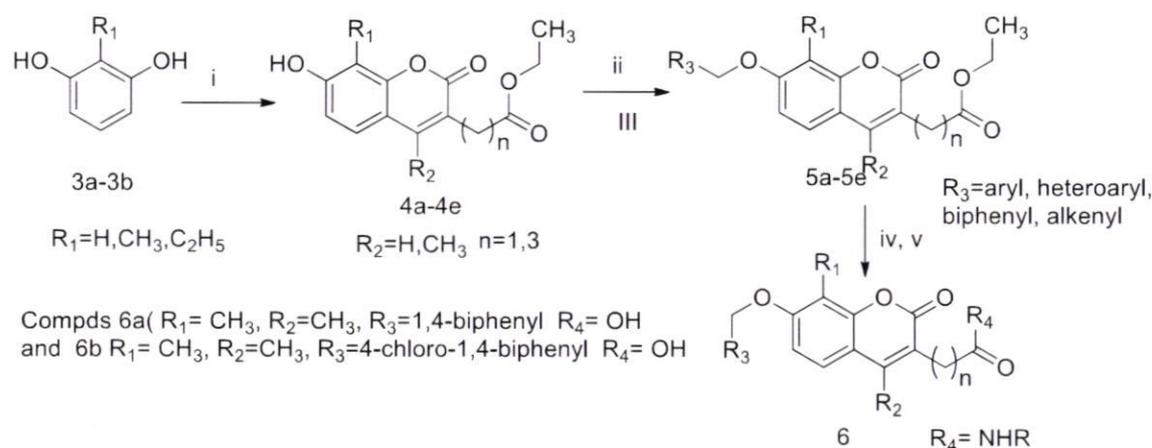


Scheme 17 Metal complexes with Coumarin chalcones-cloioquinol.

compounds **5b** and **6b** had a notable antibacterial activity against *S. aureus*, *E. coli* and *P. aeruginosa* with ZOI as 26–34 mm in comparison to Gentamicin. In these prepared compounds, fluoro substituted phenyl in either hydrazone Schiffbase structure or attached with one of substituent in oxadiazinyl ring had been verified for antibacterial activity (Hamdi et al., 2011).

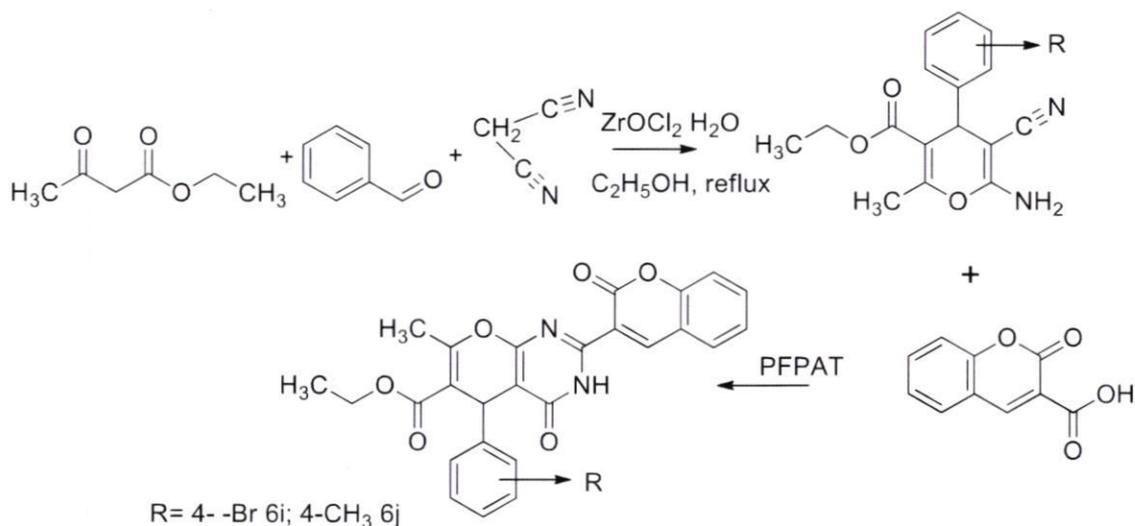
2.11. Synthesis of metal complexes-semithiocarbazone of hydroxy coumarin

Several transitional metal complexes of 4-hydroxy-3-thiocarbohydrazone **2** were formed by the reaction with chloride, acetate and nitrate salts of the following metals, Co(II),



Reagents and conditions: i) $\text{CH}_3\text{COCHCO}_2\text{C}_2\text{H}_5(\text{CH}_2)_n\text{CO}_2\text{C}_2\text{H}_5, \text{H}_2\text{SO}_4, 0^\circ\text{C}$, or $\text{HCOCHCO}_2\text{C}_2\text{H}_5(\text{CH}_2)_2\text{CO}_2\text{C}_2\text{H}_5\text{H}_2\text{SO}_4, \text{RT}$ ii) $\text{R}_3\text{CH}_2\text{X}, \text{Na}_2\text{CO}_3, \text{DMF}$, rt iii) N_2N , NaOH , rt iv) $(\text{COCl})_2, \text{DMF}$, THF v) R_4NH_2

Scheme 18 3,4,7,8-Tetra substituted coumarin derivatives.



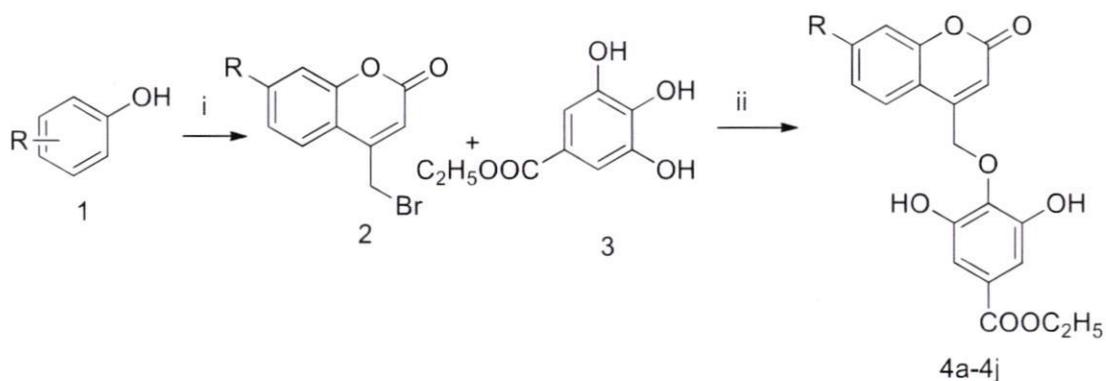
Scheme 19 Thiazolyl- pyrazoline coumarin derivatives.

Cu(II) , Ni(II) and Cr(III) in ethanol. The desired ligand was prepared by the reaction of 3-formyl 4-hydroxy coumarin with a mixture of carbon disulfide and thiosemicarbazide in the presence of dimethylformamide. In these complexation reactions, the ligand acted as monobasic tridentate ONS electron donors in all metal complexes. Furthermore, the obtained ligand with complexes structurally interpreted by thermal gravimetric analysis. Among all the metal complexes the compounds in which, the one with cobalt coordinated with ligand **2** had good antibacterial activity against *E. coli* at the MIC value, 100 $\mu\text{g/mL}$ (Mosa et al., 2011).

2.12. Synthesis of coumarin bearing triazine derivatives

Two series of quinolonyl/ coumarinyl triazine derivatives **7a-7e** and **8a-8d** were synthesized. In this synthesis, an intermediate

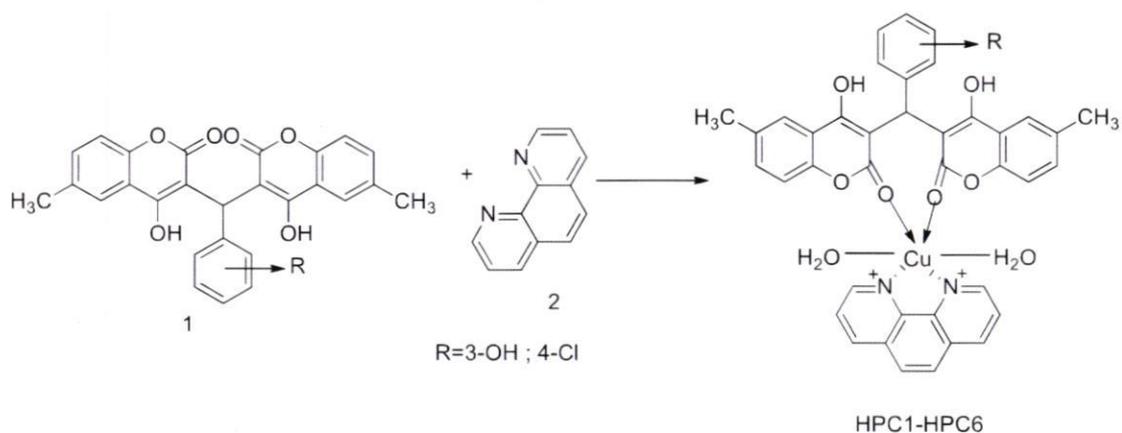
subtract, 4-((4,6-dichloro-1,3,5-triazin-2-yl)amino)-2-(trifluoromethyl)benzonitrile **3** was synthesized by the reaction mixture of 4-amino-2-trifluoromethyl benzonitrile **1** and trichloro-1,3,5-triazine **2** in the presence of triethylamine by nucleophilic displacement of chlorine atom from triazine nucleus. The obtained product further reacted with either 4-hydroxy coumarin or 1-methyl quinolone in the presence of sodium hydride in THF to produce another precursor of title compound **5**. Finally, the desired compounds corresponding 4-((4-chloro-6-((1-methyl-2-oxo-1,2-dihydroquinolin-4-yl)oxy)-1,3,5-triazin-2-yl)amino)-2-(trifluoromethyl)benzonitrile **7** and 4-((4-((2-oxo-2H-chromen-4-yl)oxy)-6-(4-phenylpiperazin-1-yl)-1,3,5-triazin-2-yl)amino)-2-(trifluoromethyl)benzonitrile **8** were prepared by nucleophilic displacement of another chlorine atom of product **5** with 4-substituted aryl piperazinyl **6** in the presence of 1,4-dioxane and potassium carbonate. The compounds



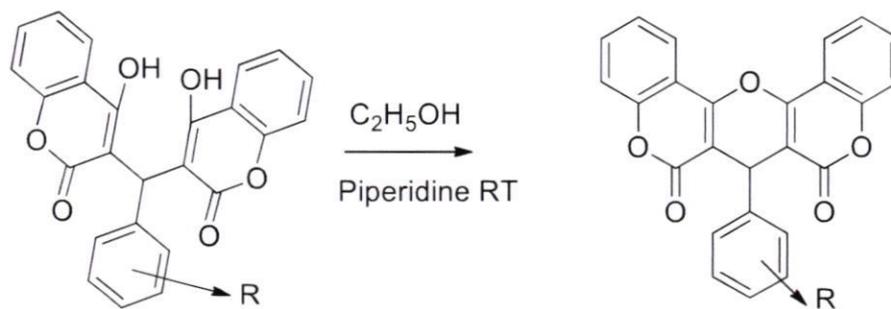
R= 6-CH₃ 3a; 7-CH₃ 3b; 6-Cl 3c; 7-Cl 3d; 6-OCH₃ 3e; 7-OCH₃ 3f; 5,7-CH₃ 3g ; 7,8-CH₃ 3h; 5,6-benzo 3i; 7,8-benzo 3j;

Reagents and conditions: i) BrCH₂COCH₂COOC₂H₅, H₂SO₄ ii) CH₃COCH₃, K₂CO₃

Scheme 20 4-Aryloxymethyl substituted coumarin derivatives.

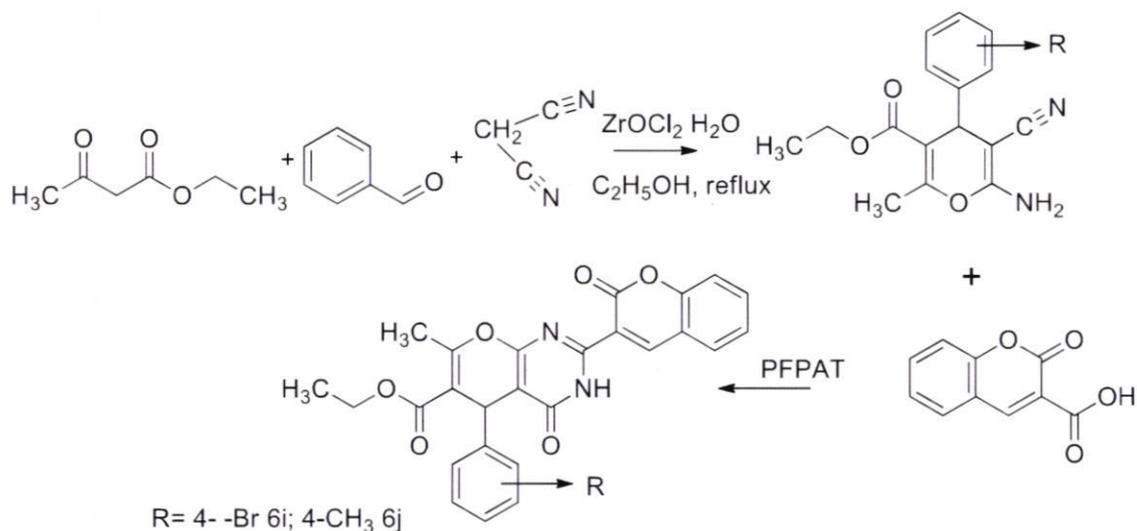


Scheme 21 Metal complexes of 4,4'-Bis-hydroxy coumarin with phenanthroline.

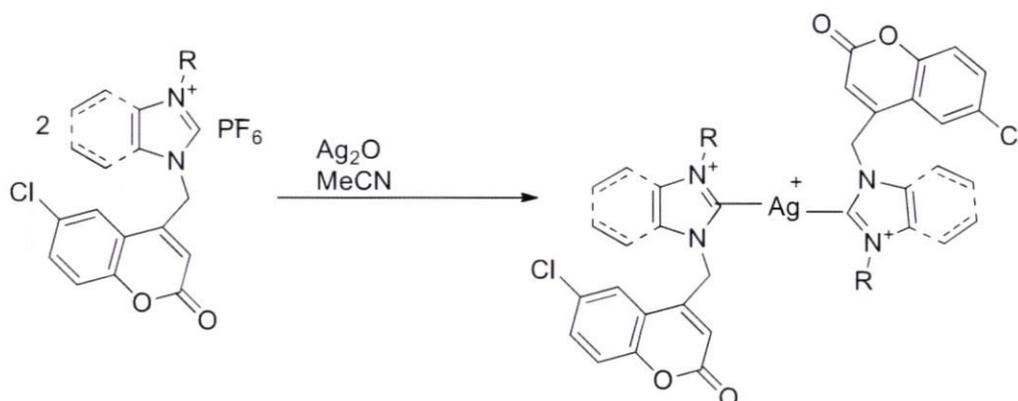


R= 3-CH₃; 3-MBH 3-MDH
R= 4-CH₃; 4-MBH 4-MDH

Scheme 22 Bis-4-Hydroxy coumarin derivative.



Scheme 23 Chromeno[2,3-*d*]pyrimidinone with coumarin derivatives.



Scheme 24 Silver salt of benzimidazolium with coumarin derivatives.

bearing quinolone as **7c** and **7d** had shown good antibacterial activity against *S. aureus* as 27 mm ZOI with the MIC value 6.25 µg/mL in comparison to the standard Ciprofloxacin. Moreover, these derivative **8d** contain coumarin ring in structure had shown good antibacterial action against *E. coli* at the MIC value 12.5 µg/mL (Patel et al., 2011).

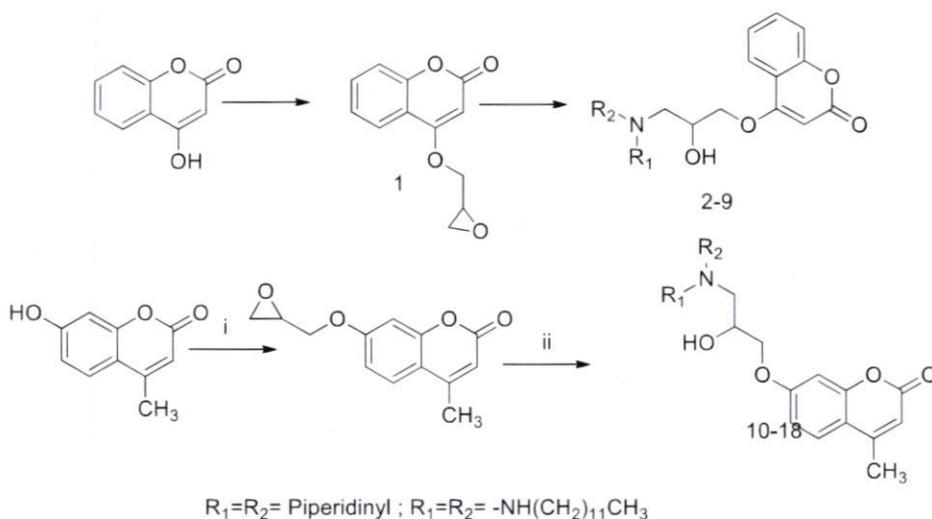
2.13. Synthesis of metal complexes-coumarin Schiff base derivatives

In this scheme, Co(II), Cu(II) and Ni(II) complexes of substituted (*E*)-*N'*-(2-hydroxynaphthalen-1-yl)methylene)-2-oxo-2*H*-chromene-3-carbohydrazide **6a-6b** have been synthesized with respective Schiff base derived from 2-hydroxy naphthaldehyde **5** and substituted coumarin 3-carbohydrazide **4a-4b**. The desired the Schiff base had been acting as tetradentate ligand, which metal ions coordinated to azomethine nitrogen, lactone carbonyl oxygen, hydrazide of nitrogen and naphthyl hydroxyl group and complexation through deprotonation of naphthanol. Then, the obtained Schiff base analogues **6a-6b** were liberated by the condensation of respective coumarin 3-carbohydrazide **4a-4b** with 2-hydroxy naphthaldehyde **5** in

ethanol. Consequently, the formation of respective complexes, **7a-7f** by treating the desired Schiffbase condenses with the corresponding hydrated metal chloride in ethanol and add few quantity of sodium acetate. The desired Schiff base and its complexes have been evaluated for *in vitro* antibacterial activity against several pathogenic bacterial strains by bacterial DNA cleaving method. Among all the complexes the compounds **7a** and **7b** had shown as good antibacterial activity against *E. coli* and MIC value at 10 µg/mL in comparison to Gentamicin (Patil et al., 2011a).

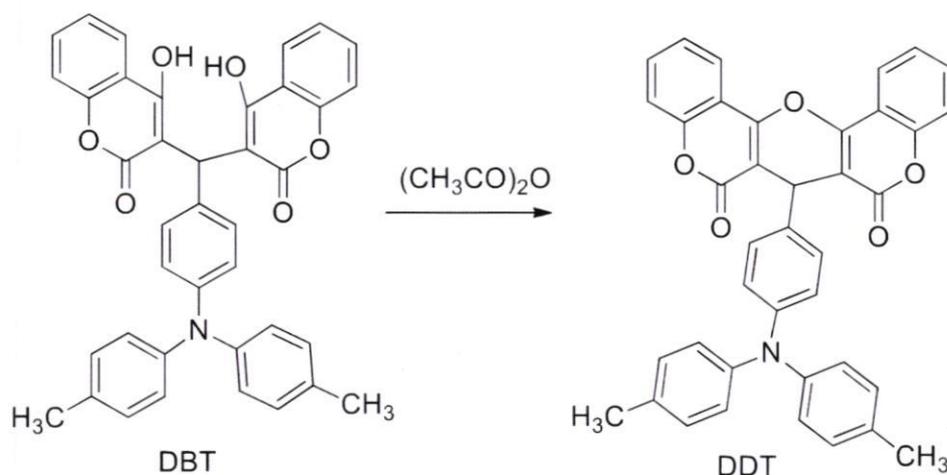
2.14. Synthesis of metal complexes of coumarin-thiosemicarbazones derivatives

A series of transitional metal complexes of Schiff base derived from 3-substituted aryl thiosemicarbazides **4a-4b** and 8-acetyl-7-hydroxy 4-methylcoumarin **3** had been synthesized. The acetylated compound **3** was prepared by the equimolar quantity of 7-hydroxy-4-methyl-coumarin and acetic anhydride in the presence of anhydrous aluminium chloride in the oil bath at 160–165 °C for 4 h. The desired Schiff base was prepared by the condensation of 8-acetyl 7-hydroxy coumarin **3** and



Reagents and conditions: i) Epichlorohydrine, anhydrous K_2CO_3 , reflux, 4h ii) corresponding amines, $\text{C}_2\text{H}_5\text{OH}$, 3-5h

Scheme 25 Coumarinyl piperazine bearing propanol derivatives.

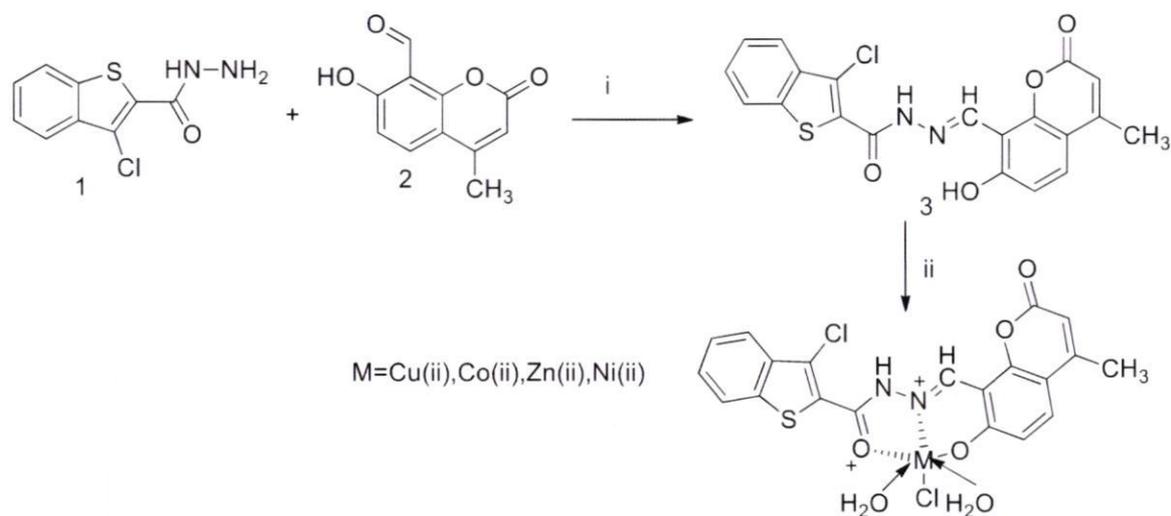


Scheme 26 Dicoumarol derivatives.

3-substituted aryl thiosemicarbazides **4a-4b** in ethanol with few drops of hydrochloric acid and followed respective metal complexes were formed by heating with the reaction mixture of equimolar concentration alcoholic solution metal chlorides and corresponding thiosemicarbazones in ethanol. During reaction, the metal ions were coordinated to azomethine group, thione-thiols and phenolic hydroxyl group of the structures through deprotonation, which indicate that the desired Schiff base act as tridentate ligand. The cobalt complexes of (*E*)-*N*-(3-chlorophenyl)-2-(1-(7-hydroxy-4-methyl-2-oxo-2*H*-chromen-8-yl)ethylidene)hydrazinecarbothioamide **5a** had shown antibacterial activity against *S. typhi*, *P. aeruginosa* and *E. coli* at MIC value 10 $\mu\text{g/mL}$ in comparison to Gentamicin as the standard (Patil et al., 2011b).

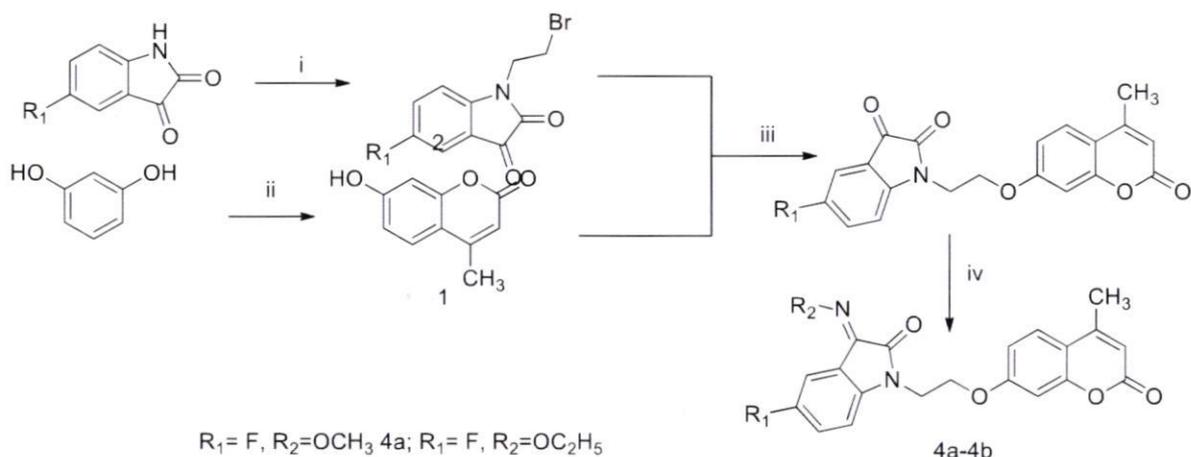
2.15. Synthesis of 3,4,7-tri substituted coumarin derivatives

A series of tri substituted coumarin analogues had been synthesized by heated with the reaction mixture of 3-diethylamino phenol **1** and various substituted α -cyano cinnamionitriles **2a-2f** in ethanol and base piperidine yields ethyl 2-amino-7-(diethylamino)-4-phenyl-4*H*-chromene-3-carboxylate **3a-3f** and 7-(diethylamino)-2-oxo-4-substituted phenyl-2*H*-chromene-3-carbonitrile **4a-4c**. The SARs of 4-substituted aryl coumarin analogues indicate that the substitution of chloro or fluoro at the para position of phenyl ring and attachment of cyano at C-3 position in the coumarin moiety, which may increase the inhibitory activity. The compounds bearing



Reagents and conditions: i) C_2H_5OH, CH_3COH ii) MCl_2, CH_3OH

Scheme 27 Metal complexes Thiophene hydrazine-coumarin.



Reagents and conditions: i) Dibromoethane, DMF ii) ethyl acetoacetate, conc. H_2SO_4 iii) DMF, K_2CO_3 iv) methoxyamine/ethoxyamine, Na_2CO_3

Scheme 28 Indolin-2,3-dione-coumarin derivatives.

4-substituted phenyl in either 4*H*-chromene 3-carboxylate or 2*H*-chromene 3-nitrile such as, **3a**, **3f** and **4c** were found notable antibacterial activity against *S. aureus* and *S. epidermidis* at 14 and 15 mm of ZOI diameter in comparison to Ampicillin (Sabry et al., 2011).

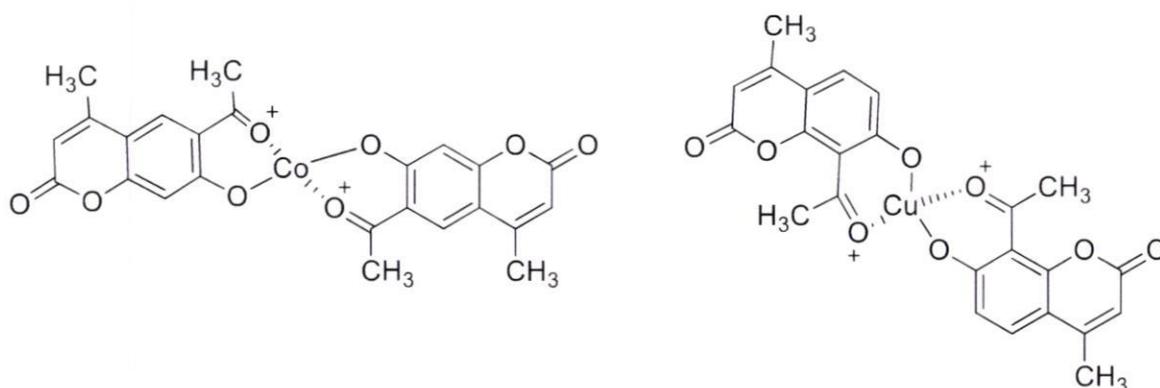
2.16. Synthesis of 4-amino alkylated coumarin derivatives

A series of compounds O-aminoalkyl substituted 7-hydroxy coumarin from substituted 7-hydroxy coumarin **a-d** were evaluated antibacterial activity with several bacterial strains. These, compounds were obtained from 7-hydroxy, C4, C6 or C8 substituted coumarin, in the initial step- the respective O-alkyl amino coumarin derivatives synthesized by alkylation

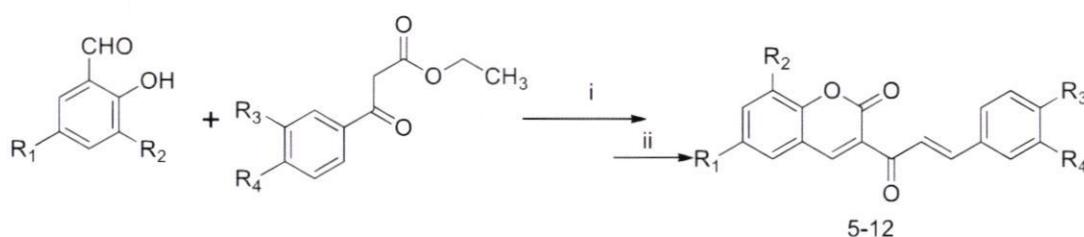
of phenol moiety of derivatives of coumarin with suitable chloro ethyl substituted amines in dry acetone as well as, potassium carbonate and the final derivatives were converted into hydrochloride salts. Furthermore, the antibacterial activity of the compound alkylamino substituted phenolic OH at C-7 position and 6-acetyl-4-methyl-7-(2-morpholinoethoxy)-2*H*-chromen-2-one had been reported with notable inhibitory activity against *Bacillus* strains (Trykowska Konc et al., 2011).

2.17. Synthesis of metal complexes with coumarin chalcones-cloquinol

A series of metal complexes of coumarin compounds were derived from corresponding chalcones **3a-3f**, which had



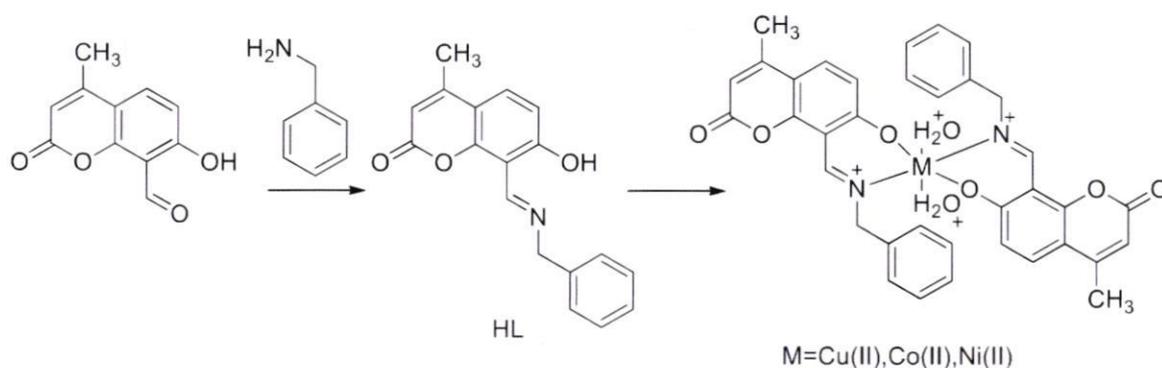
Scheme 29 Metal complexes of Acetyl coumarin derivatives.



5. $R_1=R_2=R_3=H, R_4=NH_2$ 6. $R_1=R_2=H, R_3=OCH_3, R_4=NH_2$
 7. $R_1=R_2=H, R_3=OCH_2CH_3, R_4=NH_2$, 8. $R_1=Br, R_2=R_3=H, R_4=NH_2$
 9. $R_1=R_2=R_4=H, R_3=NH_2$ 10. $R_1=R_4=H, R_2=OCH_3, R_3=NH_2$,
 11. $R_1=R_4=H, R_2=OCH_2CH_3, R_3=NH_2$, 12. $R_1=Br, R_2=R_4=H, R_3=NH_2$

Reagents and conditions: i) C_2H_5OH , piperidine, reflux, 2-5hr ii) $SnCl_2$, $2H_2O$, C_2H_5OH , reflux, 3-7h

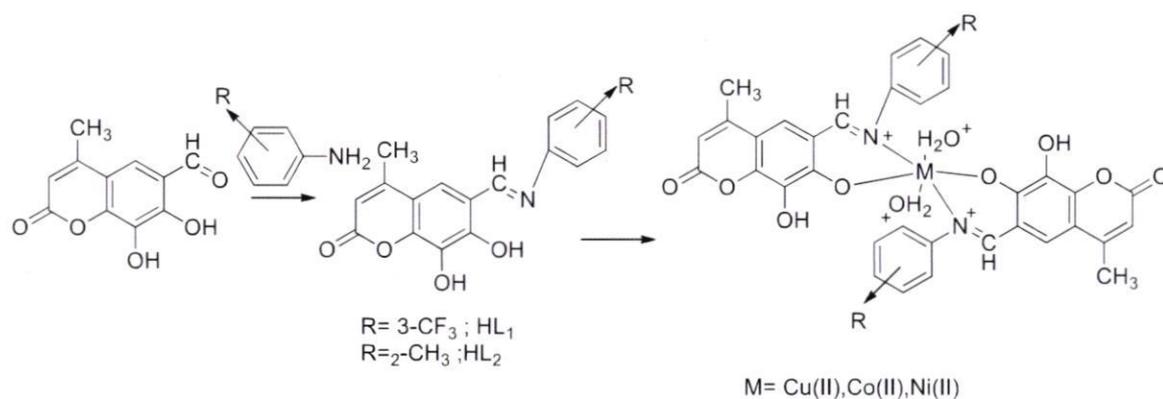
Scheme 30 Arylchalcone of coumarin derivatives.



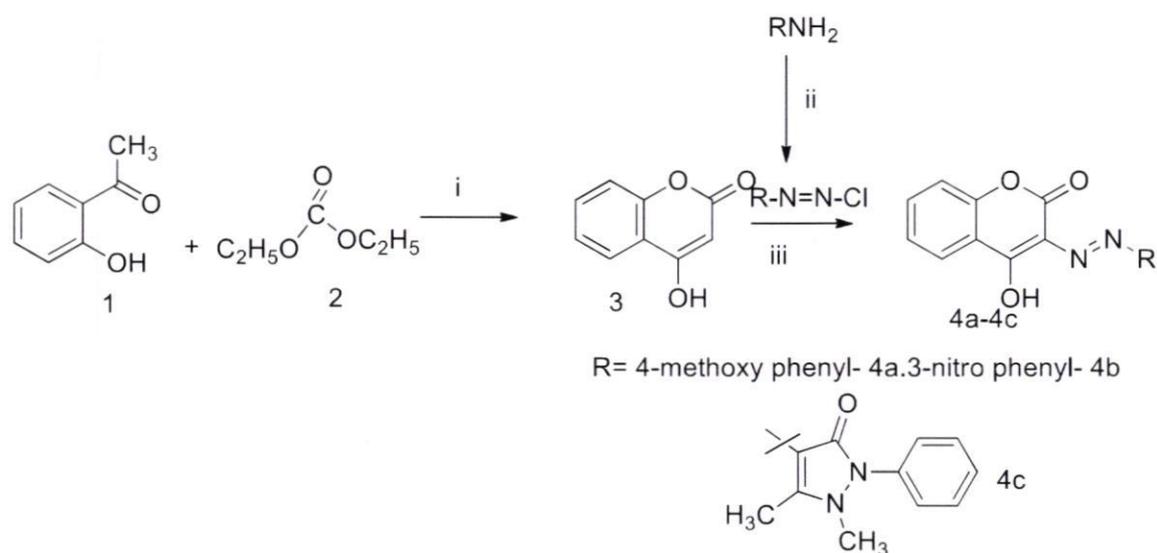
Scheme 31 Metal chelate of Schiff base with coumarin derivatives.

been prepared by the condensation of 6-bromo-3-acetyl coumarin **1** with the appropriate substituted aryl aldehyde **2a-2f** in piperidine base and ethanol. Cu(II) complexes had been synthesized by the heating the mixture compounds of cupric nitrate with corresponding chalcones **3a-3f** and Cliquinol **4**. The newly formed copper complexes as octahedral structure were evaluated *in vitro* antimycobacterial and antibacterial

activities. Among all the compounds, copper complexes of ligands 6-bromo-3-(3-(3-hydroxyphenyl)acryloyl)-2H-chromen-2-one **5c** and 6-bromo-3-(3-(4-hydroxyphenyl)acryloyl)-2H-chromen-2-one **5d** with Cliquinol had found good antibacterial and antimycobacterial agent at MIC value 25 $\mu\text{g/mL}$ each in compare to Isoniazid and Ethambutol (Patel et al., 2012).

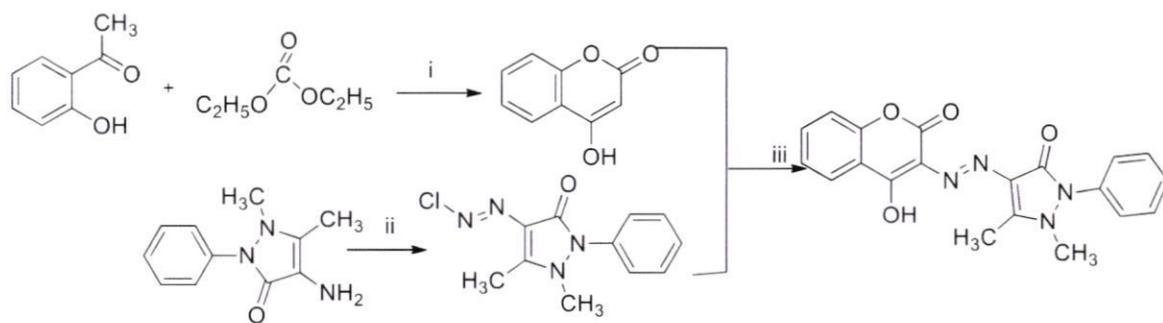


Scheme 32 Transitional metal complexes with coumarin derivatives.



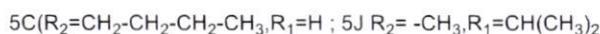
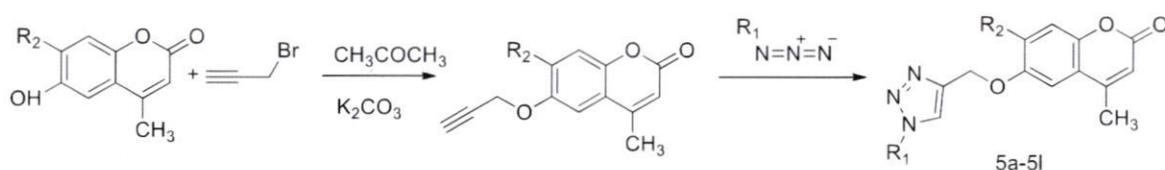
Reagents and conditions: i) NaH, toluene ii) NaNO_2/HCl , $0\text{-}5^\circ\text{C}$ iii) 10% NaOH

Scheme 33 3-Arylaazo-4-hydroxy coumarin derivatives.



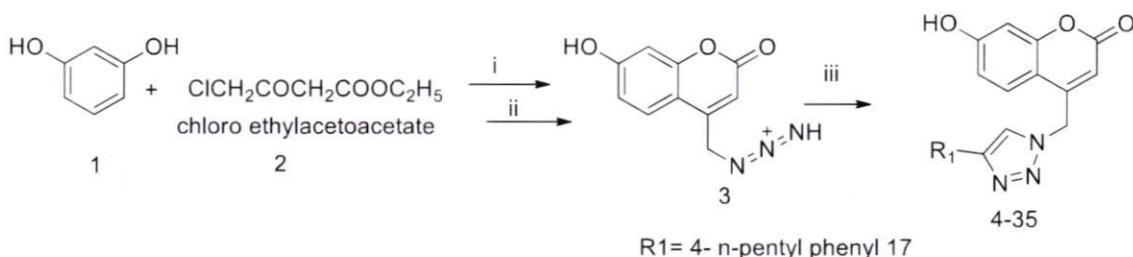
Reagents and conditions: i) NaH, toluene ii) NaNO_2/HCl , $0\text{-}5^\circ\text{C}$ iii) 10% NaOH

Scheme 34 Antipyrinyl azo-coumarin derivatives.



Reagents and conditions: i) CH_3COCH_3, K_2CO_3 ii) alkylazide, $CuSO_4 \cdot H_2O$, ascorbate

Scheme 35 1,2,3-Triazolyl substituted coumarin derivatives.



Reagents and conditions: i) $Conc, H_2SO_4, -5^\circ C$ ii) $NaN_3, CH_3CN, reflux$ iii) corresponding azide $1M CuSO_4$, tert.butanol: $H_2O(1:1)$, $DMF, 80^\circ C$

Scheme 36 Coumarin bearing triazole derivatives.

2.18. Synthesis of 3,4,7,8-tetra substituted coumarin derivatives

A series of biphenyl coumarin based bacterial helicase inhibitors of *B. anthracis* and *S. aureus* were designed and synthesized from substituted resorcinol. The title coumarin had been synthesized in three step reaction; in which initially 2-methyl/ethyl resorcinol was reacted with several β -ketoester provided 7-hydroxy coumarin carboxylate as an intermediate, which further underwent the hydrolysis of the desired ester, which provided the corresponding coumarin 3-carboxylic acid. The corresponding acids undergo alkylation by the treatment with either alkyl, aryl and alkenyl, or biphenyl halides in sodium carbonate and dimethyl formamide. These were further converted to respective amides by treatment with several amines. The desired amides were prepared from coumarin 3-carboxylic acid compounds. The compound 2-(7-([1,1'-biphenyl]-4-yloxy)-4,8-dimethyl-2-oxo-2H-chromen-3-yl)acetic acid had exhibited more significant activity against *B. anthracis* and *S. aureus* at MIC value 1.25 and 2.00 μM , respectively. SARs of these compounds indicate that the compounds having biphenyloxy with different substitution such as, fluoro, chloro, cyano and trifluoromethyl groups on the distal end phenyl ring would be responsible for antibacterial efficacy compared to unsubstituted phenyl ring (Li et al., 2012).

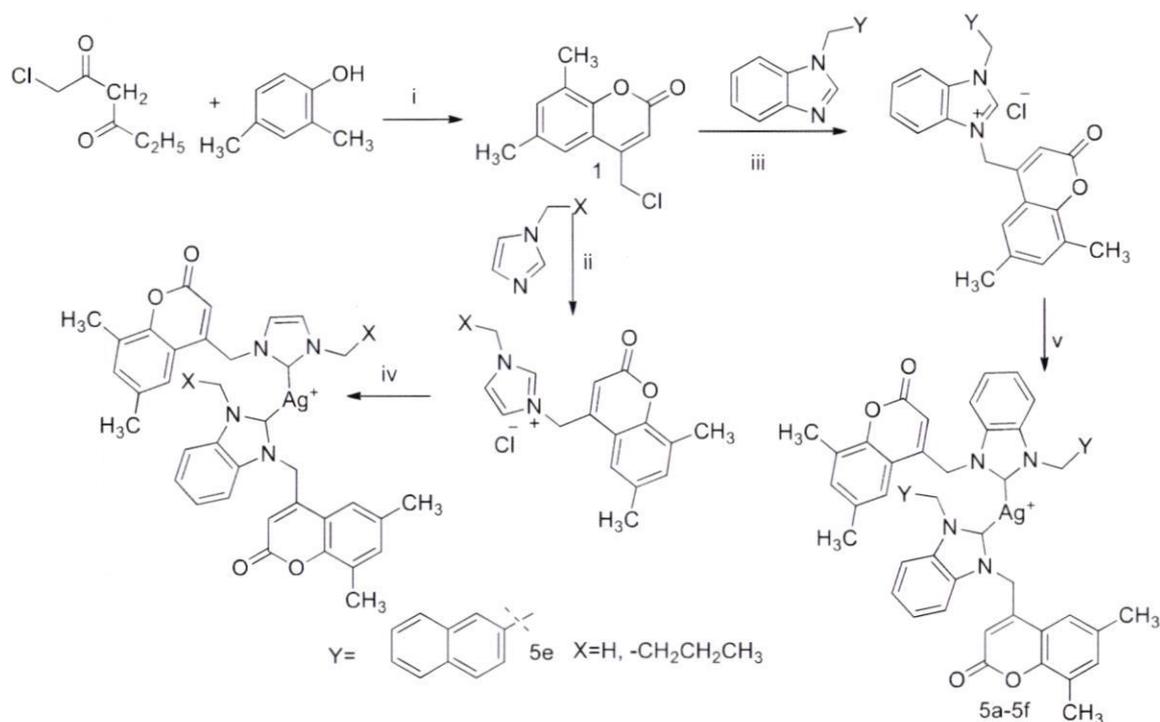
2.19. Synthesis of thiazolyl-pyrazoline coumarin derivatives

A novel series of coumarin compounds bearing thiazolyl and pyrazolone linked derivatives were synthesized by the reflux

condensation of alcoholic solution of 6-bromo 3-bromoacetyl coumarin **1** with another intermediate reactant 5-hydroxy-3,5-bis(trifluoromethyl)-4,5-dihydro-1H-pyrazole-1-carboxamide **2**. The intermediate trifluoromethyl pyrazolone carboxamide **2** was synthesized by the reaction of 2,2,2-trifluoroethyl 4,4,4-trifluoro-3-oxobutanoate with thiosemicarbazide. Moreover, the compound **6c** had notable activity against *B. subtilis* and *S. epidermidis* at 25 and 20 mm as ZOI in comparison to Cefixime (Aggarwal et al., 2013).

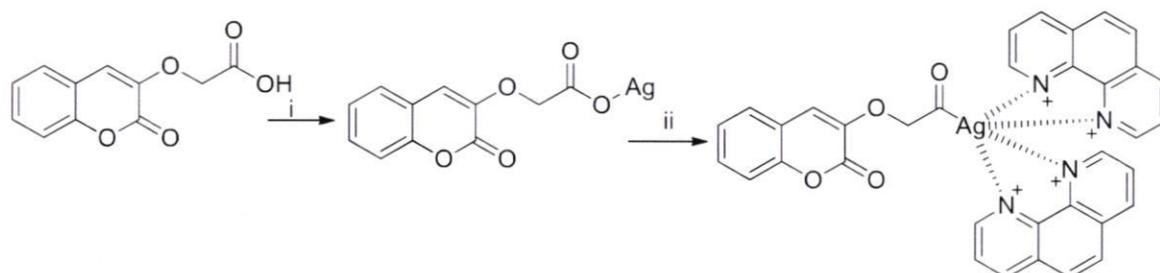
2.20. Synthesis of 4-aryloxymethyl coumarin derivatives

A series of potent 4-acyloxymethyl substituted coumarin derivatives were derived from the reaction 4-bromomethyl coumarin with methyl gallate, and the obtained hybridized products had good antibacterial activity. An organic synthesis 4-bromomethyl coumarin was obtained from brominated ethyl acetoacetate with various substituted phenols under cyclisation in the presence of condensing agent sulfuric acid, then it was reacted with methyl gallate in dry acetone and potassium carbonate to yield the desired gallate ether by involving nucleophilic displacement reactions. Thereafter, a monitored antibacterial activity result revealed that the compounds **3a-3j** were more active against *E. faecalis* and *S. aureus*. The compound ethyl 4-((7,8-dimethyl-2-oxo-2H-chromen-4-yl)methoxy)-3,5-dihydroxybenzoate **3h** was an effective congener against *E. faecalis* at MIC value 0.2 $\mu g/mL$ in comparison to Ciprofloxacin (Revankar et al., 2013).



Reagents and conditions: i) H₂SO₄ (70%), rt, 24h ii) DMF, 80°C, 24hr iii) DMF, 80°C, 24hr iv) 0.5Ag₂O, DCM, rt, 24h v) 0.5Ag₂O, DCM, rt, 24h

Scheme 37 Silver complexes with Heterocyclic substituted coumarin derivatives.



Reagents and conditions: i) NaHCO₃, H₂O, AgNO₃ ii) 1,10-phenanthroline, C₂H₅OH

Scheme 38 Silver complexes with coumarinyloxy phenoxy acetic acid derivatives.

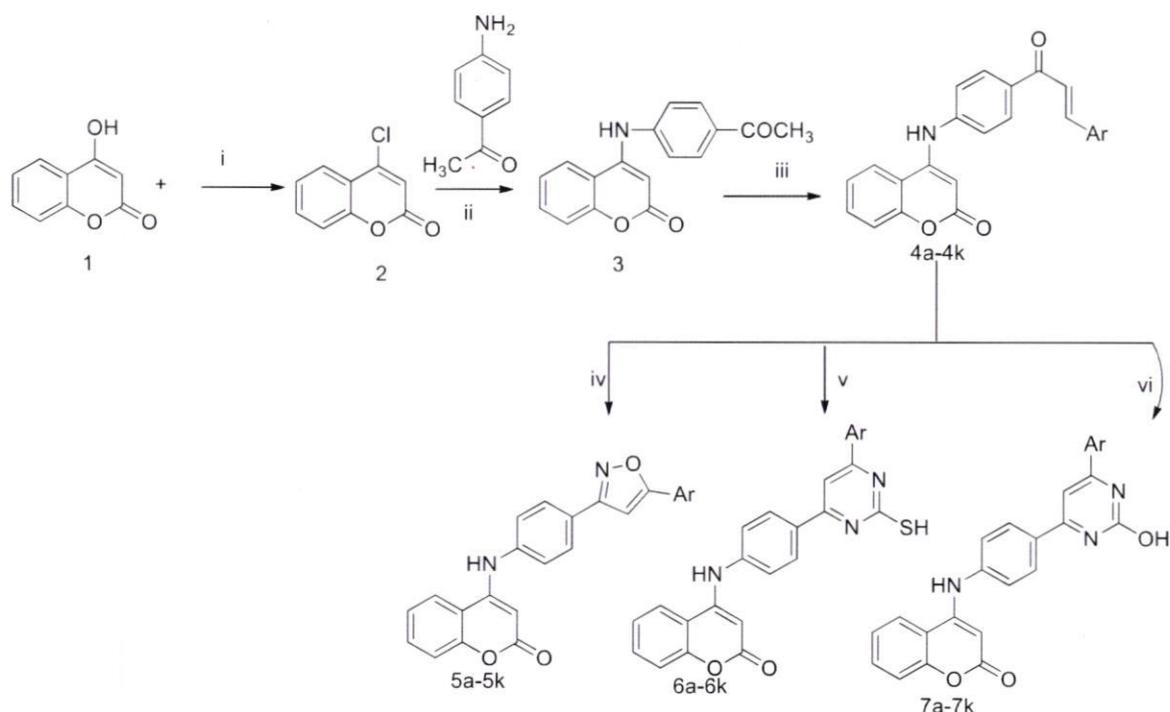
2.21. Synthesis of metal complexes of bis hydroxy coumarin with phenanthroline

A series cupric complexes containing bis-hydroxy substituted coumarin and 1,10-phenanthroline have been synthesized. The reactant ligand **1** was prepared from 4-hydroxy coumarin by treatment with a substituted aryl aldehyde in ethanol with the addition of sulfuric acid as catalyst. The desired copper complexes **HPC1-HPC6** were prepared by the mixing of bis-hydroxy substituted coumarin and 1,10-phenanthroline with cupric nitrate in ethanol. The compounds **HPC2** and **HPC3** had shown a comparatively good *in vitro* controlling activity against *S. pyogenes*. Concomitantly, the compound **HPC2** had remarkable inhibitory action against *M. tuberculosis* at

the MIC value, 3.125 µg/mL in comparison to Ciprofloxacin, Streptomycin, Isoniazid and Ethambutol as standards (Dholariya et al., 2013).

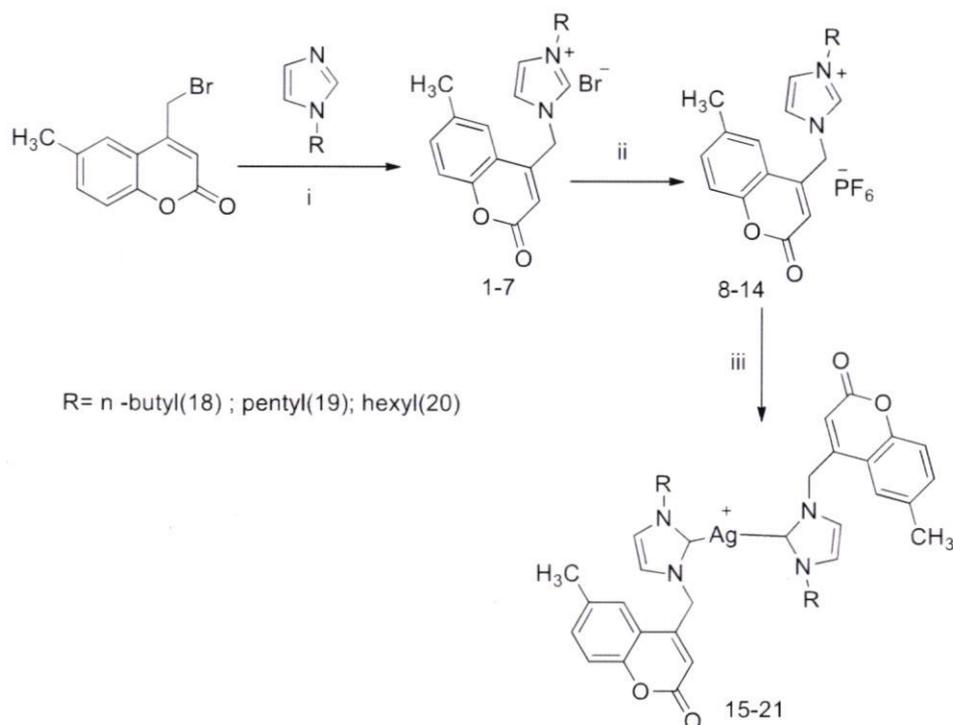
2.22. Synthesis of bis-4-hydroxy coumarin derivatives

The bis 3,3'-4-hydroxy coumarin moiety was prepared for the Bis-4-hydroxy coumarin derivative which was screened for antibacterial activity against methicillin sensitive *S. aureus* and MRSA. The compound 3,3'-(*m*-tolyl methylene)bis(4-hydroxy-2*H*-chromen-2-one) **3-MBH** had a notable bactericidal action against *S. aureus* at the MIC value 64 µg/mL. The compound **3-MBH** was synthesized by the condensation of 4-hydroxy coumarin with 3-methyl benzaldehyde(*m*-



Reagents and conditions: i) POCl_3 , reflux, 1h ii) CH_3COCH_3 , reflux, 5-8h iii) ArCHO , DMF/KOH , rt, 24h
iv) $\text{NH}_2\text{OH HCl}$ v) NH_2CSNH_2 vi) NH_2CONH_2

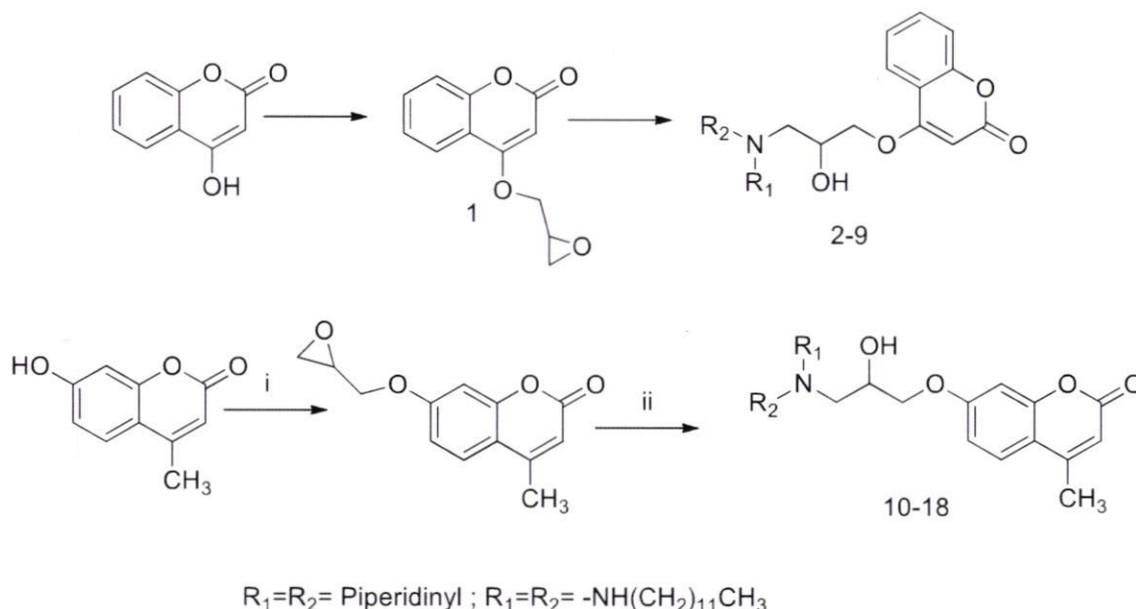
Scheme 39 4-Amino coumarin derivatives.



$\text{R} = n$ -butyl(18); pentyl(19); hexyl(20)

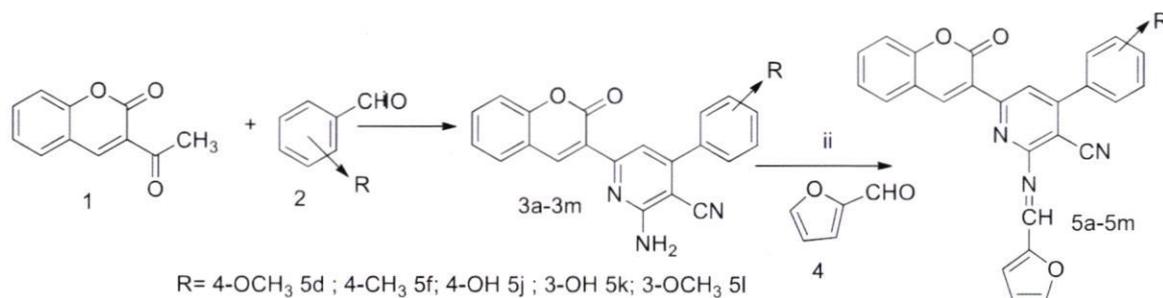
Reagents and conditions: i) 1,4-dioxane, 85°C , 24h ii) KPF_6 , $\text{CH}_3\text{OH}/\text{H}_2\text{O}$, 3-4h, rt, iii) Ag_2O , CH_3CN , 45°C , 24h

Scheme 40 Imidazolium methyl coumarin silver complexes.



Reagents and conditions: i) Epichlorohydrine, anhydrous K_2CO_3 , reflux, 4h ii) corresponding amines, $\text{C}_2\text{H}_5\text{OH}$, 3-5h

Scheme 41 7-Coumarinyloxy amino propanol derivatives.



Reagents and conditions: i) $(\text{CH}_2)_2\text{CN}$, NH_4OAc , $\text{C}_2\text{H}_5\text{OH}$ (99%) 350W, 130°C , 6-10mins
ii) CH_3COOH , ZnCl_2 , 100°C , 200W, 8-10mins

Scheme 42 3-Heteroaryl substituted coumarin derivatives.

tolualdehyde in ethanol, then those compounds converted to 3-MDH by cyclization under the basic piperidine in ethanol (Li et al., 2013).

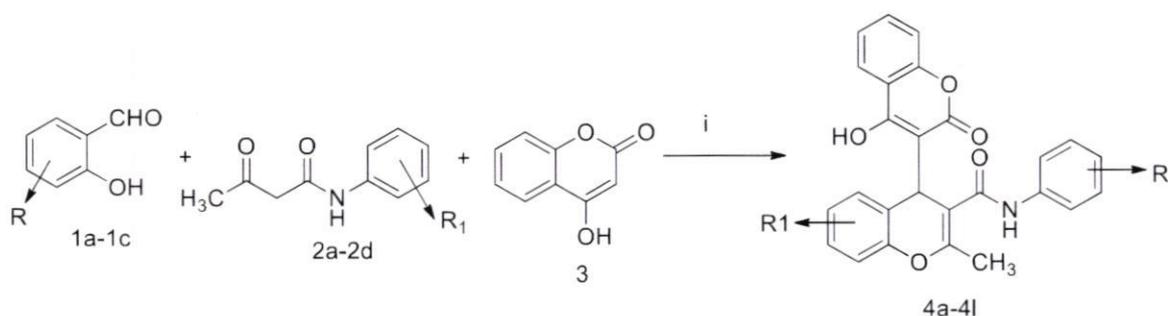
2.23. Synthesis of chromeno[2,3-d]pyrimidinone with coumarin derivatives

In this scheme, the coumarin-linked scaffold, chromeno[2,3-d]pyrimidinone 6a-6j had been synthesized by catalytic condensation of ethyl 6-amino-5-cyano-2-methyl-4-subst.phenyl-4H-pyran-3-carboxylate with coumarin 3-carboxylic acid **5**, in the presence of organocatalyst 5 mol of pentafluoro ammonium triflate (PFPA). In this reaction, the intermediate reactant **1** was synthesized by the reflux condensation of ethyl acetoacetate with substituted aryl aldehyde 2a-2j and malonic dinitrile in ethanolic of ZrOCl_2 known as Bignelli reactions.

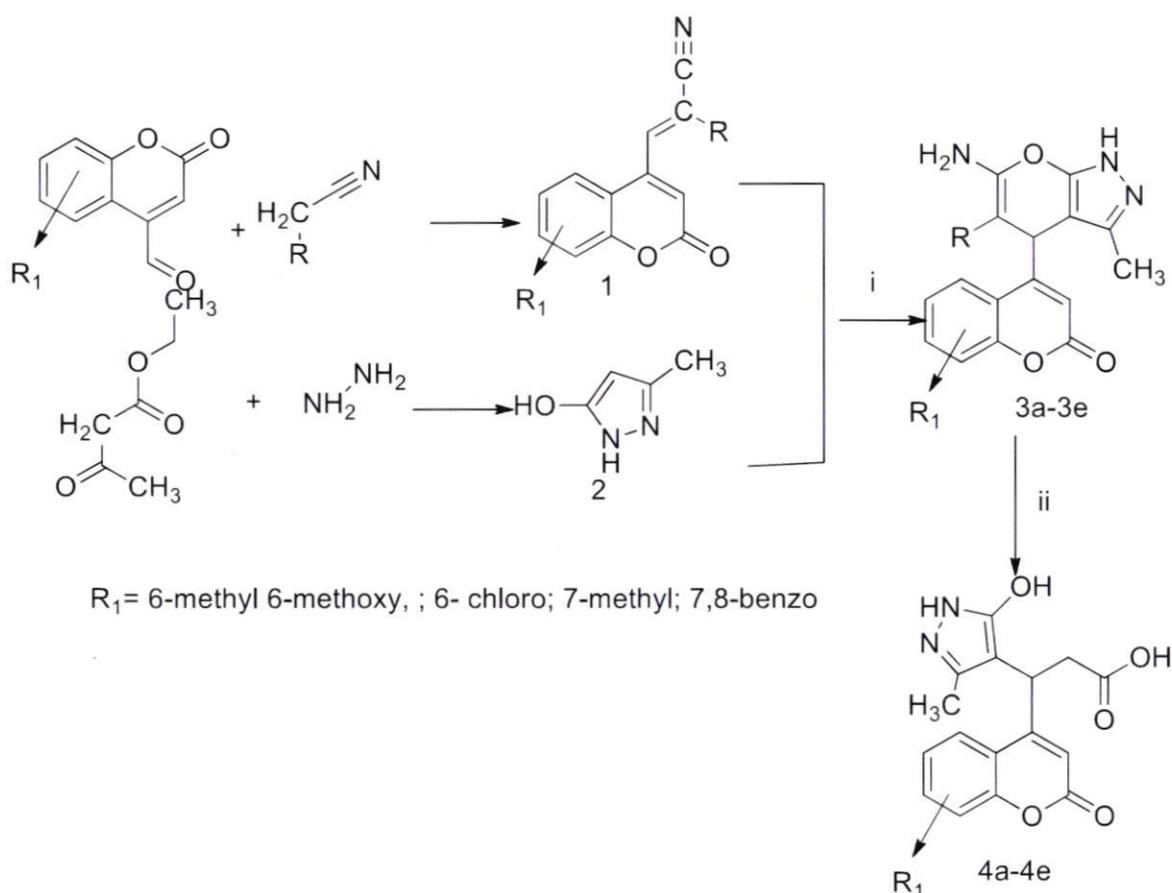
The compound ethyl 5-(4-bromophenyl)-7-methyl-4-oxo-2-(2-oxo-2H-chromen-3-yl)-4,5-dihydro-3H-pyranopyrimidine-6-carboxylate **6i** had been reported against *E. coli* at the MIC value 6.25 $\mu\text{g/mL}$ with the inhibitory zone 18 to 20 mm; concomitantly, ethyl 7-methyl-4-oxo-2-(2-oxo-2H-chromen-3-yl)-5-(p-tolyl)-4,5-dihydro-3H-pyranopyrimidine-6-carboxylate **6j** had shown control against both *E. coli* and *S. aureus* at MIC value 6.25 and in comparison to Ciprofloxacin as the standard (Ghashang et al., 2014).

2.24. Synthesis of silver salt of benzimidazolium with coumarin derivatives

A series of steric modulate chlorocoumarin derivative containing imidazolium/benzimidazolium salts and their bis-silver complexes were synthesized. Initially, the reaction of *N*-alkyl



Scheme 43 3-Chromenyl carboxamide of coumarin derivatives.

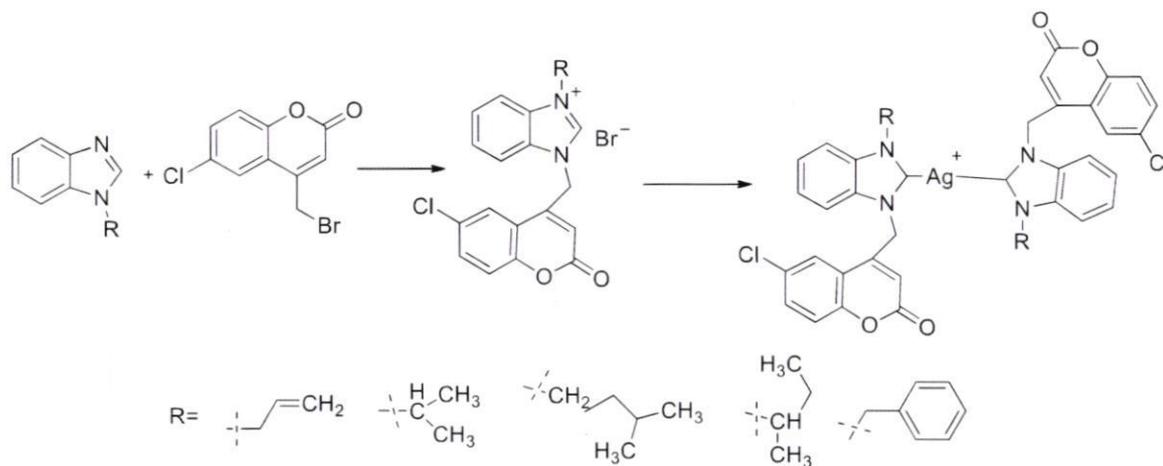


Reagents and conditions: i)Base, C₂H₅OH.(20%), 2-3h, stir. ii)HCOOH or AcOH/Ac₂O

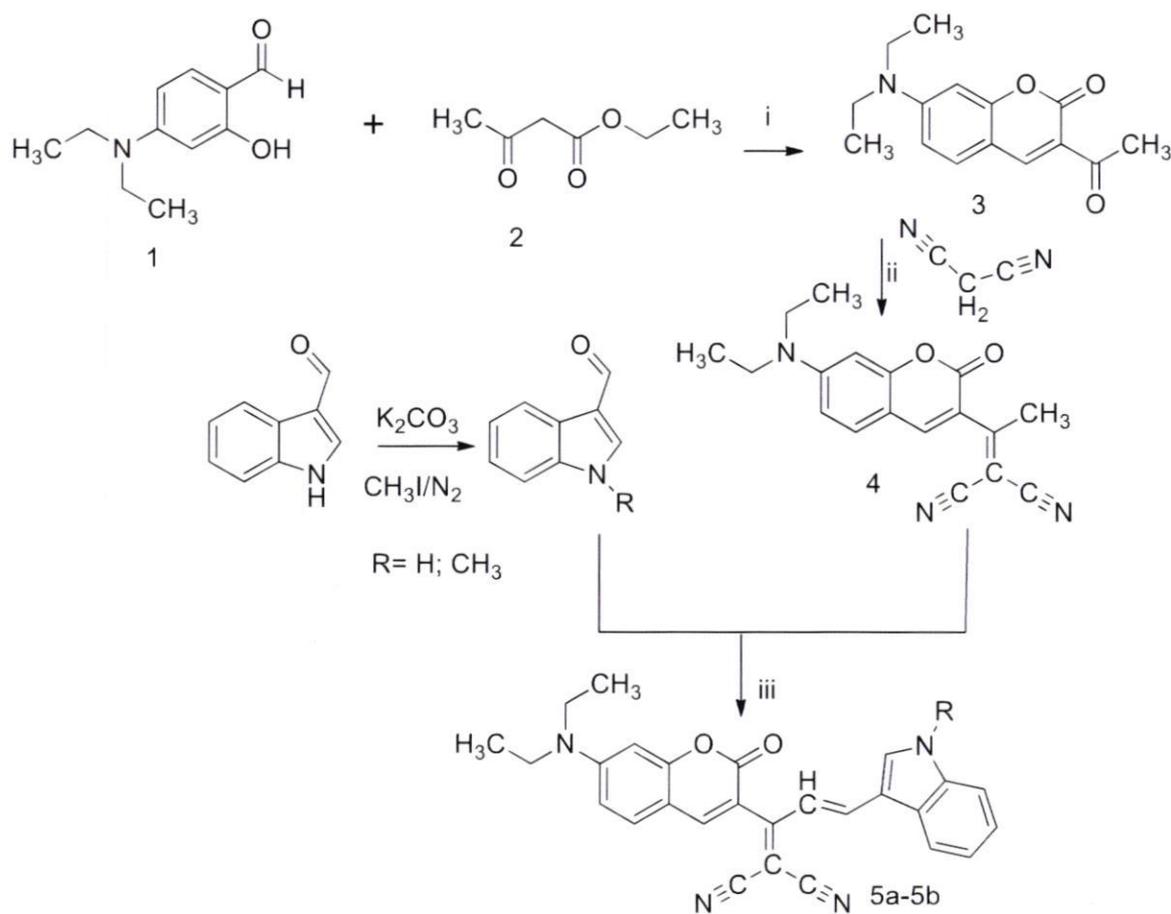
Scheme 44 Coumarin based pyano[2,3-c]pyrazole derivatives.

benzimidazolium with 4-bromomethyl-6-chlorocoumarin in 1,4-dioxane solvent, which further liberated bromide salt; this was reacted with phosphorous hexafluorophosphate to get the corresponding 1-alkyl-3-(6'-chloro-4'-methylene coumarin) imidazolium hexafluorohexaphosphate **2**. Thereafter, the silver(I) complexes were prepared by the reaction (benz)imidazolium salt with moist silver oxide in acetonitrile solution, stirred at

45 °C at 24 h. The liberated silver(I) complexes NHC were recrystallised from acetonitrile. these complexes with mixture of salts were evaluated antibacterial activity. The silver(I) complex bearing n-butyl (benz)imidazolium chloro coumarin had shown a good antibacterial agent against *E. coli* and *P. aeruginosa* at MIC value 4 µg ml⁻¹ individually in comparison to standard drug Ampicillin (Achar et al., 2017a).

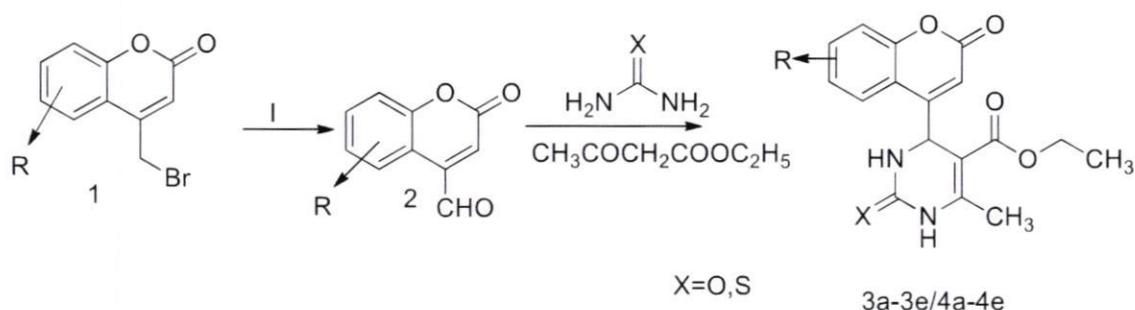


Scheme 45 Silver complexes of bis benzimidazolium methyl with coumarin.



Reagents and conditions: i) Piperidine, C₂H₅OH ii) CH₂(CN)₂, NH₄OAc, benzene, CH₃COOH
iii) piperidine, C₂H₅OH

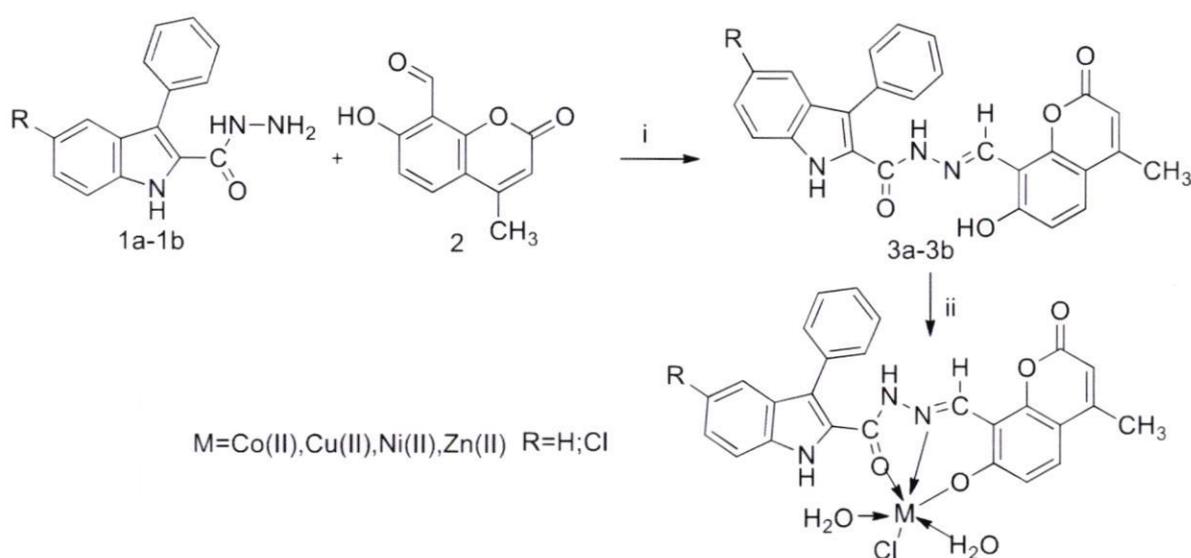
Scheme 46 Indolidenyl of coumarin derivatives.



R= 6-methyl; 6-methoxy; 6-chloro; 7-methyl, 7,8-benzo

Reagents and conditions: i) DMSO, Na₂CO₃(aq), 110°C, 6-16h ii) CAN, C₂H₅OH

Scheme 47 Pyrimidinone bearing coumarin derivatives.



Reagents and conditions: i) CH₃COOH, C₂H₅OH ii) CuCl₂, CH₃OH, Na₂CO₃

Scheme 48 Indole carboxahydrazide Schiff base coumarin derivatives.

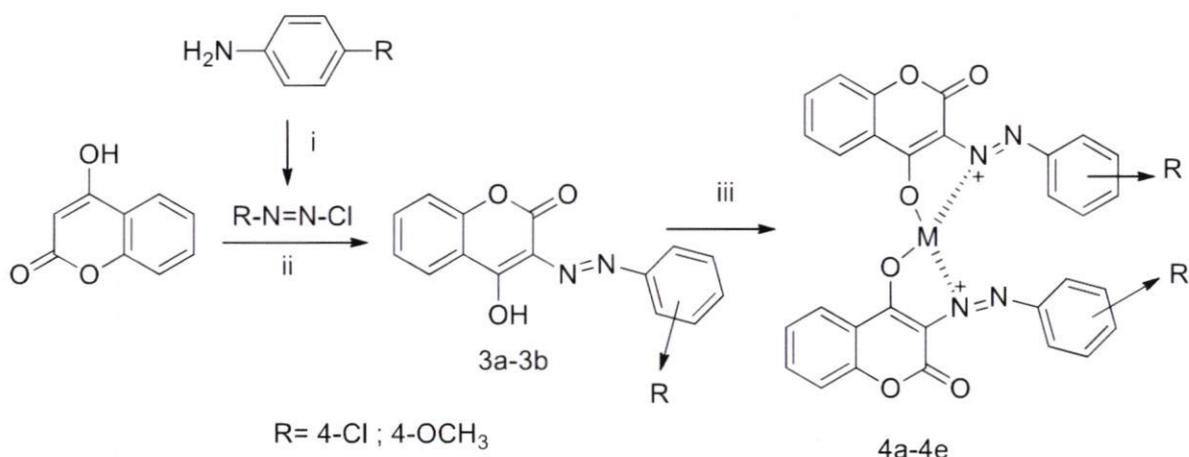
2.25. Synthesis of coumarinyl piperazine bearing propanol derivatives

A series of coumarin bearing piperazinyl derivatives 4a-4n were synthesized. The obtained compounds were tested for antibacterial activity against both Gram + and Gram- bacteria, *S. aureus*, *B. subtilis*, *E. coli* and *P. Aeruginosa* and those were potent antibacterial agents. Initially, the 4-hydroxycoumarin was reacted with epichlorohydrin in dry acetone and potassium carbonate yield 4-(oxiran-2-ylmethoxy)-2H-chromen-2-one **1**, which further reacted with 1-(4-substituted phenyl)piperazine in dimethyl formamide and potassium carbonate to produce the desired compounds. The compound 4-(2-hydroxy-3-(4-(4-methoxyphenyl)piperazin-1-yl)propoxy)-2H-chromen-2-one **4g** had shown a good antibacte-

rial activity against *S. aureus* and *B. subtilis* at the MIC value 0.23 and 0.35 µg/mL in comparison to Penicillin G as the standard (Wang et al., 2014).

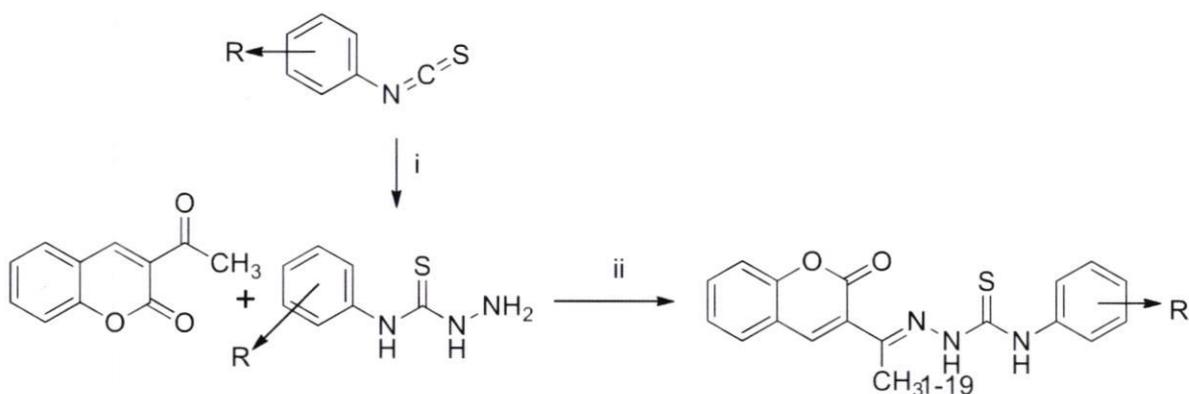
2.26. Synthesis of dicoumarol derivatives

Dichromene derivatives were synthesized by the intermediate 3,3'-((4-(di-p-tolylamino)phenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) DBT and the condensation of equimolar of ethanolic solution of 4-hydroxy coumarin; 4-(di-4-tolyl-amino) benzaldehyde in presence of piperidine, which further undergoes cyclisation in acetic anhydride yield 7-(4-(di-p-tolylamino)phenyl)-6H-pyrano[3,2-c:5,6-c']dichromene-6,8 (7H)-dione DDT. The synthesized two bis hydroxy coumarin derivatives were evaluated against *S. aureus* and MRSA with



Reagents and conditions: i) NaNO₂/HCl, 0-5°C ii) 10% NaOH, iii) MCl₂. x.H₂O, CH₃OH

Scheme 49 Metal complexes 3-Aryazo 4-hydroxy coumarin.



Reagents and conditions: i) NH₂NH₂ ii) CH₃COOH, C₂H₅OH, reflux, 70-80°C 3-4h

Scheme 50 Coumarinyl- 3-semithiocarbazone congeners.

subsequent MIC values 32 and 64 µg/mL, in comparison to Levofloxacin, Cefazidime, Ceftriaxone and Gentamicin as the standard (Li et al., 2014).

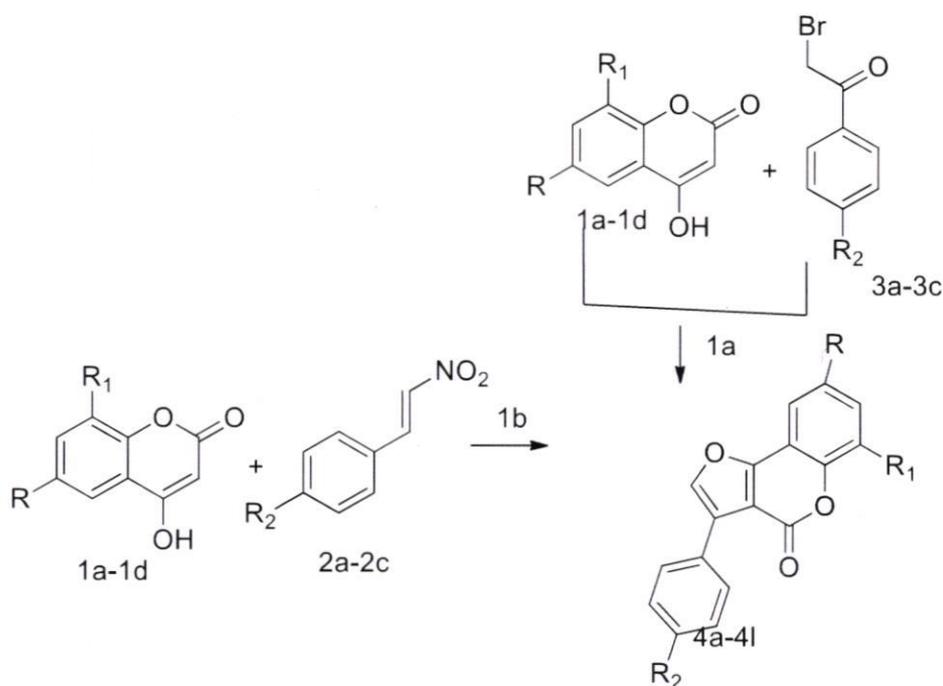
2.27. Synthesis of metal complexes thiophene hydrazone - coumarin Schiff base

A series of transitional divalent metal ion complexes were synthesized from Schiff base ligand named as 3-chloro-*N*'-(7-hydroxy-4-methyl-2-oxo-2*H*-chromen-8-yl)methylene)benzo[b]thiophene-2-carbohydrazide **3**, which was synthesized by equimolar mixture of methanolic solution of 8-formyl-7-hydroxy-4-methyl coumarin **2** and 3-chloro-benz[b]thiophene 2-carbohydrazide **1** with a few drops of glacial acetic acid. Then, finally metal complexes were obtained by the mixture of hot alcoholic solution of intermediate ligand **3** with respec-

tive metals chlorides. The obtained complexes were characterised by different spectral studies. Based spectral studies, the chelating ability of the ligand had been confirmed in complexation with Cu(II), Ni(II), Co(II) and Zn(II) ions whereas, ligands acted as ONO electron donor tridentate chelate to metal ions. During the complexation reactions, deprotonation was carried out at phenolic hydroxyl group of the ligand structure (Mahendra Raj et al., 2014).

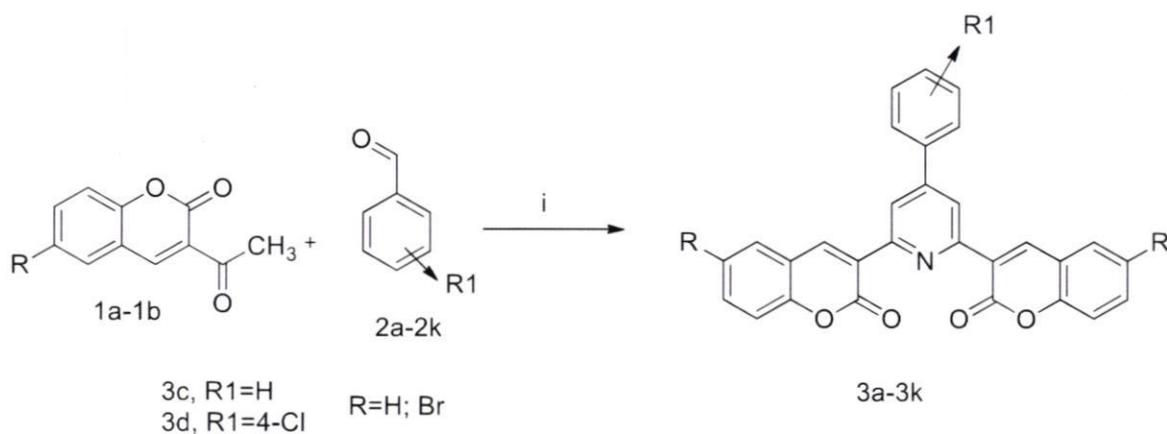
2.28 Synthesis of indolin-2,3-dione - coumarin derivatives

Linked through ethylene, a series of Isatin-coumarin compounds, **4a-4b** were synthesized and screened their antimycobacterial action against *M. tuberculosis* and multidrug resistant tuberculosis. The targeted compounds **3a-3b** were synthesized from two intermediate reactant named as 7-



Reagents and conditions: 1a) NH_4OAc , CH_3COOH , MW 1b) piperidine CH_3OH , MW

Scheme 51 Coumarin fused furan derivatives.



Reagents and conditions: CH_3COOH , NH_4OAc , reflux, 8-10h

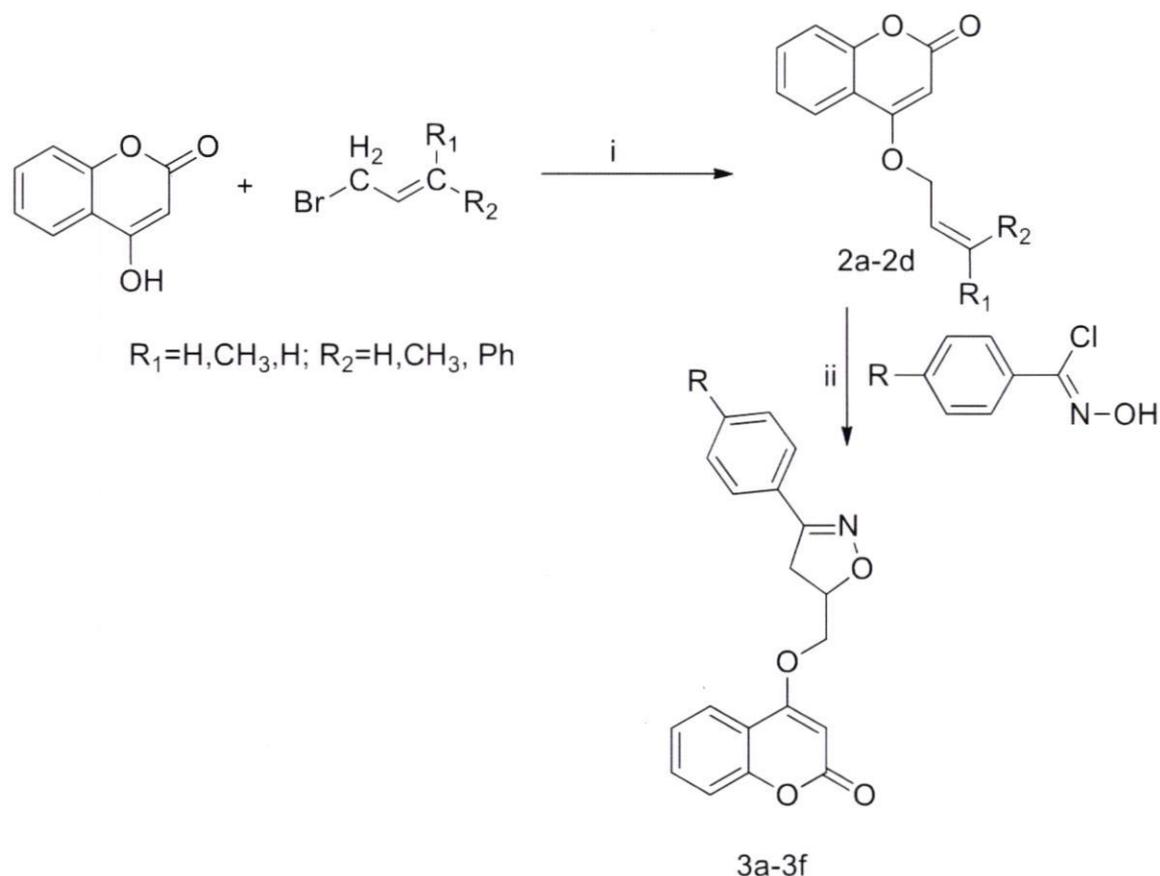
Scheme 52 Bis 3,3'- Coumarin derivatives.

hydroxy-4-methyl coumarin **1** and 1-bromomethyl 5-substituted isatin **2**, which were mutually reacted in the presence of dimethyl formamide (DMF) and potassium carbonate. From the converted to 3-(alkoxyimino)-1-(2-((4-methyl-2-oxo-2H-chromen-7-yl)oxy)ethyl)indolin-2-one **4a-4b** by treated with methoxyamine or ethoxyamine hydrochloride in sodium carbonate solution, respectively. The reactant 7-hydroxy-4-methyl coumarin was prepared by the Pechmann condensation of resorcinol and ethyl acetoacetate using polyphosphoric acid, whereas another product was alkylated of substituted isatin with dibromoethane in DMF. These tethered isatin-coumarin

compounds were evaluated for their antibacterial actions. The compounds **4a** and **4b** were reported as good antibacterial agent against *M. tuberculosis* at MIC 50 $\mu\text{g}/\text{mL}$ in comparison to standard drugs Isoniazid and Rifampicin, respectively (Gao et al., 2018).

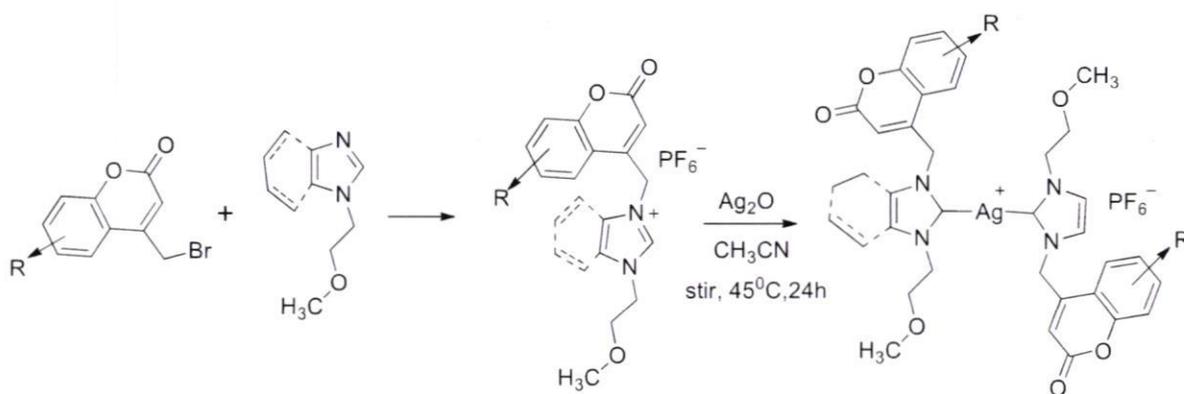
2.29. Synthesis of metal complexes of acetyl coumarin derivatives

The new Cu(II) complexes with 6-acetyl-7-hydroxy 4-methylcoumarin and 8-acetyl-7-hydroxy 4-methylcoumarin



Reagents and conditions: i) $\text{CH}_3\text{COCH}_3, \text{K}_2\text{CO}_3$ ii) toluene, $(\text{C}_2\text{H}_5)_3\text{N}$

Scheme 53 Coumarinyloxy bearing oxazoline derivatives.

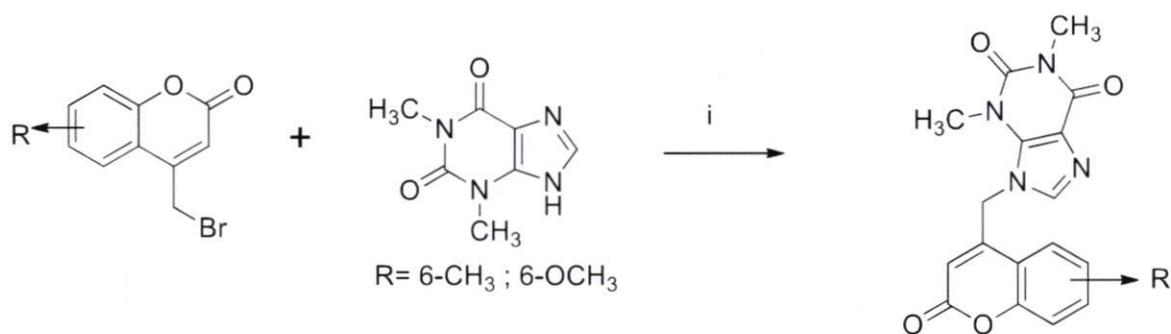


Scheme 54 Silver complexes with imidazolium methyl coumarin derivatives.

had been prepared by electrochemical method. During the reaction, the two bidentate coumarin ligands bind to $\text{Cu}(\text{II})$ through the acetyl group and deprotonation of phenolic hydroxy group. Among all the tested compounds, the complex with acetyl at C6 coumarin **2** had been reported as a good antibacterial agent against *Micrococcus luteus* at MIC value 0.017 mg/mL (Klepka et al., 2015).

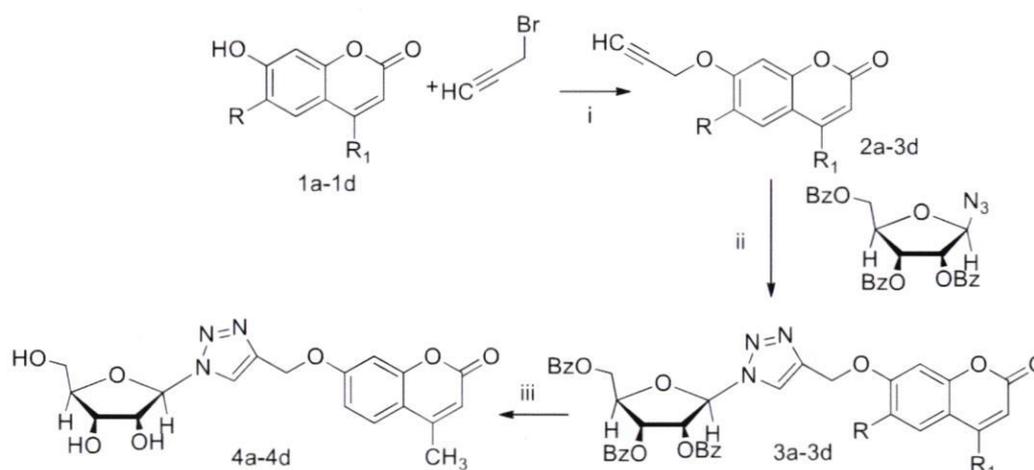
2.30. Synthesis of arylchalcone of coumarin derivatives

A series of aryl chalcone hybrids derived from coumarin (**5–12**) had been synthesized efficiently with moderate yield through the principle of knowingly reaction treating the appropriate salicylaldehyde and respective ethyl sub-



Reagents and conditions: i) Activated K₂CO₃, CH₃COCH₃, rt, 6-8h 3a-3j

Scheme 55 7- Coumarinyl methyl theophylline derivatives.



Reagents and conditions: i) K₂CO₃, CH₃COCH₃, 8-10h, reflux, ii) sodium Ascorbate, CuSO₄ tert.butanol/H₂O, 10-15h, rt iii) NaOCH₃, CH₃OH, 5-6h, rt

Scheme 56 Ribofuranosyl-coumarinyloxy bearing 1,2,3- triazole derivatives.

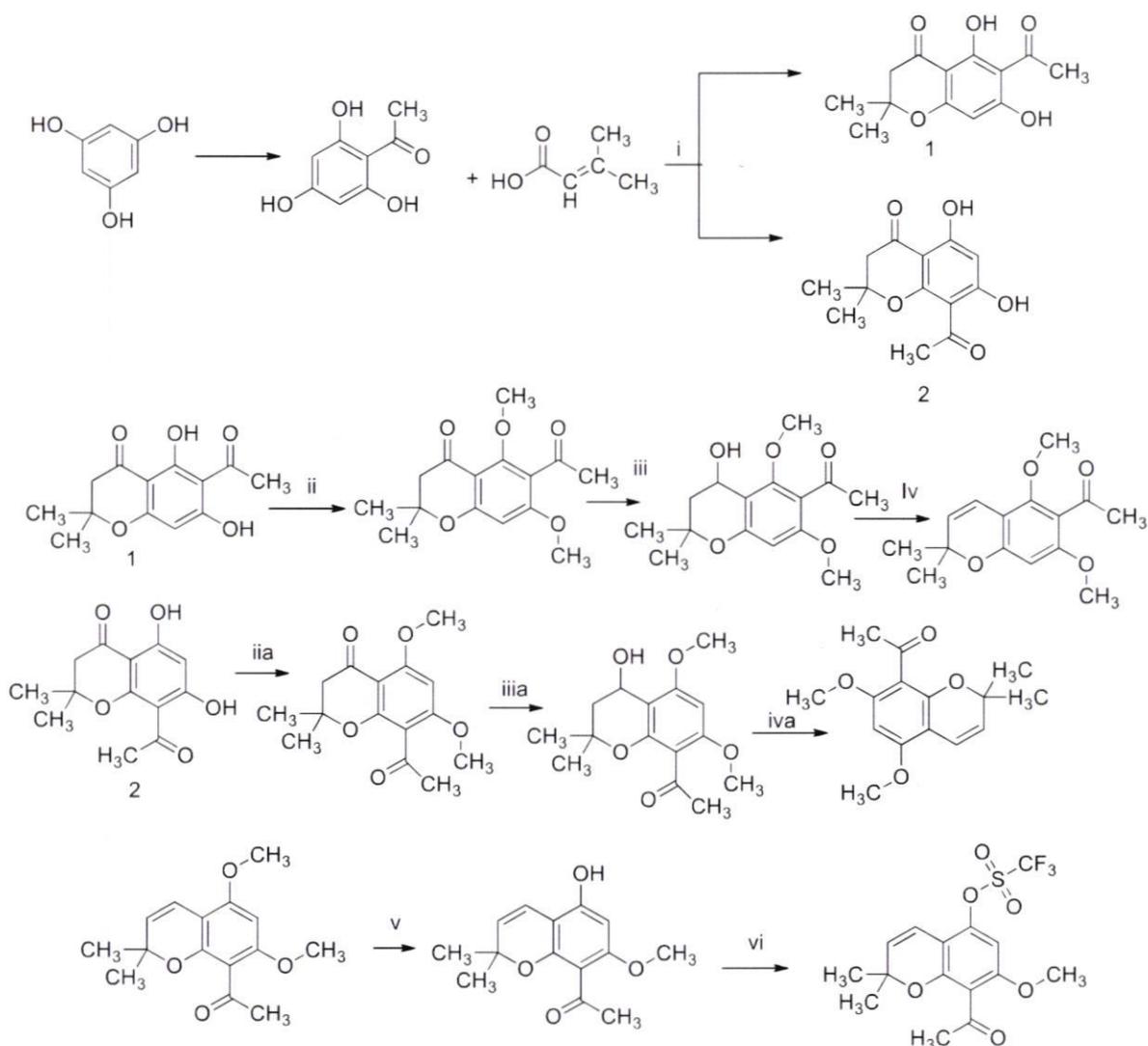
stituted benzoyl acetate with piperidine in ethanol under reflux for 2–5 h, which further were synthesized nitro substituted of benzoyl coumarin derivative from intermediate precursor undergone nitration and subsequently reduction of these derivatives with stannous chloride yield corresponding amino substituted coumarin chalcones. All these compounds had been tested for their antibacterial activity against seventeen marine gram negative bacterial strains including different species of *Tenacibaculum*. The compounds with amino substituted coumarin chalcones such as, (*E*)-3-(3-(3-amino-4-methoxy phenyl)acryloyl)-2H-chromen-2-one **6**, (*E*)-3-(3-(3-amino-4-ethoxyphenyl)acryloyl)-2H-chromen-2-one **7**, (*E*)-3-(3-(4-aminophenyl)acryloyl)-8-methoxy-2H-chromen-2-one **10** and (*E*)-3-(3-(4-aminophenyl)acryloyl)-8-ethoxy-2H-chromen-2-one **11** had been reported as antibacterial agent against *Tenacibaculum maritimum* in comparison to Oxolinic A, Enrofloxacin and Ampicillin as standards (Vazquez-Rodriguez et al., 2015).

2.31. Synthesis of a metal chelate of Schiff base with coumarin derivatives

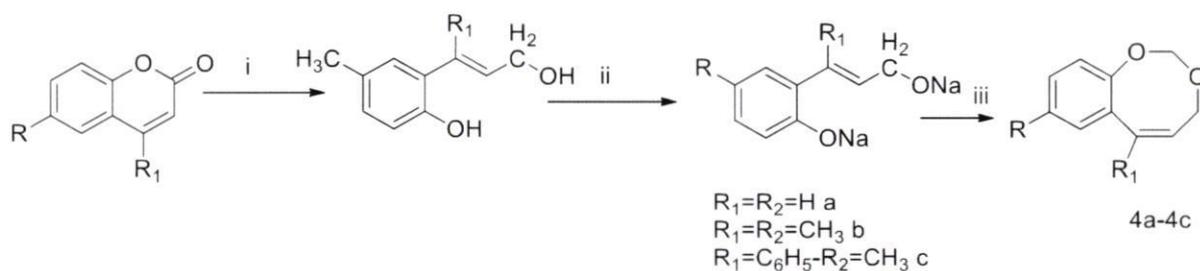
A new series of Cu(II), Ni(II) and Co(II) complexes had been synthesized with Schiff base (HL), which was derived from 8-formyl-7-hydroxy-4-methyl coumarin with benzylamine. The coordination of metal ions to the ligand through phenolic hydroxy group of coumarin and nitrogen of azomethine groups were prepared by the principle of Schiff bases. The obtained compound **3** had been reported as having good antibacterial control against *E. coli*, *P. aureginosa*, *K. pneumonia* and *S. aureus* (Prabhakara et al., 2015).

2.32. Synthesis of transitional metal complexes with coumarin derivatives

A series of Cu(II), Ni(II) and Co(II) complexes of Schiff base coumarin derivatives had been synthesized from 6-

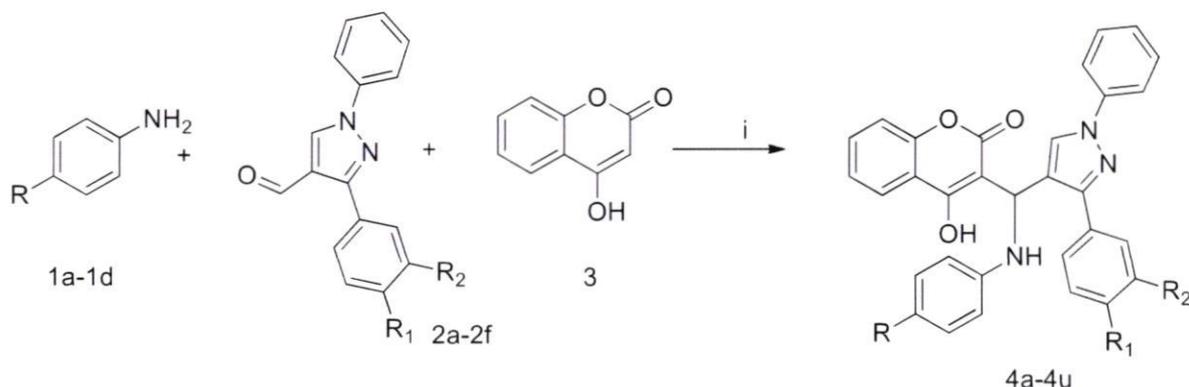


Scheme 57 Disubstituted of chromane derivatives.

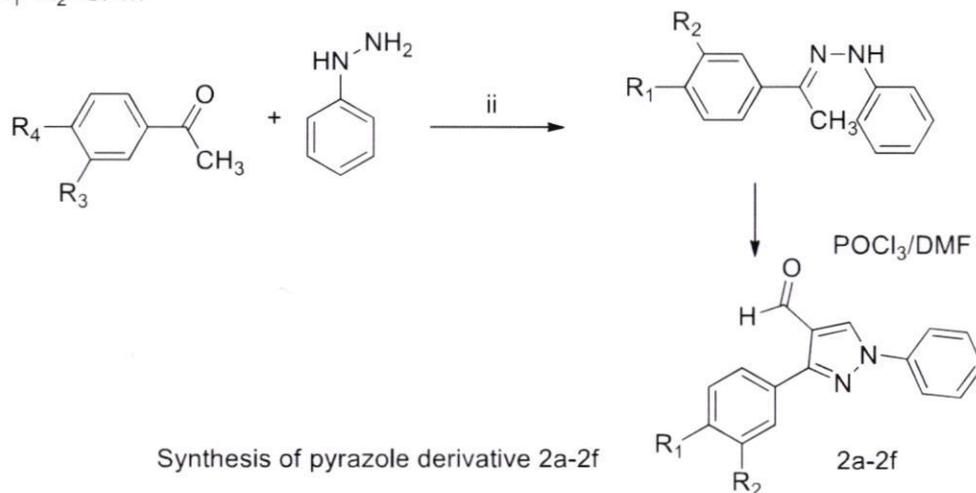


Reagents and conditions: i) $LiAlH_4$, dry ether ii) 1N NaOH iii) CH_2I_2 , dry ethyl acetate

Scheme 58 Coumacine derivatives.



R=4-CH₃, R₁=4-Cl 4b; R=OCH₃, R₁=Cl 4c; R=CH₃, R₁=NO₂ 4h; R=OCH₃, R₁=NO₂ 4i; R=OCH₃, R₁=R₂=Cl 4k



Synthesis of pyrazole derivative 2a-2f

Reagents and conditions: i) CH₃OH, reflux, 5h ii) CH₃OH, CH₃COOH, rt, 6h

Scheme 59 Pyrazole-anilino connected coumarin derivatives.

formyl, 7,8-dihydroxy coumarin with either 3-trifluoromethyl aniline or *o*-toluidine. The Schiff base ligands had been reported as an intermediate of 7,8-dihydroxy-4-methyl-6-(((3-(trifluoromethyl)phenyl)imino)methyl)-2*H*-chromen-2-one HL1 and 7,8-dihydroxy-4-methyl-6-((*o*-tolylimino)methyl)-2*H*-chromen-2-one HL2. The compound with copper metal ion complexed of 7,8-dihydroxy-4-methyl-6-(((3-(trifluoromethyl)phenyl)imino)methyl)-2*H*-chromen-2-one had shown as good antibacterial agent against *Klebsiella* sp. at 12 mm and MIC value 100 µg/mL in comparison to Gentamicin as the standard (Patil et al., 2015).

2.33. Synthesis of 3-aryloxy 4-hydroxy coumarin derivatives

A series of 3-aryloxy 4-hydroxy coumarin derivatives were obtained by azo coupling reaction of substituted aryl diazonium salt including antipyrine with 4-hydroxy coumarin. The hydroxy coumarin analogues, 4-hydroxy-3-((4-methoxyphenyl)diazonyl)-2*H*-chromen-2-one **4a**, 4-hydroxy-3-((3-nitrophenyl)diazonyl)-2*H*-chromen-2-one **4b** had exhibited potential antibacterial activities against bacterial strains with MIC 31.25 µg/mL whereas the compound (4-((4-hydroxy-2-oxo-2*H*-chromen-3-yl)diazonyl)-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one **4c** had as good antibacterial agent against *K. pneumoniae* at 26.50 mm inhibitory zone (Sahoo and Sudhir Kumar, 2015).

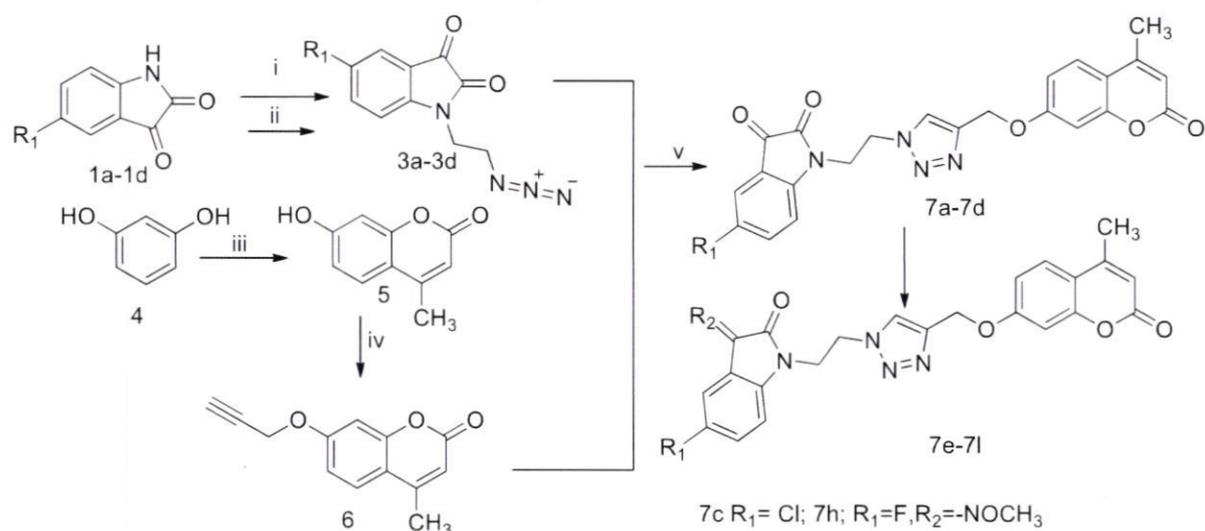
2.34. Synthesis of antipyrinyl azo-coumarin derivatives

2.34. Synthesis of antipyrinyl azo-coumarin derivatives

A series of 3-hetaryloxy coumarin was prepared from 4-hydroxycoumarin by azo coupling reaction by several heteroaryldiazonium salts reacted with the coupling component 4-hydroxy coumarin in NaOH solution, the obtained azo coumarin analogues were evaluated *in vitro* for antibacterial studies by agar diffusion method. The compound (4-((4-hydroxy-2-oxo-2*H*-chromen-3-yl)diazonyl)-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one **4e** had a good antibacterial activity against *S. aureus* as the ZOI, 25 mm with the MIC value 31.25 µg/mL in comparison to Ampicillin (Sahoo et al., 2015).

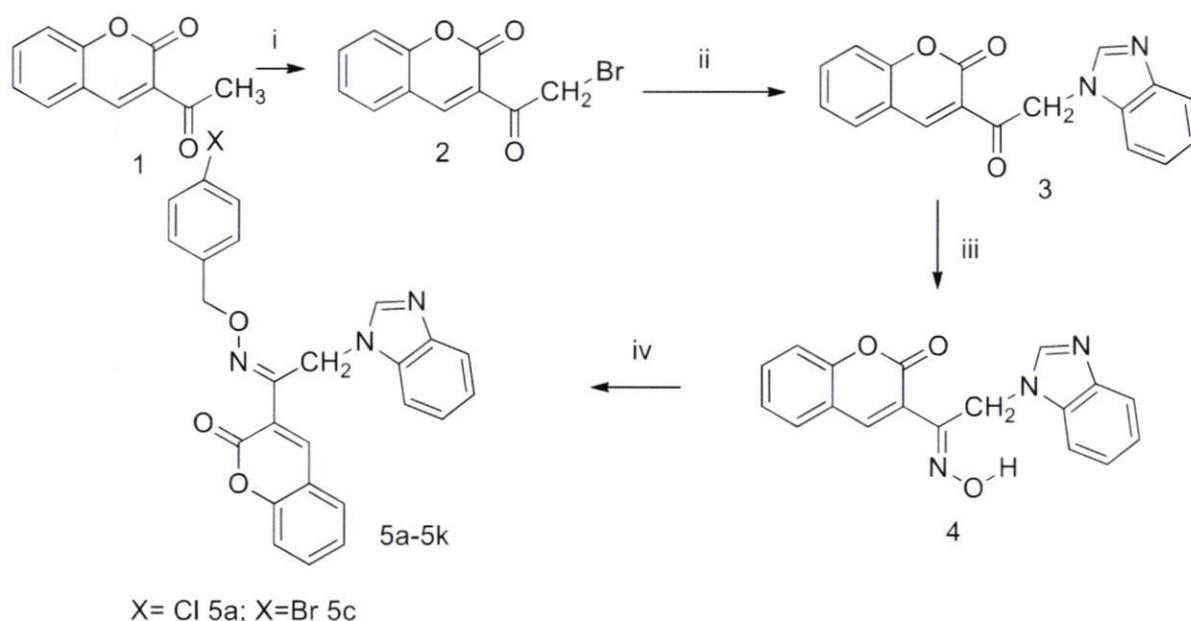
2.35 Synthesis of 1,2,3-triazolyl substituted coumarin derivatives

A series of twelve triazolylmethoxy substituted alkyl coumarin analogues, **5a-5 l** were synthesized by reacting with



Reagents and conditions: i) dibromoethane, K₂CO₃, DMF, rt ii) NaN₃, K₂CO₃, DMSO, 60°C iii) ethyl acetoacetate, conc. H₂SO₄, 100°C, 2h iv) propargyl bromide, K₂CO₃, DMF, 50°C v) Cu(OCOCH₃)₂, DMF, rt, 6h vi) RNH₂, NaHCO₃, THF/H₂O, 60°C, 12h

Scheme 60 Isatin-linked 1, 2, 3-triazole with coumarin.

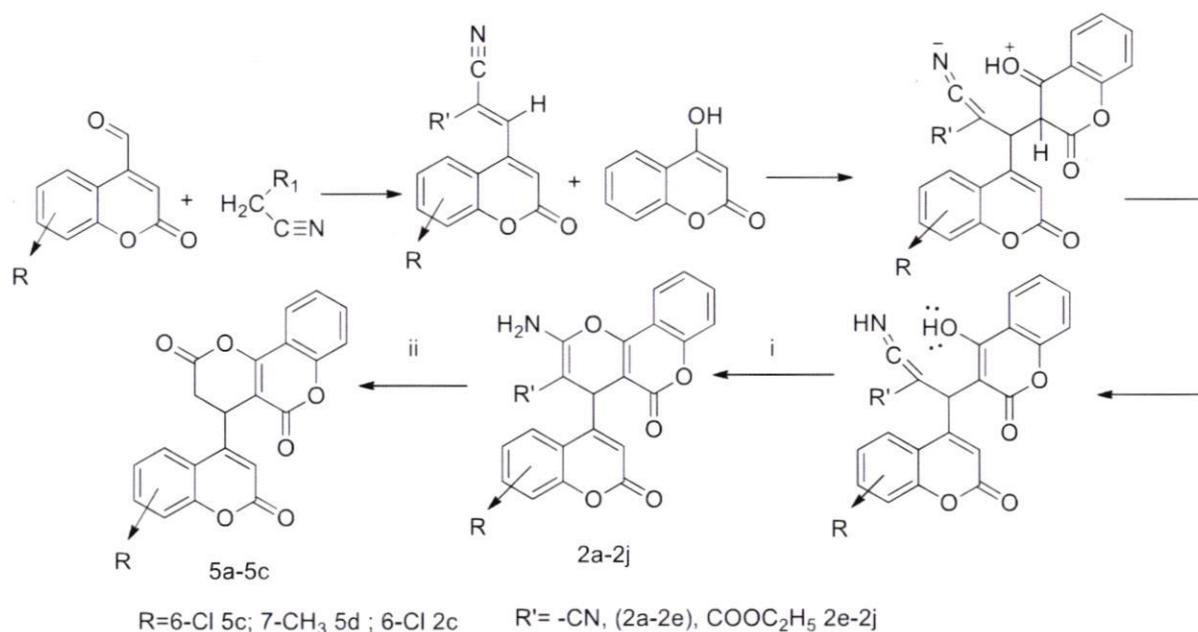


Reagents and conditions: i) Br₂, dry CHCl₃ ii) benzimidazole, CH₃CN, stir, 3h. iii) NH₂OH.HCl, C₂H₅OH, 70-80°C, 3h iv) 4-chlorobenzyl bromide, potassium tert.butoxide, DMSO

Scheme 61 N-Benzimidazolyl oxime of acetyl coumarin derivatives.

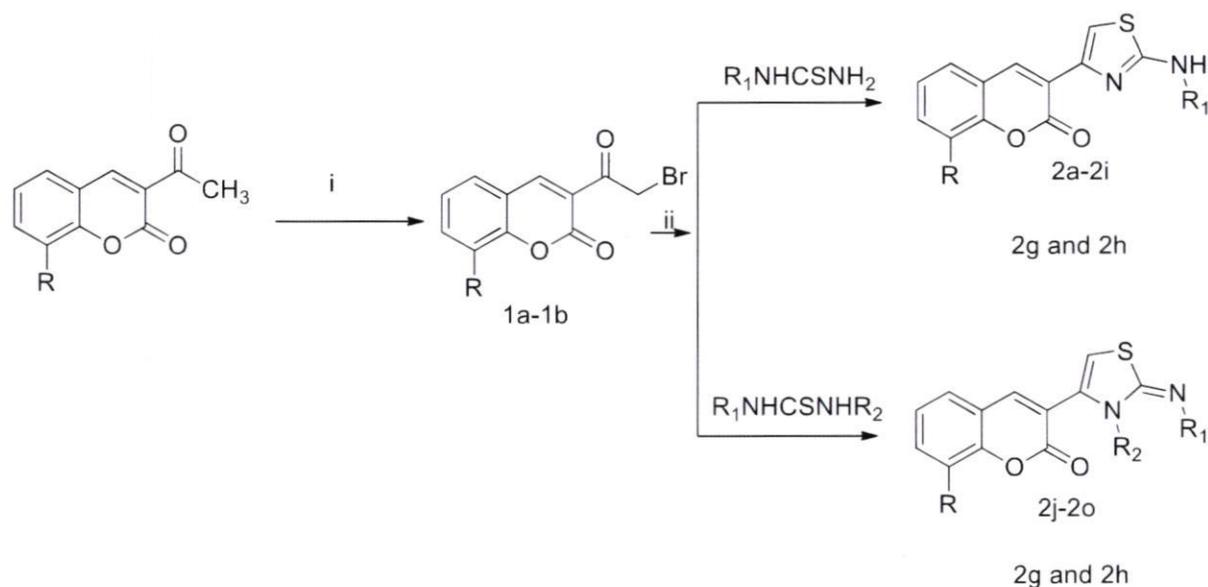
4-methyl-6-(prop-2-ynoxy)-2H-chromen-2-one and substituted alkyl azide in the principle, 'click reaction', and the obtained compound 4-methyl-6-((1-subst. alkyl-1H-1,2,3-triazol-4-yl)methoxy)-2H-chromen-2-one yielded **5a-5 l**. The compound **5c** having n-butyl substituted coumarin linked

with triazolylloxymethyl at C-6 position and isopropyl substituted coumarin linked with triazolylloxymethyl **5j** had been reported as a good *in vitro* antibacterial agent against *E. coli* and *S. aureus* at MIC values 8 and 7 µg/mL, respectively (Kolicchala et al., 2018).



Reagents and conditions: i) Catalyst L proline/ water, 75°C, 2h ii) HCOOH, 8-10h, 130°C

Scheme 62 Coumarin based pyrano[3,2-c] coumarin derivatives.

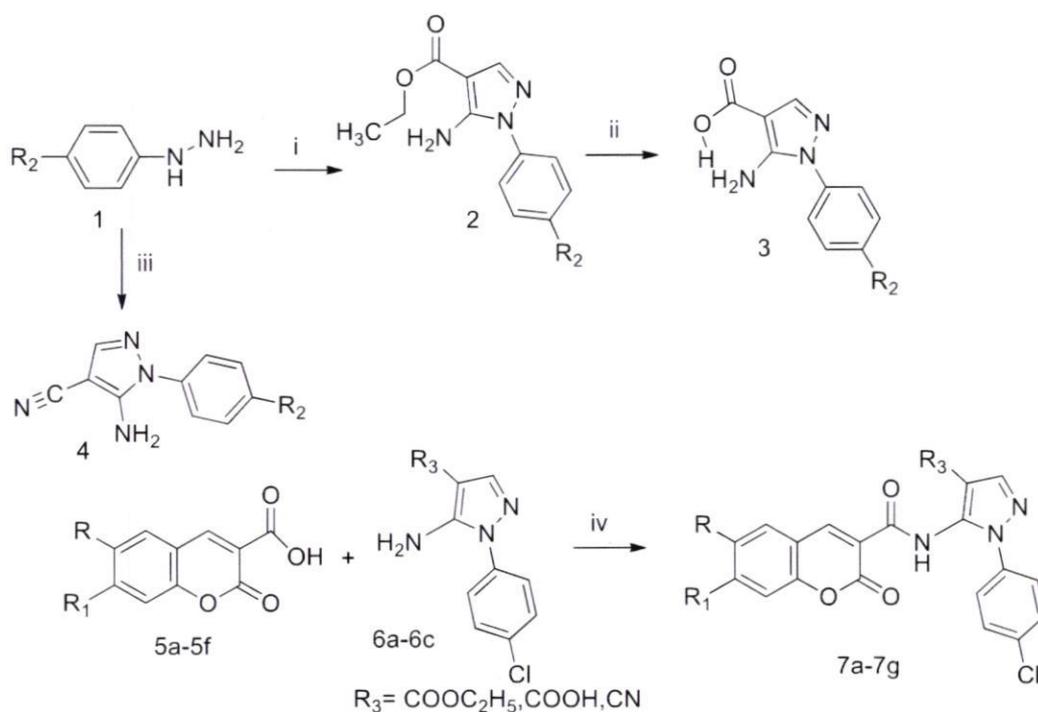


Scheme 63 3-Thiazolylamino coumarin derivatives.

2.36. Synthesis of coumarin bearing triazole derivatives

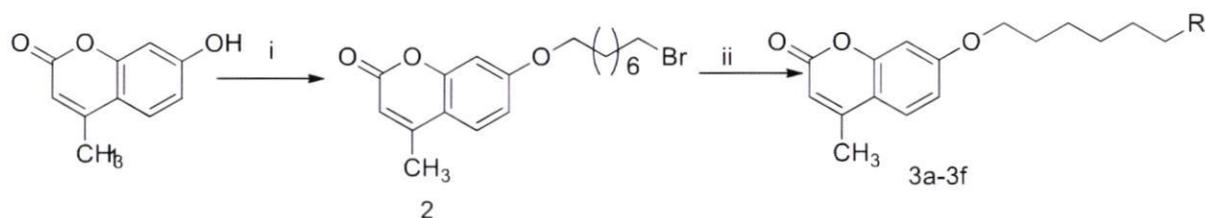
In this scheme, coumarin derivatives having substituted triazole ring attached were designed and synthesized using copper(I) catalysed by 'Huisgen 1,3-dipolar' reaction of terminal alkyne with treatment azide. In the scheme, intermediate 4-azidomethyl coumarin derivatives were liberated as sodium

azide with 4-chloromethyl- 7-hydroxy coumarin and 4-chloromethyl- 7-methyl coumarin, then these were prepared by the reaction of 3-hydroxy phenol and 3-methyl phenol with chloro ethyl acetoacetate under cyclisation in the presence of dehydrating agents concentrated sulfuric acid. Furthermore, these 1,2,3-triazole-coumarin hybrids were obtained by 'click chemistry' of 4-azidomethyl coumarin derivative and



Reagents and conditions: i) Cyano-ethoxy-acrylate, H_2O , NaOH , EtOH , 3h, reflux ii) H_2O , NaOH , EtOH , 3h, reflux iii) 2-(ethoxymethyl)malonylnitrile, H_2O , NaOH , EtOH , 3h, reflux iv) POCl_3 , pyridine

Scheme 64 N-Pyrazolyl coumarin-3-carboxamide derivatives.



$R = \text{imidazol-2-yl- } 3a; 2\text{-methyl imidazol-2-yl- } 3b; 4\text{-methyl imidazol-2-yl- } 3c; \text{benzimidazol-2-yl } 3d; 1,2,4\text{-triazolyl } 3e$

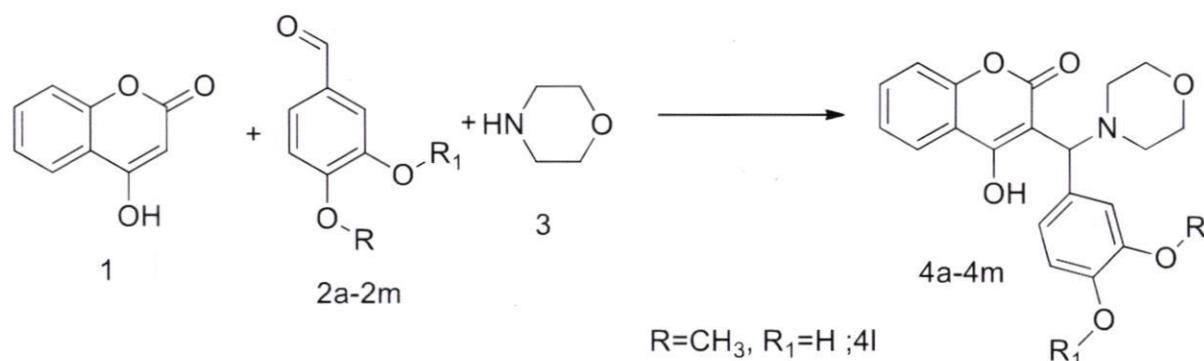
Reagents and conditions: i) 1,6-dibromohexane, K_2CO_3 , dry CH_3COCH_3 , reflux, 20--24h ii) 2-methyl imidazole, K_2CO_3 , CH_3CN , reflux, 20-24h

Scheme 65 1-Alkyloxy imidazolyl coumarin derivatives.

substituted alkynes in the presence of catalyst Cu(I) , which generally was prepared *in situ* by copper sulfate and metallic copper. Moreover, the synthesis of 1,2,3-triazolyl substituted aryl sulfonamide of coumarin azide and the corresponding *N*-propargylated aryl sulfonamides were prepared. The compound 7-hydroxy-4-((4-(4-pentylphenyl)-1*H*-1,2,3-triazol-1-yl)methyl)-2*H*-chromen-2-one **17** had shown as a good antibacterial agent against *E. faecalis* at MIC value 8 $\mu\text{g/mL}$ (Kraljević et al., 2016).

2.37. Synthesis of silver complexes with heterocyclic substituted coumarin derivatives

A series of coumarin silver(I) metal complexes bearing *N*-heterocyclic carbene (NHC) were synthesized by interaction of the corresponding imidazolium or benz[*d*]imidazolium chloride and silver oxide in dichloromethane at room temperature; and the obtained corresponding metal complexes were characterised by elemental analysis, $^{13}\text{C}/^1\text{H}$ NMR and



Scheme 66 Coumarin mannich based derivatives.

mass spectral studies. These silver NHC complexes were prepared in the reactions consisting of three diverse steps, initially with coumarin ligand 4-(chloromethyl)-6,8-dimethyl-2*H*-chromen-2-one **1**, precursor prepared from chloroethyl acetoacetate, which was condensed with 2,4-dimethylphenol in the presence of concentrated sulphuric acid; thereafter, silver NHC complexes were prepared by the treatment of *N*-substituted imidazolium or *N*-substituted benzimidazolium chloride with equivalent silver oxide in DCM after 24 h. Among all complexes, the compound with coumarin bearing *N*-naphyl benzimidazolium in structural frame **5e** was the good antibacterial agent against *S. aureus* and *E. faecalis* at MC value 25 µg/mL, individually in comparison to Ampicillin and Ciprofloxacin (Karataş et al., 2016).

2.38. Synthesis of silver complexes with coumarinyloxyphenoxy acetic acid derivatives

Silver metal complexes bearing coumarin with phenathroline adducts were prepared by the ligand 2-((2-oxo-2*H*-substituted chromen-3-yl)oxy)acetic by condensation of an appropriate substituted 3-hydroxy coumarin with bromoethyl acetate with further hydrolysis of the obtained ester on in acetone: water mixture. The silver(I) complexes were prepared by deprotonation of substituted coumarinyl phenoxy acetic acid ligands using stoichiometrically equal amount of sodium bicarbonate and silver nitrate. Furthermore, silver (I)-phenathroline adduct were obtained by mixing of the silver substituted coumarinyl phenoxy acetate with phenathroline in ethanol. The compound having silver metal complex containing phenathroline had was a good antibacterial agent against enterococci with the MIC₅₀ value 16 µM in compare to Vancomycin as the standard (Mujahid et al., 2016).

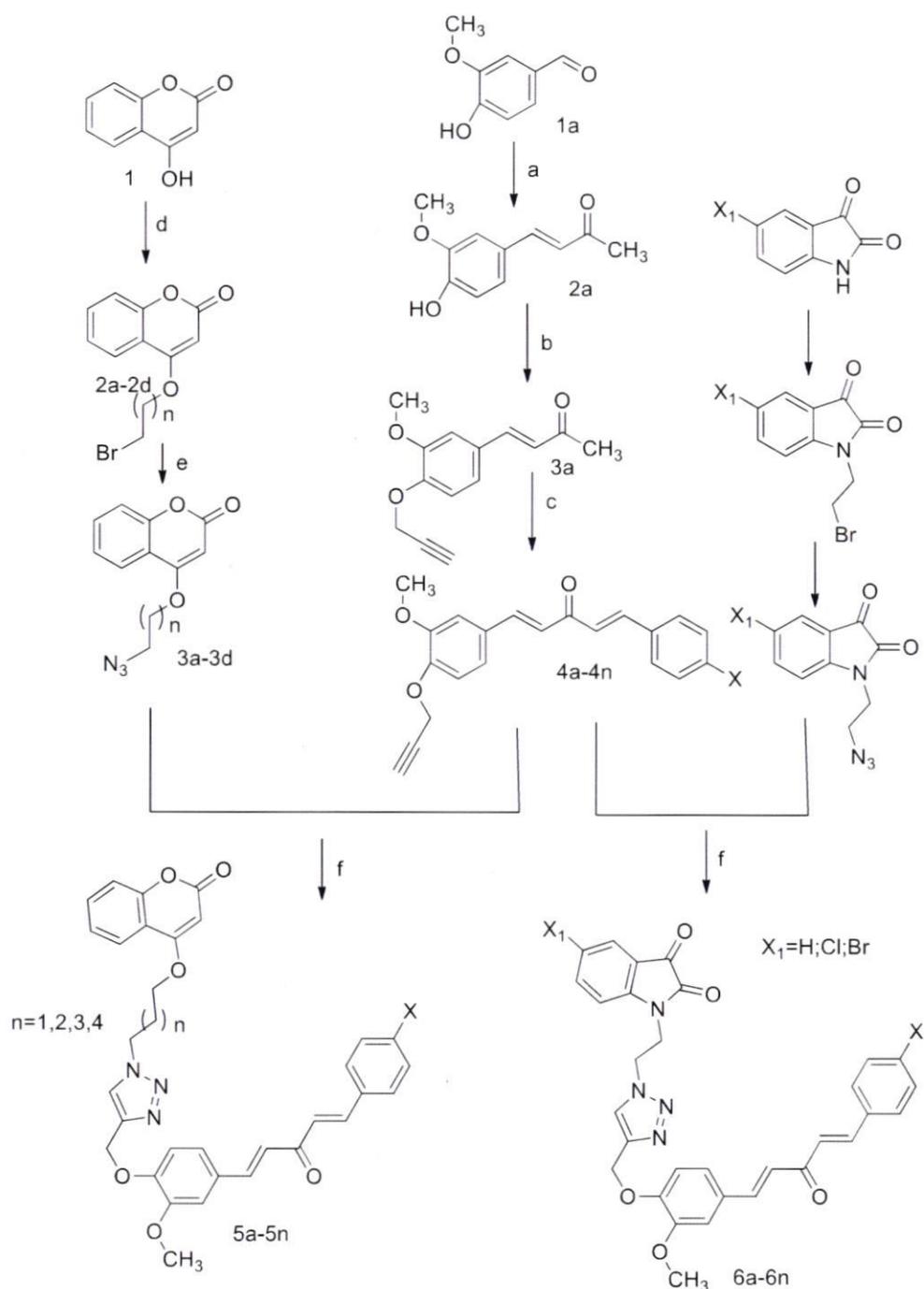
2.39. Synthesis of 4-amino coumarin derivatives

Three series of 4-amino substituted were designed and synthesized from intermediate corresponding chalcones yielded 4-amino coumarin derivatives. Phenyl substituted isoxazole, pyrimidin-2-one moiety were connected coumarin at C-4 position of 4-amino coumarin. These three series compounds were synthesised by cyclisation of coumarin derived chalcones treated with hydroxylamine, thiourea and urea. In this synthesis, the starting material 4-chloro coumarin **2** was prepared by heating of the mixture of 4-

hydroxy coumarin **1** with phosphorous oxytrichloride, which further yielded the corresponding 4-((4-acetylphenyl)amino)-2*H*-chromen-2-one **3** form an intermediate by reacting with 4-amino acetophenone in the presence of sodium carbonate. A series of α,β -unsaturated carbonyl substituted 4-amino coumarin (**4a-4k**) was prepared by mixing of the 4-((4-acetylphenyl)amino)-2*H*-chromen-2-one **2** in DMF solution with an appropriate aryl aldehyde in the presence of potassium hydroxide at room temperature. Thereafter, an individual chalcone derivative in alcoholic solution was mixed with hydroxylamine hydrochloride to produce the corresponding 4-((4-(5-subst.phenylisoxazol-3-yl)phenyl)amino)-2*H*-chromen-2-one **5a-5k** by the cyclisation; whereas, other derivatives such as, 4-((4-(6-(subst.phenyl)-2-hydroxypyrimidin-4-yl)phenyl)amino)-2*H*-chromen-2-one **7a-7k**, 4-((4-(subst.phenyl)-2-mercaptopyrimidin-4-yl)phenyl)amino)-2*H*-chromen-2-one analogues **6a-6k** were prepared by cycliation of chalcones with urea and thiourea, respectively. These derivatives had been monitored for antibacterial activities against eight bacterial strains. The compound 4-((4-(6-(3,4-dimethoxyphenyl)-2-mercaptopyrimidin-4-yl)phenyl)amino)-2*H*-chromen-2-one **7j** was reported as having good antimycobacterial activity at MIC value 25 µg/mL. Furthermore, compounds 4-((4-(6-(3,4-dihydroxyphenyl)-2-mercaptopyrimidin-4-yl)phenyl)amino)-2*H*-chromen-2-one **7i** and 4-((4-(6-(4-methoxyphenyl)-2-mercaptopyrimidin-4-yl)phenyl)amino)-2*H*-chromen-2-one **7k** had been reported as being as good antibacterial agents against *S. aureus*, *B. cereus* and *E. coli* at 28 mm ZOI and with the MIC value 3.12 µg/mL (Patel et al., 2017).

2.40. Synthesis of imidazoliummethyl coumarin silver complexes

A series of coumarin tethered *N*-heterocyclic carbene (NHC) silver (I) complexes (**5-21**) was designed and synthesized. NHC ligands bearing *N*-substituted imidazolium and benzimidazolium had been developed by integument of transition metal complexes and those had been suitable for biological uses for treating several bacterial infections. The coumarin tethered bis imidazolium salt had been developed through two steps alkylation reactions, initially alkylation of either imidazole or benzimidazole to obtain 1-alkyl imidazole or 1-alkyl benzimidazole, respectively, which further the formation respective imidazolium bromide salt had been accomplished



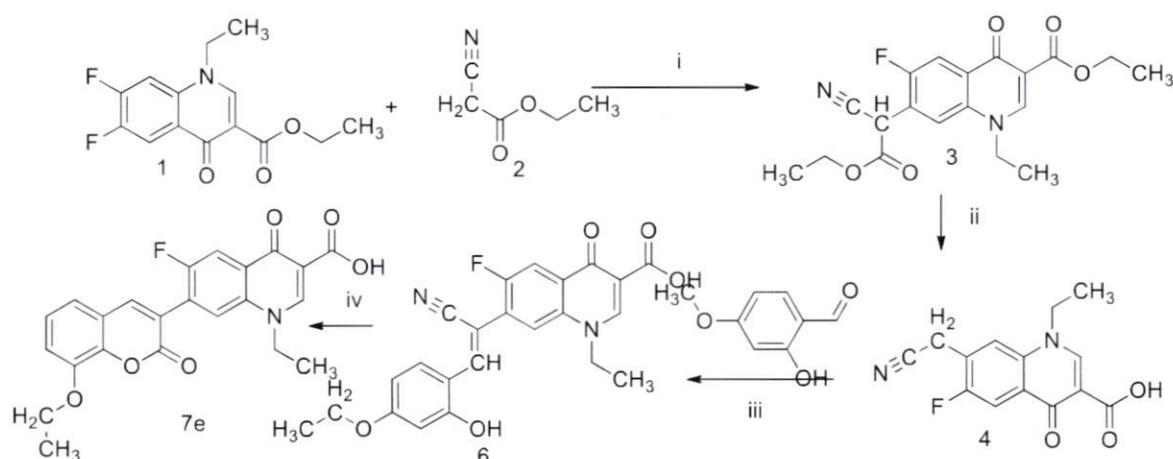
$X = \text{H}; 4\text{-Cl}; 4\text{-Br}; 3\text{-Br}; 4\text{-OCH}_3; 4\text{-NO}_2; 3\text{-NO}_2; 3\text{-OCH}_3; 2\text{-OCH}_3; 3,4\text{-di OCH}_3; 3,4,5\text{-tri OCH}_3; \text{naphthyl-}; \text{furanyl-}; \text{thienyl-}$

Reagents and conditions: a) $\text{CH}_3\text{COCH}_3, 40\% \text{KOH}, 2\text{h}, \text{stir}, \text{rt}$ b) propargyl bromide, $\text{K}_2\text{CO}_3, \text{DMF}, 2\text{h}, \text{stir}, \text{rt}$ c) corresponding aromatic aldehydes, $5\% \text{NaOH}, \text{CH}_3\text{OH}, 2\text{h}, \text{stir}, \text{rt}$ d) dibromoalkane, $\text{K}_2\text{CO}_3, \text{DMF}, 2\text{h}, \text{stir}, \text{rt}$ e) $\text{NaN}_3, \text{DMF}, 1\text{h}, \text{stir}, \text{rt}$ f) Na ascorbate, $\text{CuSO}_4, \text{DMF}, 10\text{-}15\text{m}$

Scheme 67 Curcumin and isatin linked coumarin derivatives.

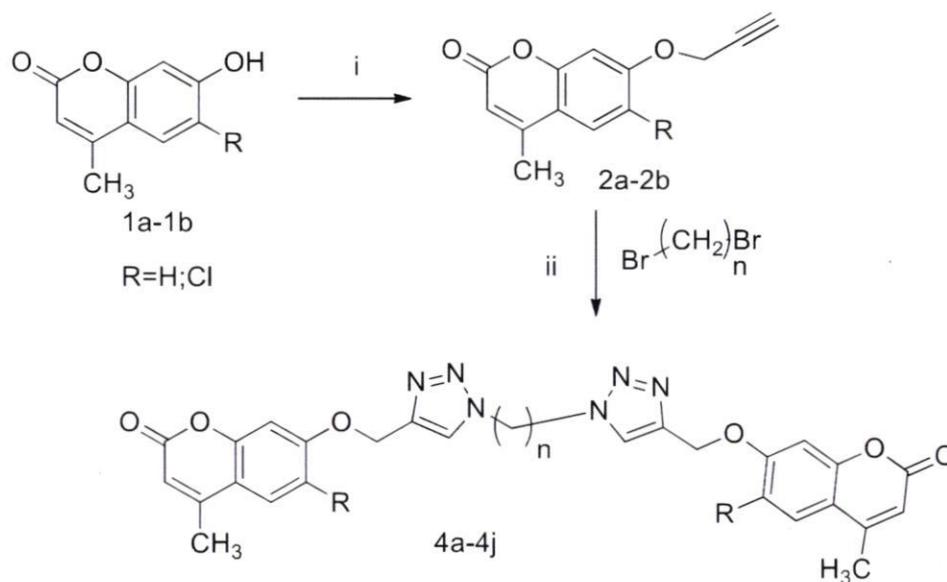
in the successive *N*-alkylation through addition of equimolar concentration of *N*-alkylated with 4-bromomethyl-6-methyl-coumarin in 1,4-dioxane solution at 85°C for 24 h. Further-

more, the imidazolium bromide salts with potassium hexafluorophosphate in methanol and water (9:1) produced the desired sterically tethered coumarin-imidazolium hex-



Reagents and conditions: i)DMF, K_2CO_3 , 70-80°C ii) H_2SO_4 , CH_3COOH , H_2O (1:20:20), 110°C iii)DMF, piperidine iv)3-5% H_2SO_4 , 110°C

Scheme 68 Fluoroquinolone based coumarin derivatives.



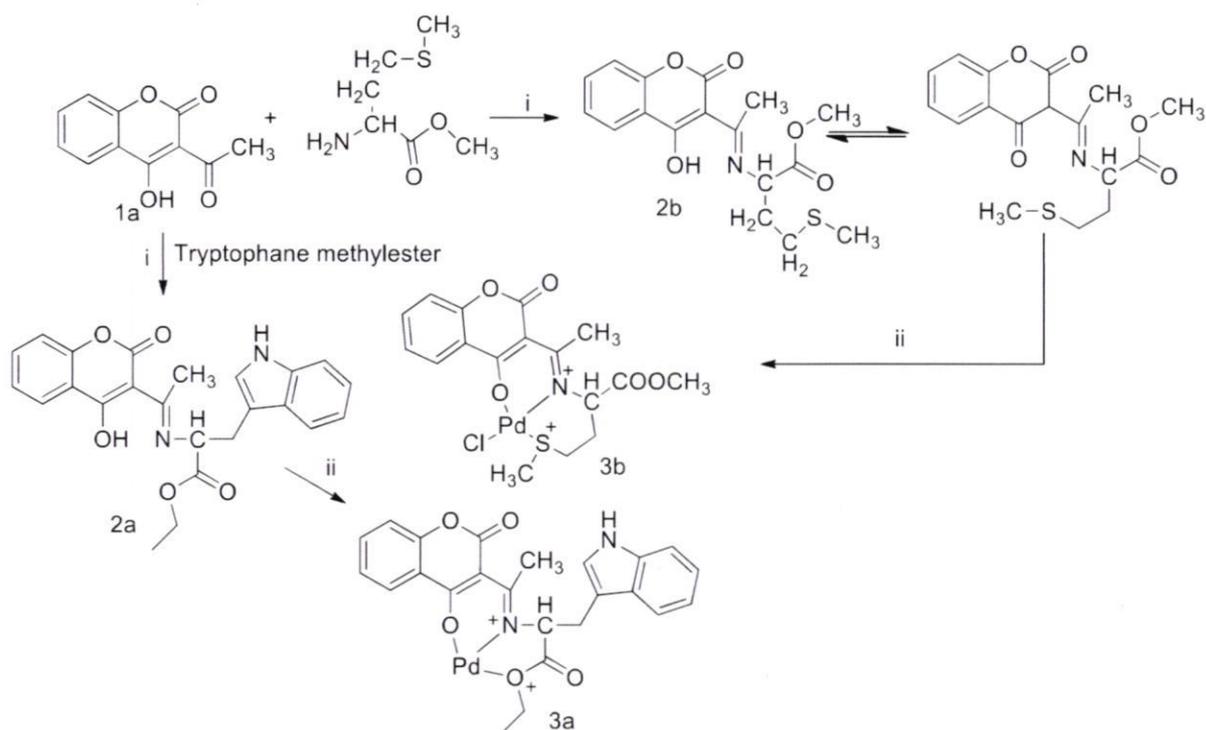
R= Cl 4e; H 4j; n=8 R= Cl; n=6 4i

Reagents and conditions: i)propargyl bromide, anhydrous K_2CO_3 , acetone, reflux, 18h; ii) NaN_3 , CuI , $DMF:H_2O$, 80°C, 24h

Scheme 69 Dimer of Triazole-coumarin hybrids derivatives.

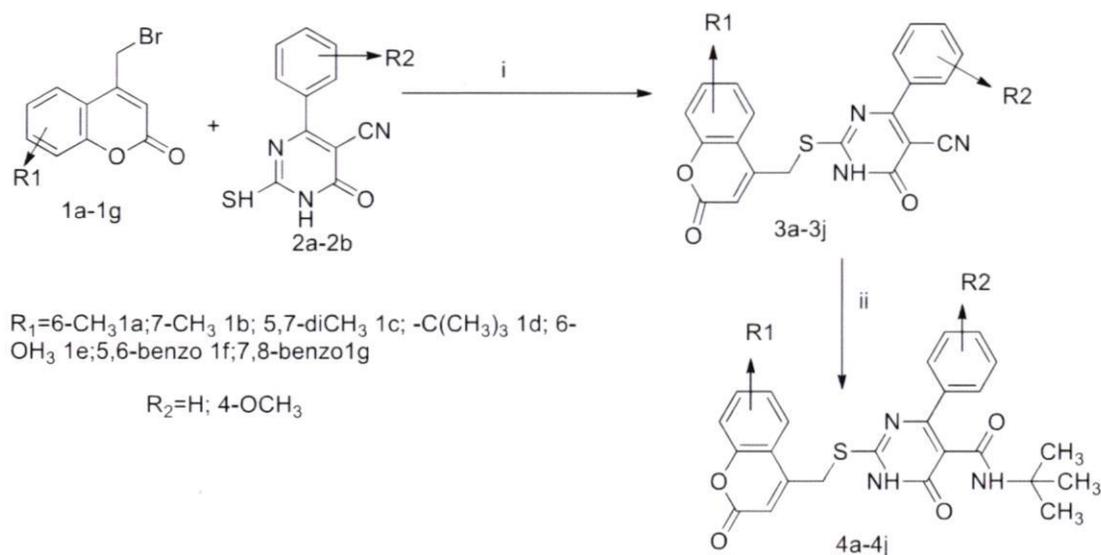
afluropentaphosphate in high yields and these compounds acted as NHC ligands upon *in situ* reaction deprotonation at C-2. Finally, the targeted silver- bis NHC complexes had been synthesized by reactions of the corresponding imidazolium salts with silver oxide (Ag_2O) in acetonitrile at 45 °C under dark condition. Among all the tested compounds, silver salts of bis butyl, pentyl and hexyl substituted imidazolium of 6-

methyl coumarin had shown good antibacterial action against *E. coli* at MIC value 8 $\mu g/mL$. Moreover, the compound **18**, **19** and **20** were reported as having good antibacterial activity against *E. coli* at MIC value 8 $\mu g/mL$. Similarly, compounds **15** and **18** were reported active against *P. aeruginosa* at MIC value 8 $\mu g/mL$ in comparison to Ciprofloxacin as the standard (Achar et al., 2017b).



Reagents and conditions: i) $\text{CH}_3\text{OH}, (\text{C}_2\text{H}_5)_3\text{N}$, ii) $\text{K}_2\text{PdCl}_4, \text{CH}_3\text{OH}$

Scheme 70 Palladium complexes coumarin derivatives.



$\text{R}_1 = 6\text{-CH}_3$ 1a; 7-CH_3 1b; $5,7\text{-diCH}_3$ 1c; $-\text{C}(\text{CH}_3)_3$ 1d; 6-OH_3 1e; $5,6\text{-benzo}$ 1f; $7,8\text{-benzo}$ 1g

$\text{R}_2 = \text{H}; 4\text{-OCH}_3$

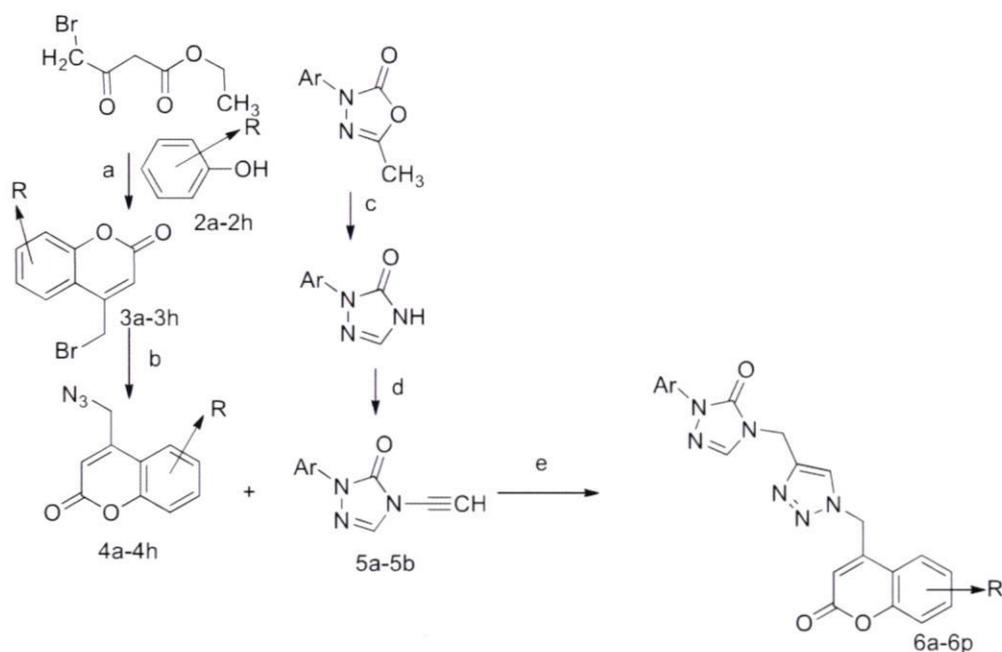
Reagents and conditions: i) Acetone K_2CO_3 , rt ii) MW, 200W, 6-10min, 75°C OR $(\text{CH}_3)_3\text{COOH}$, acetic acid, 2-3h

Scheme 71 Coumarinyl pyrimidinone derivatives.

2.41. Synthesis of 7-coumarinyloxy amino propanol derivatives

In the scheme, 7/4-coumarinyloxy propanol substituted amine were synthesised from starting material either 7-hydroxy 4-

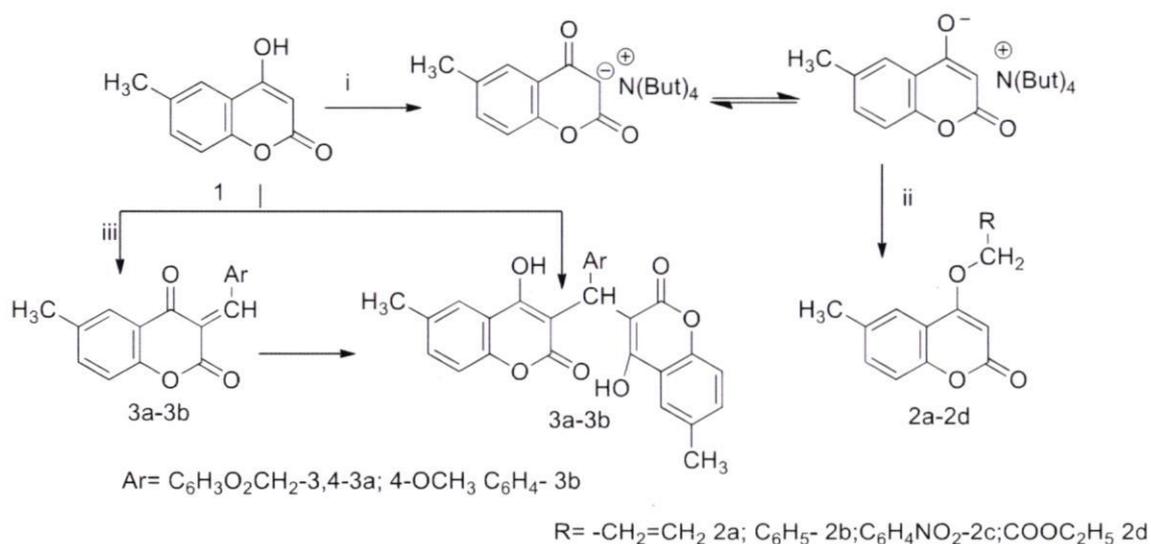
methyl coumarin or 4-hydroxycoumarin. The synthesis of the desired compounds comprised two step-reactions among which, 7-hydroxy-4-methyl coumarin/4-hydroxy coumarin was treated with epichlorohydrin in the presence of anhydrous



6e (Ar= phenyl R= 7-CH₃ ; 6f Ar= phenyl , R=5,6-benzo; 6i Ar= 4-Anisyl ,R= 6-CH₃ ;6j (Ar=4-Anisyl R= 6-Cl; ,6k(Ar=4-Anisyl R= 6-'t'-butyl; ,6l (Ar=4-Anisyl R= 7-CH₃; ,6n(Ar=4-Anisyl R= 5,6-benzo);

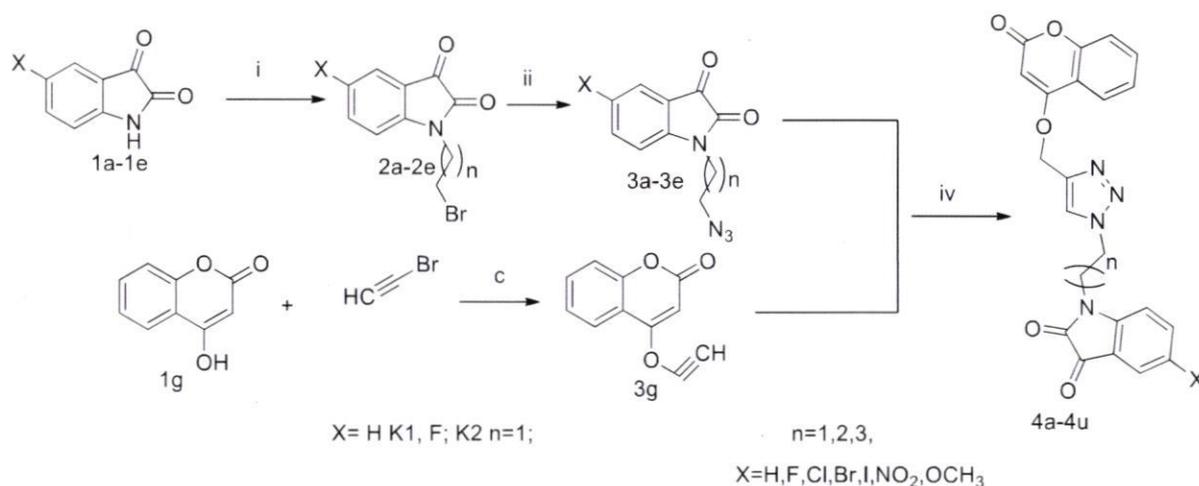
Reagents and conditions: a)Conc.H₂SO₄,0-5°C b)NaN₃,acetone,water,rt c)NH₂CHO,180°C d)propargyl bromide, anhydrous K₂CO₃, acetone,rt

Scheme 72 1,2,3-Triazole substituted coumarin derivatives.



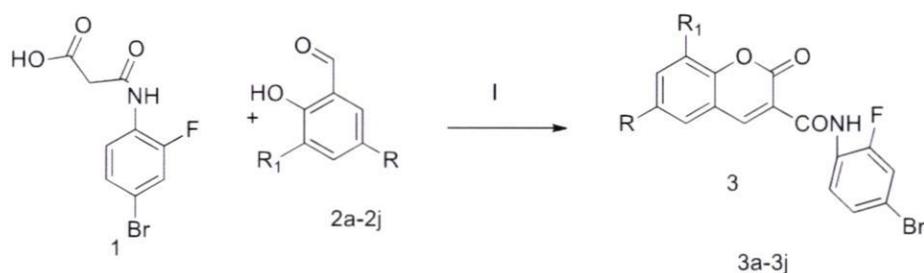
Reagents and conditions: i) K₂CO₃/acetone, TBAC ii)ClCH₂COOC₂H₅,acetone rt, 3h for 2d iii).Piperidine C₂H₅OH

Scheme 73 Coumarinyloxy derivatives.



Reagents and conditions: i) K_2CO_3 , DMF, 2h, stir, rt ii) NaN_3 , DMF, 1h, stir, rt; iii) I_2CO_3 , DMF, 2h, stir, rt iv) sodium ascorbate, CuSO_4 , DMF, 15min, rt

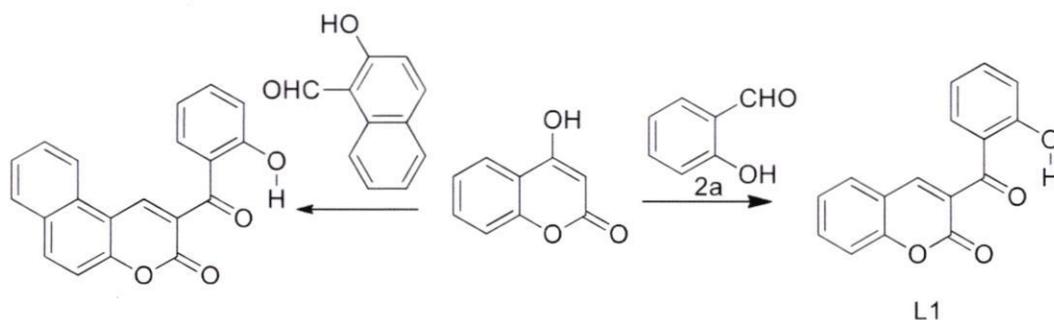
Scheme 74 Isatin-triazolyl coumarin derivatives.



$R = \text{Br}, R_1 = \text{H}$ 3a, $R = \text{Cl}, R_1 = \text{H}$ 3b, $R = \text{NO}_2, R_1 = \text{H}$ 3c; $R = R_1 = \text{Br}$ 3d; $R = R_1 = \text{Cl}$ 3e; $R = \text{Cl}, R_1 = \text{NO}_2$ 3f; $R_1 = \text{H}, R_2 = \text{NO}_2$

Reagents and conditions: i) Piperidine, acetone, 3h, reflux

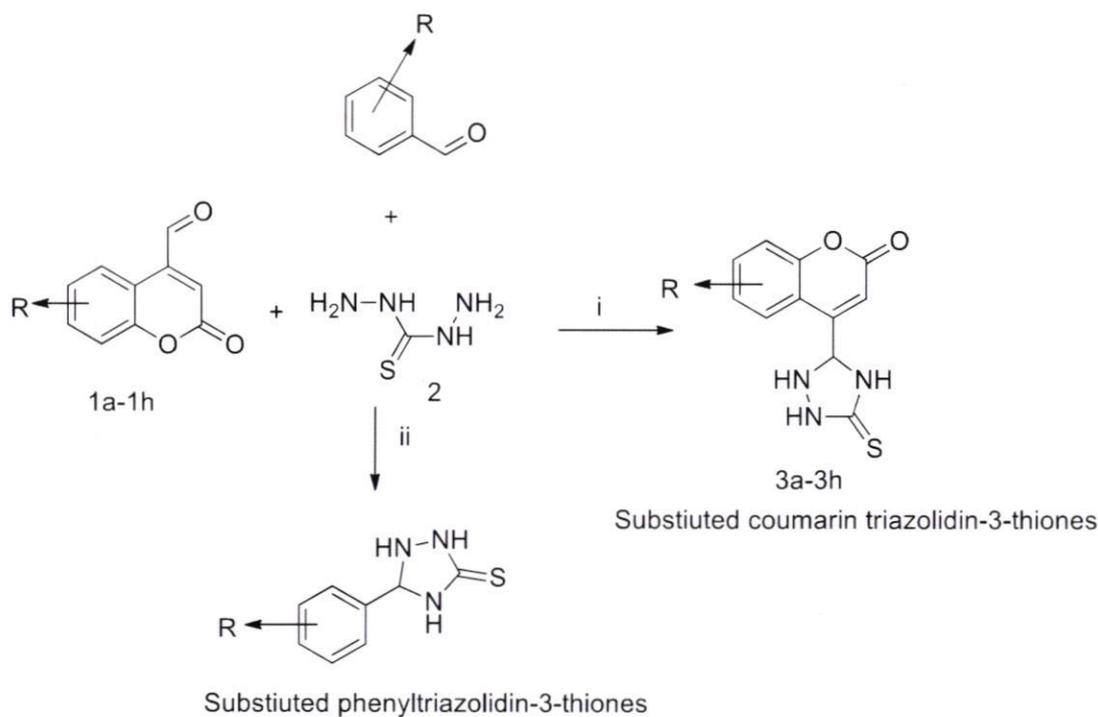
Scheme 75 Coumarin 3-carboxamide derivatives.



Scheme 76 3-Aroyl substituted coumarin derivatives.

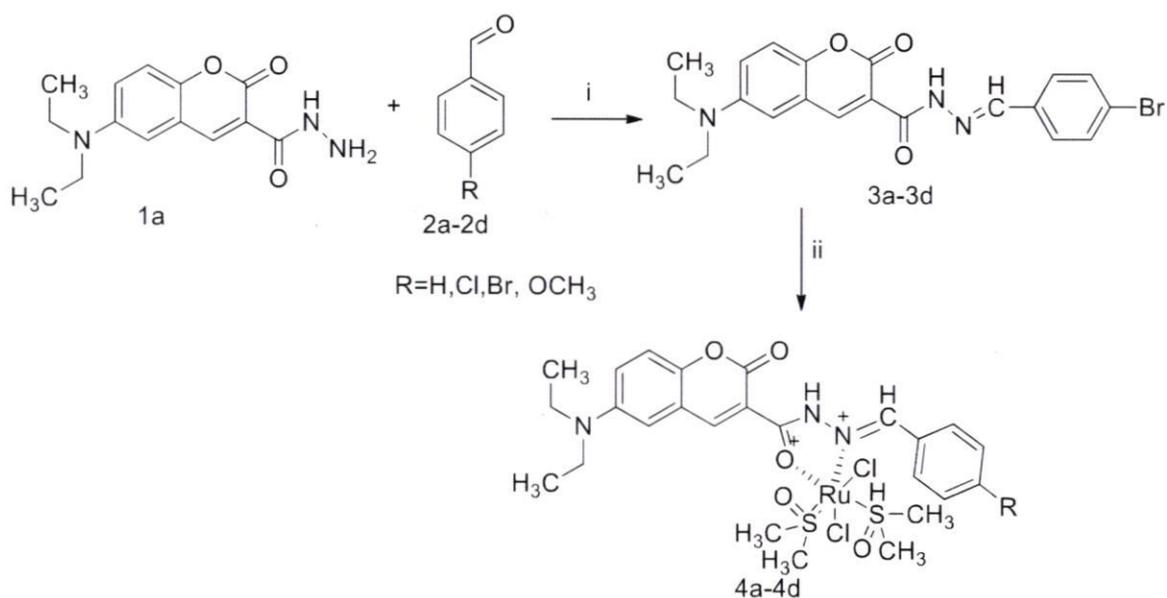
potassium carbonate to produce the corresponding oxirane derivative with a high yield; thereafter, those intermediates reacted with various primary or secondary cyclic amines in

ethanol at room temperature yielded coumarinyl oxy propanol amines analogues. During the reaction, nucleophilic oxirane ring had opened by the treatment with amines to produce



Reagents and conditions: i) Different solvents like water or PEG, ethanol so on

Scheme 77 4-Triazolidin-thione coumarin derivatives.

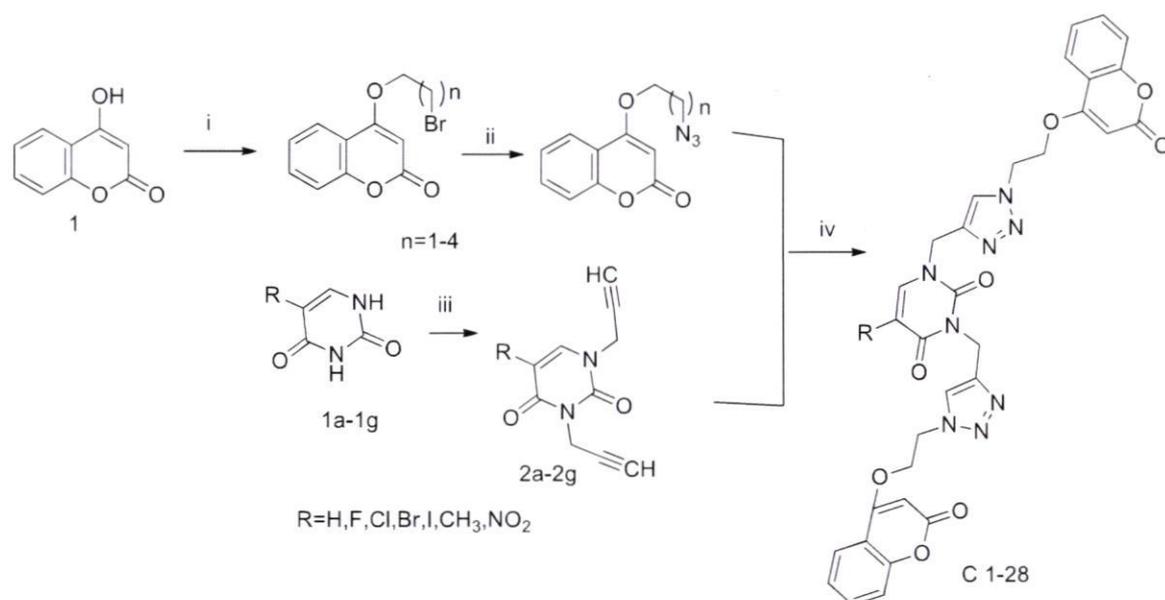


Reagents and conditions: i) HCl, C₂H₅OH, rt, 1h ii) cis[RuCl₂(DMSO)₄], C₂H₅OH, reflux, 4h

Scheme 78 Ruthenium complexes of 3-acetohydrazone coumarin.

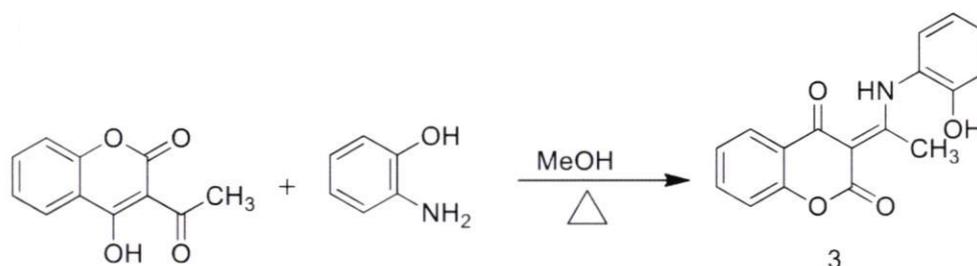
coumarinyl amino alcohol derivatives. SAR analyses of these compounds revealed that compounds with 7-substituted amino propanol of 4-methyl coumarin derivatives had greater (stronger)

antibacterial action compared to 4-substituted amino propanol coumarin derivatives, due the methyl group which might have induced the hydrophobic interaction leading to a better



Reagents and conditions: i) Dibromoalkane, K_2CO_3 , DMF, 2h, stir, rt ii) NaN_3 , DMF, 1h, stir, rt, iii) propargyl bromide, K_2CO_3 , DMF, 2h, stir, rt iv) sodium ascorbate, $CuSO_4$, DMF, 15min, stir, rt

Scheme 79 Bis-Triazole uracil based coumarin derivatives.



Scheme 80 4-Hydroxy- 3-acetyl coumarin derivatives.

penetration of the compounds through bacterial cell membranes. Furthermore, the compounds with long alkyl chain substituent on side of amino group had been shown active than short alkyl chain amine substituent compounds. Among all the tested compounds, long alkyl chain contains twelve carbons had promising bioactive compound due to high lipophilicity effects. The antibacterial action results indicated that the compound 7-(3-(di(piperidin-1-yl)amino)-2-hydroxypropoxy)-4-methyl-2H-chromen-2-one **17** and compound bearing dodecyl amine substituent at C-7 of 4-methyl coumarin **18** was reported as good antibacterial agent against *K. pneumoniae* at MIC value 6.25 $\mu\text{g/mL}$, in comparison to Novobiocin and Vancomycin. The compound with oleyl group substituent at amine of 4-methyl coumarin **16** had shown a good antibacterial activity against four bacterial strains namely, *S. aureus*, *E. coli*, *K. pneumoniae* and *P. aeruginosa* with MIC value ranging 6.25–25 $\mu\text{g/mL}$ (Priyanka et al., 2017).

2.42. Synthesis of 3-heteroaryl substituted coumarin derivatives

In this scheme, a series of 2-((furan-2-ylmethylene)amino)-6-(2-oxo-2H-chromen-3-yl)-4-substituted phenylnicotinonitrile derivatives **5a-5m** were designed and synthesized through the microwave irradiation of the reaction mixture of 2-amino-6-(2-oxo-2H-chromen-3-yl)-4-substituted phenylnicotinonitrile **3a-3m** and furan 2-carboxaldehyde **4** in glacial acetic acid to which with zinc chloride was added at 300 W. Initially, 3-acetyl coumarin **1** was reacted with different aryl aldehyde **2** malononitrile and ammonium acetate in microwave irradiation at 130 $^{\circ}\text{C}$ for 6–10 min, and the liberated 2-amino-6-(2-oxo-2H-chromen-3-yl)-4-substituted phenylnicotinonitrile. Then the compounds **5d**, **5f**, **5j**, **5k** and **5l** with 4-methoxy, 4-methyl, 4-hydroxy, 3-hydroxy and 3-methoxy respectively; those were attached to phenyl substituent of nicotinyl nitrile

Table 1 Isolation of coumarin from several sources.

Sl. No.	Plant name	General name	Family	Reference
1.	<i>Agathosma betulina</i>	Boegoe	Rutaceae	(Moolla and Viljoen, 2008)
2.	<i>Ammi majus</i>	Bishop's weed	Apiaceae	(Bartnik and Mazurek, 2016)
3.	<i>Amyris elemifera</i>	Torchwood	Rutaceae	(Burke and Parkins, 1979)
4.	<i>Angelica archangelica</i>	Garden angelica	Apiaceae	(Kumar et al., 2013)
5.	<i>Apium graveolens</i>	Leaf celery	Apiaceae	(Ramezani et al., 2009)
6.	<i>Artemisia tridentata</i>	Sagebrush	Asteraceae	(Barua et al., 1980)
7.	<i>Asclepiascurassavica</i>	Scarlet milkweed	Apocynaceae	(Reddy, 2012)
8.	<i>Calophyllum brasiliense</i>	Árbol de Santa Maria	Calophyllaceae	(Ruiz-Marcial et al., 2007)
9.	<i>Calophyllum dispar</i>	Laurelwood	Calophyllaceae	(Guilet et al., 2001)
10.	<i>Cinnamomum cassia</i>	Chinese cassia	Lauraceae	(Woehrlin et al., 2010)
11.	<i>Cinnamomum verum</i>	Ceylon cinnamon	Lauraceae	(Ballin and Sørensen, 2014)
12.	<i>Citrus reticulata</i>	Citrus plant	Rutaceae	(Takemura et al., 1993)
13.	<i>Citrus hassaku</i>	Jagada	Rutaceae	(Nakatani et al., 1987)
14.	<i>Cleome hasslerian</i>	Spider flower	Capparaceae	(Kumar et al., 1988)
15.	<i>Coleonema album</i>	Aasbossie	Rutaceae	(Gray, 1981)
16.	<i>Daucus carota</i>	Carrot	Apiaceae	(Gilani et al., 2000)
17.	<i>Daucus carota</i>	Wild carrot	Apiaceae	(Abenavoli et al., 2003)
18.	<i>Dipteryx odorata</i>	Tonka bean	Fabaceae	(Gleye et al., 2003)
19.	<i>Eriostemon brucei</i>	Philotheca	Rutaceae	(Jefferies and Worth, 1973)
20.	<i>Eryngium campestre</i>	Field eryngo	Apiaceae	(Erdelmeier and Sticher, 1985)
21.	<i>Ferula asafetida</i>	Asafetida	Apiaceae	(Iranshahy and Iranshahi, 2011)
22.	<i>Ferula communis</i>	Gaint fennel	Apiaceae	(Appendino et al., 1988)
23.	<i>Galium odoratum</i>	Sweet woodruff	Rubiaceae	(Martin and Bodson, 2010)
24.	<i>Gerbera jamesoni</i>	Transvaal daisy	Asteraceae	(Inoue et al., 1989)
25.	<i>Helianthus annuus</i>	Sunflower	Asteraceae	(Serghini et al., 2001)
26.	<i>Heracelum thomsoni</i>	Cowparsnip	Apiaceae	(Patnaik et al., 1987)
27.	<i>Hierochloe odorata</i>	Vanilla grass	Poaceae	(Sinha et al., 2008)
28.	<i>Hierochloe odorata</i>	Sweet grass	Poaceae	(Brown et al., 1960)
29.	<i>Kielmeyera elata</i>	White gul mohur	Calophyllaceae	(Gramacho et al., 1999)
30.	<i>Melilotus officinalis</i>	Sweet clover	Fabaceae	(Brown et al., 1960)
31.	<i>Micromelum minutum</i>	Lime berry	Rutaceae	(Rahmani et al., 1994)
32.	<i>Murraya exotica</i>	Orange jessamine	Rutaceae	(Negi et al., 2005)
33.	<i>Notopterygium forbesii</i>	Notopterygium Root	Apiaceae	(Ma et al., 2008)
34.	<i>Paramignya trimera</i>	Xao tam phan	Rutaceae	(Dang et al., 2017)
35.	<i>Pelargonium sidoides</i>	South African geranium	Geraniaceae	(Krone et al., 2001)
36.	<i>Pelea barbigera</i>		Rutaceae	(Higa and Scheuer, 1974)
37.	<i>Peucedanum formosanum</i>	Milk parsley	Apiaceae	(Chen et al., 2008)
38.	<i>Peucedanum mogoltavicum</i>	Milk parsley	Apiaceae	(Nikonov, 1972)
39.	<i>Peucedanum ostruthium</i>	Milk parsley	Apiaceae	(Urbain et al., 2005)
40.	<i>Phebalium stenophyllum</i>	Narroe leafed phebalium	Rutaceae	(Bevalot et al., 1988)
41.	<i>Polygala paniculata</i>	Milkwort	Polygalaceae	(Hamburger et al., 1985)
42.	<i>Prangos tschimganica</i>	Prangos	Apiaceae	(Shikishima et al., 2001)
43.	<i>Prunus armeniaca</i>	Siberian apricots	Rosaceae	(Kayano et al., 2004)
44.	<i>Prunus avium</i>	Cherries	Rosaceae	(Santamour and Riedel, 1994)
45.	<i>Ribes nigrum</i>	Black currants	Grossulariaceae	(Knox et al., 2003)
46.	<i>Setaria italicia</i>	Foxtail	Poaceae	(Jain et al., 1991)
47.	<i>Toddalia asiatica</i>	Orange climber	Rutaceae	(Oketch-Rabah et al., 2000)
48.	<i>Verbascum thapsus</i>	Mullein	Scrophulariaceae	(Pardo et al., 1998)
49.	<i>Zanthoxylum dipetalum</i>	Kawa'u,Hea'e	Rutaceae	(Fish et al., 1976)
50.	<i>Zanthoxylum suberosum</i>	Fagara	Rutaceae	(Krajniak et al., 1973)

of coumarin at C-3 position; it had shown as good antibacterial activities against *E. coli* and *P. aeruginosa* in comparison to Ampicillin. SAR analyses of these nicotinyl nitrile derivatives indicated that the compound having electron donating capability such as, methoxy, hydroxy and methyl substituent in structure might have contributed to the leading antimicrobial actions (Desai et al., 2017).

2.43. Synthesis of 3-chromenyl carboxamide of coumarin derivatives

A series compounds of chromenyl carboxamide **4a-4l** was synthesized from multi reactant reaction between substituted salicylaldehyde **1a-1c** and derivatives acetoacetanilides **2a-2d** with

4-hydroxy coumarin **1** in the presence of catalyst ceric ammonium nitrate (CAN) under solvent-free condition by the principle 'Knoevenagel-michael' reaction. The reactant acetoacetanilide with both electron donating and electron withdrawing substituents attached to phenyl terminal of the structure reacted with salicylaldehyde and 4-hydroxycoumarins in CAN to yield the respective chromenyl carboxamide derivatives. In this synthesis, a lot of catalysts was used in during reactions, but among all CAN had efficient for cyclisation that afforded good yield under solvent free condition. The solvatochromic properties of the compounds were analysed by using solvents of the increasing order of polarity. The compound 6'-bromo-4-hydroxy-2'-methyl-2-oxo-*N*-(*p*-tolyl)-2*H*,4'*H*-[3,4'-bichromene]-3'-carboxamide **4k** and 6'-bromo-4-hydroxy-2'-methyl-2-oxo-*N*-(4-chlorophenyl)-2*H*,4'*H*-[3,4'-bichromene]-3'-carboxamide **4l** had shown good antibacterial actions against *S. aureus* and *B. subtilis* at MIC value 9.3 µg/mL in comparison to Ampicillin (Chitreddy and Shanmugam, 2017).

2.44. Synthesis of pyrazole bearing coumarin derivatives

A series of coumarin based pyrano[2,3-*c*]pyrazole derivatives **3a-3e** had been synthesized by multiple component reaction steps. In the synthesis, initially the mixture of hydrate hydrazine and alcoholic solution of ethyl acetoacetate was stirred for few minutes then added with substituted coumarin 4-carboxaldehyde, malononitrile or ethyl cyanoacetate with base and stirred for 2 h to produce an intermediate ethyl 2-cyano-3-(2-oxo-2*H*-chromen-4-yl)acrylate **1**, which further reacted with pyrazolin-5-one **2** by 'Michael addition and intramolecular cyclisation' to produce coumarin bearing pyrano[2,3-*c*]pyrazole derivatives **3a-3e**. Moreover, these derivatives were the substituted coumarin based pyrazolyl propanoic acid **4a-4e** in the presence of various acidifying agents for yielding, whereas the mechanism of organic reaction involves open pyran ring system under an acidic condition as nitrile hydrolysis and decarboxylation one of the carboxyl groups. The obtained coumarin bearing pyrano[2,3-*c*]pyrazole derivatives and the corresponding coumarin based pyrazolyl propanoic acid were monitored for *in vitro* antibacterial activity against some bacterial strains. Concomitantly, molecular docking studies had been performed with bacterial dihydropteroate synthetase (DHPS). These tested compounds were subjected to *in vitro* antibacterial activities, which revealed that the compounds **4b** and **4c** had good antibacterial activities against *E. coli* at 0.78, 1.56 µg/mL, respectively. Similarly, the compound **4b** was had *in vitro* control against *E. faecalis* at MIC value 1.56 µg/mL in comparison to Ciprofloxacin (Chougala et al., 2017).

2.45. Synthesis of silver complexes of bis benzimidazolium methyl with coumarin

In this scheme, a series of silver complexes bearing *N*-heterocyclic carbenes (NHC) **5a-5e** had been synthesized and were explored for biological actions. In the synthesis of complexes, initially benzimidazole underwent alkylated with different bromides such as, allyl, isopropyl, benzyl, isopentyl and 2-methylpropyl, to produce *N*-alkyl/allyl benzimidazole **1a-1e**, which products were reacted with 4-bromomethyl-6-chloro coumarin **2** in 1,4-dioxane at room temperature for 10 min

and those were the refluxing mixture for 24 h to afford off-white solid product of the corresponding 1,3-disubstituted benzimidazolium bromide salts **3a-3e**; and further those were recrystallized from acetonitrile and were made to react with hexafluorophosphate form as respective bromides salts into their 1,3-disubst. benzimidazolium hexafluorophosphate **4a-4e** products through metathesis reaction and the obtained salts were recrystallised from acetonitrile:diethyl ether mixtures. In the coumarin tethered silver salts bis-NHC **5a-5e** were prepared by the corresponding hexafluorophosphate that was stirred with silver oxide (Ag₂O) in acetonitrile at 45 °C for 24 h. The antibacterial activity of these compounds was tested, which indicated the compounds **5a**, **5b**, **5c**, **5d** and **5e** had been good antibacterial agents against *P. aeruginosa* at MIC value 8 µg/ mL in comparison to Ampicillin (Achar et al., 2017c).

2.46. Synthesis of indolidenyl of coumarin derivatives

In the scheme, an intermediate 2-(1-(6-(diethylamino)-2-oxo-2*H*-chromen-3-yl)ethylidene)malononitrile **4** was prepared by the condensation of malonyl dinitrile and 3-acetyl-7-(diethylamino)-2*H*-chromen-2-one **3** in the presence of ammonium acetate and acetic acid using benzene as the solvent. An intermediate 3-acetyl-7-(diethylamino)-2*H*-chromen-2-one **3** was prepared by the reaction of diethylamino salicylaldehyde and ethyl acetoacetate in a few drops of piperidine. Furthermore, the reaction of 2-(1-(6-(diethylamino)-2-oxo-2*H*-chromen-3-yl)ethylidene)malononitrile with *N*-subst. indole-3-carboxaldehyde in piperidine solution afforded the title target compound 2-(1-(6-(diethylamino)-2-oxo-2*H*-chromen-3-yl)-3-(subst.1*H*-indol-3-yl)allylidene) malononitrile **5**. The compound **5a** had been reported as an agent against *E. coli* at 10 mm ZOI in agar well diffusion plates (Aksungur et al., 2017).

2.47. Synthesis of pyrimidinone bearing coumarin derivatives as good antibacterial

A series of compounds, dihydro pyrimidin 2-one/thione containing coumarin derivatives **3a-3e/4a-4e** had been synthesized from 4-formyl substituted coumarin and ethyl acetoacetate using thiourea and urea in the presence of an effective catalyst CAN. Those condensations had involved by the principle of 'Biginelli' reactions. During the synthesis, the mechanism with CAN was catalysed in the Biginelli reaction indicated that the reaction would proceed through the formation of acylamino in intermediate ion, which is formed *in situ* by the reaction of formyl coumarin with urea/thiourea. All these synthesized were evaluated for the inherent antibacterial action against *S. aureus*. Among all the tested compounds, the compound ethyl 6-methyl-4-(7-methyl-2-oxo-2*H*-chromen-4-yl)-2-oxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate **3d** had been reported having good antibacterial activity against *S. aureus* at the MIC value, 0.2 µg/mL (Sunagar et al., 2017).

2.48. Synthesis of indole carbaxahydrazide Schiff base coumarin derivatives

In the scheme, a series of transitional metal ions Cu(II), Co(II), Ni(II) and Zn(II) complexes of the Schiff base 5-substituted *N'*-(7-hydroxy-4-methyl-2-oxo-2*H*-chromen-8-yl)methylene-

3-phenyl-1*H*-indole-2-carbohydrazide ligand **3a-3b** were synthesized. An intermediate ligand was prepared by the reaction of methanolic solution 5-substituted-3-phenyl-1*H*-indole-2-carbohydrazide **1a-1b** with 4-methyl-7-chloro 8-formyl coumarin **2** in presence glacial acetic acid then Schiff base ligand reacted with respective metal chlorides in methanolic solution to obtain the corresponding metal complexes of indole carmax hydrazide coumarin derivatives. Metal ions had been coordinated with oxygen atom of carbonyl, nitrogen of azomethine and functional group of phenolic hydroxyl through deprotonation and possess octahedral geometries. The copper complex of ligand **3a** had been reported as a good antibacterial agent against *B. subtilis*, *E. coli* and *S. aureus* at the MIC value 12.50 µg/mL individually (Mahendra raj and Mruthyunjayaswamy, 2017).

2.49. Synthesis of metal complexes 3-aryloxy 4-hydroxy coumarin

In the scheme, a series metal ion complexes of 3-aryloxy-4-hydroxy coumarin **4a-4h** were synthesized. The metal complexes bearing ligands 3-aryloxy 4-hydroxy coumarin **3a-3b** were prepared by the coupling of reaction of respective aryl diazonium salt with 4-hydroxy coumarin in sodium hydroxide at mild condition which further reacted with various metal chloride salt in hot methanol. Furthermore, two moles of ligands were joined together with an individual metal ions like Cu(II), Co(II), Ni(II) and Zn(II) through deprotonation of enolic hydroxyl group of coumarin. The antibacterial activity of ligands and the individual complexes were monitored for *in vitro* antibacterial activities by agar well diffusion method. Among all the tested compounds, the compound with cobalt complexes of (*E*)-3-((4-chlorophenyl)diazonyl)-4-hydroxy-2*H*-chromen-2-one and (*E*)-3-((4-methoxyphenyl)diazonyl)-4-hydroxy-2*H*-chromen-2-one had excellent antibacterial activities against *E. coli* (Sahoo and Paidasetty, 2017).

2.50. Synthesis of coumarin 3-semi thiocarbazon

In this scheme, (*E*)-2-(1-(2-oxo-2*H*-chromen-3-yl)ethylidene)hydrazine carbothioamide derivatives had been synthesised by the reaction of 3-acetyl coumarin, hydrated hydrazine and substituted phenyl isothiocyanate in presence of a small amount catalyst glacial acetic acid in refluxing ethanol and a good yield was liberated. An intermediate 3-acetyl coumarin was prepared ecofriendly by using catalyst starch sulfuric acid (SSA) or cellulose sulfuric acid (CSA) in Pechmann condensation reaction from which, salicylaldehyde was condensed with ethyl acetoacetate. All the liberated products were used for assessing antibacterial activities against *E. coli*, *S. aureus*, *P. aeruginosa* and *S. pyogenes*. Thereafter, the tested compound **15** containing an electron deactivating group fluoro substituted at the ortho and para position of phenyl ring and had good antibacterial activities against *E. coli* and *S. aureus* at MIC value 50 µg/mL each (Vekariya et al., 2017).

2.51. Synthesis of coumarin fused furan derivatives

Two series of the 3-aryl- furo[3,2-*c*] coumarins derivatives **4a-4l** were prepared by microwave assisted synthetic method

with two different conditions; in the first method, the synthesis of these derivatives were carried out by the reaction of substituted 4- hydroxycoumarin **1a-1d** with an appropriate 2-aryl-1-nitro ethylene **2a-2c** in methanol by the principle of Nef reaction whereas other methods. The reaction of substituted 4- hydroxycoumarin **1a-1d** with an appropriate 2-aryl methyl bromide **3a-3c** in acetic acid and ammonium acetate under the Feist-Benary reaction condition. Among these two methods, in method of feist benary condition, furan-coumarin derived products had good yields. SARs of these derivatives attributed that the compounds with fused furo [3,2-*c*] coumarin skeleton had inherent antibacterial action in addition to substituted electron donating groups like methoxy; methyl was attached to phenyl at para position that might have increased the potency of inhibition growth of bacteria. The compound 8-chloro-3-(*p*-tolyl)-4*H*-furo[3,2-*c*] chromen-4-one **4 h** was reported having a good antibacterial agent action *E. coli* at MIC value 25 mm (Kanerla et al., 2017).

2.52. Synthesis of bis 3,3-coumarin derivatives

In the scheme, a series of bis coumarin derivatives **3a-3k** were synthesized by the reaction of one pot convenient with multiple component through chichibabin reaction. These derivatives were synthesized by the reaction of 3-acetyl coumarin or bromo substituted 3-acetyl coumarin **1a-1b** with substituted aryl aldehyde **2a-2k** and ammonium acetate under acidic conditions. During this synthesis initially involves the formation of imine by 3-acetyl coumarin is treated with ammonia. Then condensation was performed in between substituted aryl aldehyde and enolic state of acetyl coumarin and finally coumarin chalcones and imines cyclisation to yield 3,3'-(4-subst. phenylpyridine-2,6-diyl)substituted bis(2*H*-chromen-2-one). The antibacterial activity revealed that the compounds **3c** and **3d** were reported as having good antibacterial activities against *P. aeruginosa*, *B. subtilis*, *E. coli* at MIC values 15 µg/mL in comparison to Amoxicillin and Gentamicin as standards (Kenchappa et al., 2017).

2.53 Synthesis of coumarinyloxy bearing oxazoline derivatives

In this scheme, a series of coumarinyloxy bearing 3-substituted phenyl/pyrrolyl oxazoline **3a-3f** compounds were synthesized from the starting precursor 4-hydroxycoumarin, then 4-hydroxy coumarin was converted into 4-allyloxy coumarin derivatives **2a-2d** by the condensation with appropriate allylic halides in the presence of anhydrous acetone and potassium carbonate for 20 h. Furthermore, the obtained intermediate 4-allyloxy coumarin was reacted with (*Z*)-*N*-hydroxybenzimidoyl chloride in the presence of anhydrous toluene by 1,3-dipolar cycloaddition, which yielded coumarinyloxy bearing isoxazolone derivatives. Introduction of the substituted phenyl or pyrrole ring at C-3 position of desired target were obtained as the isoxazolone derivative. Among all the tested compounds the analogue 4-((3-(1*H*-pyrrol-2-yl)-4,5-dihydroisoxazole-5-yl)methoxy)-2*H*-chromen-2-one **3f** had shown good antibacterial action against *E. faecalis* at the MIC value 0.31 mg/mL in comparison to Gentamicin as the standard (Zghab et al., 2017).

2.54. Synthesis of silver complexes with imidazolium methyl coumarin derivatives

In this scheme, a series of silver complexes of substituted coumarin tethered NHC carbene ligands were synthesized and their biological actions were explored. *N,N'*-Dialkylated (imidazole or benzimidazole) referred as azolium salts as NHC precursors, initially in an electrophilic substitution reaction between imidazole/benzimidazole and *n*-bromo alkyl ether in the presence of dimethyl sulfoxide and excess potassium hydroxide to give *n*-alkyl imidazole/benzimidazole, which further with ether functionalized derivatives undergone alkylation by treated with 4-bromomethyl-6-substituted coumarin in 1,4-dioxane at 24 h refluxing liberated corresponding *N*-alkoxy-*N'*-substituted coumarin azolium bromide salts. These bromides salt derivatives had been converted into the corresponding hexafluorophosphate salts by the treatment with methanol and water (1:9v/v) solution of hexafluorophosphate. Finally, the ether substituted azolium hexafluorophosphate salt was treated with silver oxide in the presence of acetonitrile at 45 °C for 24 h under the absence of light and the obtained desired silver complex substituted coumarin tethered bis azolium hexafluorophosphate salts and this reaction was performed by *in situ* deprotonation of azolium salt. The silver complex of 6-methyl coumarin bearing bis imidazolium hexafluorophosphate salt **9** had reported as good antibacterial agent against *E. coli* at MIC at 08 µg/mL in comparison to Ampicillin as the standard (Achar et al., 2018).

2.55. Synthesis of 7-coumarinyl methyl theophylline derivatives

A new series of hybrid compounds containing xanthine derivative like theophylline **3a-3j** with substituted 4-bromomethyl coumarin were synthesized. The derivatives of 1,3-dimethyl-9-((substituted 2-oxo-2*H*-chromen-4-yl)methyl)-1*H*-purine-2,6 (3*H*,9*H*)-dione **3a-3j** were synthesized by the mixture of theophylline and anhydrous potassium carbonate with substituted 4-bromomethyl coumarin in acetone for 6–8 h. The SARs of antibacterial action of coumarin-theophylline hybrid congeners revealed that the derivative, which had electron donating substitutions at coumarin nucleus possesses a good inhibition of growth and 6-methyl coumarin-theophylline congener had the maximum inhibitory action with MIC at 3.9 µg/mL against Gram + bacterium *S. aureus*, where as another derivative 6-methoxy coumarin-theophylline hybrid had MIC at 7.8 µg/mL. The compounds bearing electron donating substituent at C-6 position decrease order of efficacy methyl > *t*. butyl > methoxy > 5,6-benzo for different bacterial strains. Moreover, the compound 1,3-dimethyl-9-((6-methyl-2-oxo-2*H*-chromen-4-yl)methyl)-1*H*-purine-2,6 (3*H*,9*H*)-dione **3a** was reported as having good antibacterial action against *S. aureus* and *E. coli* at MIC 3.9 µg/mL. Similarly, the compound **3f** was recorded as good antibacterial agent against *S. aureus* and *E. coli*, *S. typhi* at MIC 7.8 µg/mL in comparison to Tetracycline as the standard (Mangasuli et al., 2018).

2.56. Synthesis of ribofuranosyl-coumarinyloxy bearing 1,2,3-triazole derivatives

A series of substituted ribofuranosyl coumarinyl 1,2,3-triazole **4a-4d** had been synthesized by cycloaddition reaction between

azido sugar and 7-alkynated 4-methyl coumarin **2a-2d** in presence of Cu(I) with good yields. During synthesis of these compounds, initially with an intermediate 7-hydroxy substituted coumarin **1a-1d** were treated with propargyl bromide in the presence of potassium carbonate to produce corresponding 7-propargyloxy substituted coumarin in an 85% yield. Then after, 7-propargyloxy coumarin was reacted with the corresponding 2-azido-2,3,5-tribenzoyl-β-D-ribofuranose in the presence of ascorbate-CuSO₄ in THF through Cu(I) mediated cycloaddition reaction to afford resultant *N'*-2,3,5-tribenzoyloxy β-D-ribofuranosyl-4-coumarinyl-7-oxymethyl-1,2,3-triazole in 70% yield. Then debenzoylation of the resulted targeted ribofuranosyl coumarinyl 1,2,3-triazole derivatives. The compound *N'*-2,3,5-tribenzoyloxy β-D-ribofuranosyl-4-methylcoumarinyl-7-oxymethyl-1,2,3-triazole and *N'*-2,3,5-tribenzoyloxy β-D-ribofuranosyl-4-coumarinyl-7-oxymethyl-1,2,3-triazole had been reported having a good inhibitory action against *M. tuberculosis* at MIC 5.1 µM (Srivastava et al., 2018).

2.57. Synthesis of disubstituted of chromane derivatives

A series of disubstituted chromane derivatives were designed and synthesized from an intermediate either with 6-acetyl-5,7-dihydroxy-2,2-dimethylchroman-4-one **1** or 8-acetyl-5,7-dihydroxy-2,2-dimethylchroman-4-one **1a**. These intermediates underwent methylation with methyl iodide to obtain corresponding 6-acetyl-5,7-dimethoxy-2,2-dimethylchroman-4-one **2** and 8-acetyl-5,7-dimethoxy-2,2-dimethylchroman-4-one **2a**, respectively. Then, reduction of carbonyl of pyrone system of **2** and **2a** in the presence of sodium tetraborohydride and methanol in mild condition afforded 1-(4-hydroxy-5,7-dimethoxy-2,2-dimethylchroman-6-yl)ethanone **3** and 1-(4-hydroxy-5,7-dimethoxy-2,2-dimethylchroman-8-yl)ethanone **3a**, respectively. Finally the desired compounds 1-(5,7-dimethoxy-2,2-dimethyl-2*H*-chromen-6-yl)ethanone **4** and 1-(5,7-dimethoxy-2,2-dimethyl-2*H*-chromen-8-yl)ethanone **4a** were obtained by dehydration of immediate precursor **3** and **3a**, respectively using MDS/TEA/MSC. The compound 8-acetyl-7-methoxy-2,2-dimethyl-2*H*-chromen-5-yl trifluoromethanesulfonate was reported as a good antibacterial agent against *K. pneumoniae*, *E. coli*, *P. aeruginosa* and *S. aureus* at MIC 0.64, 0.64, 0.84, 0.02 µgml⁻¹, respectively in comparison to Novobiocin as the standard (Ponussamy et al., 2018).

2.58. Synthesis of coumacine derivatives

A group of bicyclo twelve-membered heterocyclic coumacine rings **4a-4c** derived from substituted coumarin **1a-1c** was synthesized by convenient methods. Initially, substituted coumarin **1a-1c** were prepared through Pechmann condensation between the phenolic derivative and ethyl acetoacetate in the presence of concentrated sulfuric acid, then after reduction of corresponding substituted coumarin in the presence of Lithium aluminum tetrahydride (LAH) in ether afforded (*E*)-2-(4-hydroxybut-2-en-2-yl)-substituted phenol **2a-2c** by opened coumarin ring system. The diol derivatives were converted to their disodium salts having reacted with sodium hydroxide, which further on treatment with diiodomethane (CH₂I₂) in anhydrous condition coumacine derivatives were obtained. Finally, the reaction was performed by the reactant methylene diiodide and nucleophilicity of phenoxide with

unsaturated alcoholic group. The coumacine-1 had notable action against *E. coli*, *K. pneumoniae*, *P. aeruginosa* at MIC 5 µg/mL individually, in comparison to standard drug Ciprofloxacin (Mustafa, 2018).

2.59. Synthesis of pyrazole-anilino connected coumarin derivatives

A series of compounds with bearing triazolyl methyl aniline-linked 4-hydroxycoumarin **4a-4u** were synthesized by the reaction of corresponding alcoholic solution of pyrazole carboxaldehyde **2a-2f** and substituted aniline and 4-hydroxycoumarin **3**. Then the obtain desired 3-((1,3-diphenyl-1*H*-pyrazol-4-yl)(phenylamino)methyl)-4-hydroxy-2*H*-chromen-2-one derivatives **4a-4u**. An intermediate pyrazole carboxaldehyde were synthesized by the reaction of alcoholic solution with the substituted acetophenone and phenylhydrazine with added of phosphorous oxy trichloride. In the synthesis free solvent and catalyst chosen during the reaction. The plausible mechanism of synthetic series of coumarin pyrazolyl linked aniline derivatives was initially aniline subtract and pyrazole carbaldehyde condensed and resulting in the formation of imine. In SARs reveal that compounds **4b**, **4c**, **4h**, **4i** and **4k** with electron donating substituent methoxy and methyl, which attached at anilino side chain in structure have been evident antibacterial activity with compared to electron withdrawing substituent chloro, nitro so on in aniline ring. The compounds **4b**, **4c**, **4h**, **4i** and **4k** had been reported as good antibacterial agent against, *M. luteus*, *S. aureus*, *P. aeruginosa* at MIC values ranging between 1.9 and 7.8 µg/mL in comparison to the standard drug Novobiocin (Kovvuri et al., 2018).

2.60. Synthesis of isatin- linked 1,2,3-triazole with coumarin

A series of newly synthesized compounds 1, 2, 3-triazole tethered indolinone-coumarin hybrids **7a-7l** from two reactant substrates such as, 4-methyl 7-(prop-2-ynyloxy) coumarin **6** and azidoethyl substituted isatin **3a-3d**. In the synthesis the corresponding hybrids, initially 7-hydroxy 4-methyl coumarin **5** was prepared by Pechmann condensation of ethyl acetoacetate and resorcinol **4** in acidifying agent; and the compound **4**, being heated with 7-hydroxy 4-methyl coumarin and propargyl bromide in the presence of potassium carbonate at 50 °C. Another intermediate 1-(2-azidoethyl)indoline-2,3-dione was prepared by the reaction between isatin and 1,2-dibromoethane in the presence of potassium carbonate yield N-(2-bromoethyl isatin). Consequently, this product reacted with sodium azide at 60 °C to afforded 1-(2-azidoethyl)indoline-2,3-dione. These two precursors were further used for the synthesis of the triazole tethered coumarin isatin hybrids through copper(I)-promoted alkyne-azide cyclo addition in the presence of DMF and copper acetate. Finally the desired products were condensed with appropriate amine hydrochloride to obtain respective (Z)-3-(methoxyimino)-1-(2-(4-(((4-methyl-2-oxo-2*H*-chromen-7-yl)oxy)methyl)-1*H*-1,2,3-triazol-1-yl)ethyl)indolin-2-one derivatives **7e-7l**. The compound 5-chloro-1-(2-(4-(((4-methyl-2-oxo-2*H*-chromen-7-yl)oxy)methyl)-1*H*-1,2,3-triazol-1-yl)ethyl)indoline-2,3-dione **7c** and (Z)-3-(methoxyimino)-1-(2-(4-(((4-methyl-2-oxo-2*H*-chromen-7-yl)oxy)methyl)-1*H*-1,2,3-triazol-1-yl)ethyl)indolin-2-one **7h** had exhibited as a good antibacterial agent against *M. smegmatis*

at MIC value 50 µg/mL in comparison to standard drugs, Rifampicin and Isoniazid (Liu et al., 2018a, 2018b).

2.61. Synthesis of N- benzimidazolyl oxime of acetyl coumarin derivatives

A series of N- benzimidazolyl acetyl coumarin oxime derivatives **5a-5k** were designed and synthesized from precursor 3-acetyl coumarin **1**. Initially 3-acetyl coumarin **1** was brominated in dry chloroform afforded 3-bromomethyl coumarin **2**. Furthermore, nucleophilic substitution on compound **2** reacted with benzimidazole in acetonitrile obtained 3-(2-(1*H*-benzo[d]imidazol-1-yl)acetyl)-2*H*-chromen-2-one **3**. Consequently, condensation of hydroxylamine hydrochloride in ethanol on the compound **3** produced the corresponding oxime, **4**. Finally, the desired oxime ethereal derivatives **5a-5k** were synthesised by the esterification on compound **4** with various substituted benzyl bromide in DMSO and tertiary-butoxide. The compound (Z)-3-(2-(1*H*-benzo[d]imidazol-1-yl)-1-(((4-chlorobenzyl)oxy)imino)ethyl)-2*H*-chromen-2-one **5a** and (Z)-3-(2-(1*H*-benzo[d]imidazol-1-yl)-1-(((4-bromobenzyl)oxy)imino)ethyl)-2*H*-chromen-2-one **5c** had exhibited as good antibacterial agent against *B. subtilis* at the value, MIC 0.95 and 6.25 µg mL⁻¹ (Singh et al., 2017).

2.62. Synthesis of 3,4-benzocoumarin linked coumarin derivatives

A series of pyrano[3,2-c] coumarin derivatives **5a-5c** had been synthesized by using multiple component reactions. Initially, the precursor 2-((2-oxo-2*H*-chromen-4-yl)methylene)malononitrile was achieved by multiple condensations of 4-formyl coumarin **1**, substituted malonyl nitrile and 4-hydroxy coumarin in one-step component reactions. In the reaction, the numbers of catalysts, sodium carbonate, potassium carbonate, sodium benzoate, sodium bisulfite and some amino acids, as L-proline, cysteine used in multiple component reaction approaches for enhancing productivity of the desired compounds. As L-proline being reported as a notable organic green catalyst due to duality nature, as proton -donating and -accepting, the mechanism of synthetic reaction was involved the principle of Knoevenagel condensation of substituted 4-formyl coumarin and ethyl cyanoacetate/malononitrile to form corresponding intermediate alkene, which further underwent Michael nucleophilic addition of 4-hydroxycoumarin, followed by intramolecular cyclisation to liberate the corresponding 2-amino-5-oxo-4-(2-oxo-2*H*-chromen-4-yl)-4,5-dihydropyrano[3,2-c]chromene-3-carbonitrile derivatives **2a-2c**. The compound 2-amino-4-(6-chloro-2-oxo-2*H*-chromen-4-yl)-5-oxo-4,5-dihydropyrano [3,2-c]chromene-3-carbonitrile **2c**, 4-(6-chloro-2-oxo-2*H*-chromen-4-yl)-3,4-dihydropyrano[3,2-c]chromene-2,5-dione **5c** and 4-(7-methyl-2-oxo-2*H*-chromen-4-yl)-3,4-dihydropyrano[3,2-c]chromene-2,5-dione **5d** were had shown good antibacterial activity against *E. faecalis* at MIC 3.25 µg/mL, individually in comparison to the standard drug Ciprofloxacin (Chougala et al., 2018).

2.63. Synthesis of 3-thiazolylamino coumarin derivatives

A series of 3-thiazolyl amino derivatives **2a-2o** was synthesised from substituted 3-acetylcoumarin **1a-1b**, initially acetyl cou-

marin was converted into 3-bromoacetyl coumarin by bromination on starting 3-acetyl coumarin in the presence of bromine in chloroform. Then after, by the principle 'Hantzsch's condensation', the substituted 3-bromoacetyl coumarin with either *N*-substituted or *N,N'*-disubstituted urea through cyclisation to yielded target 3-thiazolyl amino derivatives **2a-2o**. Among all the test compounds, the 3-(2-((3,4-dichlorophenyl)amino)thiazol-4-yl)-8-methoxy-coumarin **2g** had been a good antibacterial agent against *S. pneumonia* at MIC 75 μ M. Similarly, the compound 3-(2-((4-bromophenyl)amino)thiazol-4-yl)-8-methoxy-coumarin **2h** too exhibited a good antibacterial activity against *E. coli*, *P. aerogenes*, *S. typhi*, *S. aureus* at MIC 73 μ M, individually in comparison to standard drugs Streptomycin, Kanamycin and Vancomycin (Osman et al., 2018).

2.64. Synthesis of *N*-pyrazolyl coumarin-3-carboxamide derivatives

In this synthesis, coumarinyl linked pyrazole carboxamide derivatives **7a-7g** were synthesised from an intermediate 5-(2-hydroxybenzylidene)-2,2-dimethyl-1,3-dioxane-4,6-dione. The substituted compounds **5a-5f** were synthesised by the reaction of substituted salicylaldehyde with dimedone in the presence of piperidine acetate in ethanol. The obtained products were recrystallized in ethanol, then substituted coumarin 3-carboxylic acid **5a-5f** were liberated. Concomitantly, another intermediate ethyl 5-amino-1-substituted phenyl-1*H*-pyrazole-4-carboxylate derivatives **6a-6c** were prepared by the mixture containing 4-substituted phenyl hydrazine hydrochloride and ethyl 2-cyano ethoxyacrylate/ ethoxymethylenemalonitrile in ethanol. Finally, the derivatives of compounds **6a-6c** were converted into the corresponding pyrazole bearing carboxamides in the presence of phosphorus oxytrichloride and pyridine solution in mild condition. The obtained target compounds ethyl 5-(2-oxo-2*H*-chromene-3-carboxamido)-1-phenyl-1*H*-pyrazole-4-carboxylate derivatives had been recrystallized. Moreover, the compounds **7b** 5-(6-chlorocoumarin-3-yl-carboxamido)-1-(4-chlorophenyl)-1*H*-pyrazole-4-carboxylic acid and ethyl 1-(4-chlorophenyl)-5-(7-(diethylamino)-coumarin-3-yl-carboxamido)-1*H*-pyrazole-4-carboxylate **7e** had shown antibacterial action compared to ciprofloxacin against *E. coli*, and *Salmonella* sp. at MIC value 0.25 mg/L, individually in comparison to standard drugs and further exhibited potent inhibition with Topo II and Topo IV with IC₅₀ values, 9.4-25 mg/L. SAR studies of these derivatives indicated that compounds having pyrazole carboxamide moiety linked with *p*-chloro substituted coumarin in structural frame, which might be responsible for inhibitory action of both topoisomerase II and IV of *E. coli* and *Salmonella* sp. (Liu et al., 2018a, 2018b).

2.65. Synthesis of 1-alkyloxy imidazolyl coumarin derivatives

Thirty-nine derivatives of coumarin linked with imidazole and alkyloxy group had been synthesized. These derivatives were synthesized from the starting material 7-hydroxy coumarin and 7-hydroxy 4-methyl coumarin **1a**, which was prepared by heating of the solution 2,4-dihydroxy benzaldehyde and chloroethyl acetate with phosphonium ethyl acetate in ethanol, whereas another coumarin derivative **1b** was prepared by the

principle of Pechmann condensation reaction. Then, resorcinol was reacted with ethyl acetoacetate in sulfuric acid. Then compound **1** was reacted with the corresponding α,ω -dibromo alkanes and triethylamine in anhydrous acetone produced the corresponding 7-(3-bromo alkyloxy)-2*H*-chromen-2-one derivatives **2**. The compound **2** was treated with the substituted heterocyclic amines namely, triazole, imidazole, 2-methyl imidazole, piperazine, piperidine and pyrazole, and a few more in the presences of anhydrous acetone and acetonitrile and the obtained compound 7-(3-(1*H*-heteroaryl)alkyloxy)-2*H*-coumarin derivative **3**. In SAR study indicated that the long alkyl chain linker and substituted imidazole group would be responsible for antibacterial action and inhibitory activity against FabI and FabK. Among all these compounds, 7-((6-(2-methyl-1*H*-imidazol-1-yl)hexyl)oxy)-2*H*-chromen-2-one derivatives **3b** were reported as having good antibacterial activity against *F. cloumnae*, *S. agalactiae* and *S. aureus* at 8, 8, 64 μ M in comparison to standard drugs Enrofloxacin and Norfloxacin (Hu et al., 2018).

2.66. Synthesis of coumarin mannich based derivatives

A series of 3-methylated amino 4-hydroxy coumarin derivatives **4a-4m** was designed virtually and these derivatives were validated through computational tools. These derivatives were synthesized by the reflux condensation of 4-hydroxycoumarin, substituted aldehyde **2a-2m** and secondary amines such as, morpholine, piperidine, pyrrolidine, piperazine so on with the presence of DCM by the principle of Mannich condensation reaction. The compound 4-hydroxy-3-((4-hydroxy-3-methoxyphenyl)(morpholino)methyl)-coumarin **4l** had shown as a good antibacterial agent against *S. aureus* at MIC 12.50 μ g/mL (Sahoo et al., 2019c).

2.67. Synthesis of curcumin and isatin linked coumarin derivatives

Two series of compounds of triazole tethered mono carbonyl curcumin-coumarin **5a-5n** and isatin-coumarin **6a-6n** hybrids had been synthesised. Initially vanillin was reacted with acetone in the presence of base potassium at 25 °C and the obtained compound was (*Z*)-4-(4-hydroxy-3-methoxyphenyl)but-3-en-2-one **2a**. Furthermore, these obtained products were reacted with propargyl bromide to produce (*Z*)-4-(3-methoxy-4-(prop-2-yn-1-yloxy)phenyl)but-3-en-2-one **3a**. Then the substituted arylaldehyde was reacted with **3a** in the presence of dilute NaOH in methanol, which yielded monocarbonyl of curcumin **4a-4n**. In another intermediate reactant, by the treatment of various appropriate 1,2- dibromo alkanes with 4-hydroxy coumarin **1** in the presence of potassium carbonate in DMF could produce 4-(2-bromoalkyloxy)-2*H*-chromen-2-one **2a-2d**, which further treated with sodium azide at 25 °C in DMF to yield 4-(2-azidoalkoxy)-2*H*-chromen-2-one **3a-3d**. Simultaneously, isatin derivatives were subjected to treatment with 1,2-dibromoethane in the presence of potassium carbonate at 25 °C in DMF to obtain product; which on further treatment with sodium azide produce *N*-azidoethyl indolin-2,3-dione yielded the intermediate reactants such as, 4-(2-azidoethoxy)-coumarin **3a-3d** and *N*-azidoethyl isatin. These were reacted with various propargylated analogues of curcumin **4a-4n** in the presence of copper sulfate and sodium ascorbate

in DMF and the obtained desired target triazole was linked monocarboxyl curcumin-coumarin **5a-5n** and isatin-coumarin hybrid molecules. The SAR studies revealed that substitution bromo at C-5 position of isatin and 4-chloro substitution as isatin-curcumin of phenyl ring and 4-methoxy as coumarin-curcumin hybrids would suitable for inhibition of antibacterial action. The compound 5-bromo isatin linked with mono carbonyl curcumin through 1,2,3-triazole was reported as a good antibacterial agent against *E. coli* at MIC 6.25 µg/mL (Singh et al., 2019).

2.68. Synthesis of fluoro-quinolone based coumarin derivatives

A series of fluoroquinolone derivatives bearing substituted bicyclo or tricyclic ring at C-7 position was synthesized from an intermediate ethyl 6,7-difluoro-*N*-substituted -4-oxo-1,4-dihydroquinoline-3-carboxylate **1** through several step reactions. In the synthesis initially, the compound **1** was reacted with ethyl cyanoacetate or *t*-butyl cyanoacetate in the presence of potassium carbonate at 70 °C in DMF yield **2**. By the ester hydrolysis in acidic medium yield the corresponding fluoroquinolone such as, ethyl 7-(cyanomethyl)-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid derivatives **3**. The cyanomethyl was introduced at C-7 position of quinolone ring, then the condensed compounds **3** with 2-hydroxy benzaldehyde derivatives in the presence of catalytic piperidine and in DMF and produced (*Z*)-7-(1-cyano-2-(2-hydroxy-4-methoxy phenyl)vinyl)-*N*-ethyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid **4**. Then the compound **4** was treated 3–5% sulfuric acid at 110 °C and the obtained desired compound was 1-ethyl-6-fluoro-7-(substituted-with-2-oxo-2*H*-chromen-3-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid. In nucleophilic substitution of fluoro atom of fluoroquinolone to cyanomethyl at C-7 position acted as the source of nucleophiles and was formed as bicyclic ring system (coumarin or 1,4-dihydro benzoxepine ring) through intramolecular cyclisation. The obtained quinolone derivatives were evaluated against several strains of *Mycobacterium* sp. The compound 1-ethyl-6-fluoro-7-(8-ethoxy-2-oxo-2*H*-chromen-3-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid **7e** had promising antimycobacterial action against *M. tuberculosis* (H37Rv), *M. terrae* and *M. avium* at MIC doses, 0.7 and 1.5 µg/mL respectively (Charushin et al., 2018).

2.69. Synthesis of bis 1,2,3-triazolyl methoxy coumarin derivatives

A series of dimer compounds containing bis 1,2,3-triazolyl methoxy linked with 4-methyl-7-hydroxy coumarin derivatives under microwave irradiation methods was synthesized and the obtained products had antimycobacterial and antibacterial activities. Initially an intermediate 4-methyl-7-(prop-2-yn-1-yloxy)-2*H*-chromen-2-one **2** was prepared by a two step reaction, firstly 4-methyl-7-hydroxy coumarin derivative **1** was prepared by the Pechmann condensation of substituted resorcinol and ethyl acetoacetate in the presences of acidifying agent, then after the compound **1** was reacted with propargyl bromide in the presence of dry acetone and potassium carbonate for yielding **2**. In the synthesis of dimer of triazole-coumarin derivatives initially, nucleophilic substitution of dibromoalkane with sodium azide liberated azidoalkane and

coupled with 4-methyl-7-(prop-2-yn-1-yloxy)-2*H*-chromen-2-one in the presence of copper catalysed by 1,3-cycloaddition *via* azido-alkyne reactions. The title compounds were optimised by CuI in DMF:H₂O (1:3) under microwave irradiation at 180 W for 10 min. All the desired target molecules were screened for their antimycobacterial action using resazurin microtiter assay (REMA) in comparison to standard Rifampin and isoniazid (INH). Moreover, the compound 7,7'-(((1,1'-octane 1,8-diyl) bis(1*H*-1,2,3-triazole-4,1-diyl))bis(methylene))bis(oxy))bis(6-chloro-4-methyl-2*H*-chromen-2-one) **6j** had shown good antibacterial action against *B. subtilis*, *S. aureus*, *E. coli* at MIC doses 3.125 µg/mL for each; and the compound 7,7'-(((1,1'-methylenebis(1*H*-1,2,3-triazole-4,1-diyl))bis(methylene))bis(oxy))bis(4-methyl-2*H*-chromen-2-one) **6e** had *in vitro* control over *B. subtilis*, *S. aureus* and *P. vulgaris* at MIC doses 6.25 µg/mL. Consequently, compounds **6i** and **6j** had excellent antimycobacterial action with MIC 1.56 µg/mL. In SAR studies, it was known that electron-negative chlorine substituted coumarin at C-6 position and longer lipophilic alkyl chain linker between two ring systems play an important role for significant antibacterial action (Ashok et al., 2018).

2.70. Synthesis of palladium complexes coumarin derivatives

A series of palladium (Pd) complexes coumarin based tryptophan **3a** and methionine **3b** were synthesised from 4-hydroxy-3-acetyl coumarin. In the reaction, initially prepared corresponding enamine ligands (*E*)-methyl 2-((1-(4-hydroxy-2-oxo-2*H*-chromen-3-yl)ethylidene)amino)-4-(methylthio)butanoate, (*E*)-ethyl 2-((1-(4-hydroxy-2-oxo-2*H*-chromen-3-yl)ethylidene)amino)-3-(1*H*-indol-3-yl)propanoate by the reaction of 4-hydroxy-3-acetyl coumarin with methionine methyl ester hydrochloride and tryptophan methyl ester in the presence of triethylamine and methanol solution respectively. Further these Pd complexes were processed by the reaction of aqueous solution two ligands **3** and **3a** with potassium tetrachloride palladate in methanol and the mixture was stirred for 3hr. The ligand acted as a tridentate co-ordinated by one hydroxyl oxygen atom and the carbonyl oxygen of ester coumarin nucleus and nitrogen of enamine **3b**. The compound **3b** was reported as good antibacterial activity against *S. aureus*, *P. aeruginosa*, *E. faecalis* at MIC 208, 166, 208 µg/mL respectively in comparison to standard drug Ceftriaxone, Vancomycin (Stojković et al., 2018).

2.71. Synthesis of coumarinyl pyrimidinone derivatives

A series of coumarinyl linked with 1,6-dihydro pyrimidine carboxamide derivatives had been synthesized as intermediate reactants substituted 4-bromomethyl coumarin **1a-1g** and 2-mercapto-6-oxo-4-phenyl-1,6-dihydropyrimidine-5-carbonitrile **2a-2b**. The intermediate compounds **1a-1g** were prepared by Pechmann cyclisation of phenol and bromoethyl acetoacetate, whereas two compound **2a-2b** were prepared by a mixture of equimolar quantities of ethyl cyanoacetate, thiourea and aromatic aldehyde. Thenafter, condensation of substituted 4-bromomethyl coumarin and 2-mercapto-6-oxo-4-substituted phenyl-1,6-dihydropyrimidine-5-carbonitrile in anhydrous potassium carbonate using as acetone solvent to obtain 6-oxo-2-(((2-oxo-2*H*-chromen-4-yl)methyl)thio)-4-substituted phenyl-1,6-dihydropyrimidine-5-carbonitrile derivatives **3a-3j**.

Finally, these dihydropyrimidine carbonitrile derivatives were converted the corresponding *N*-(*tert*-butyl)-6-oxo-2-(((2-oxo-2*H*-chromen-4-yl)methyl)thio)-4-phenyl-1,6-dihydropyrimidine-5-carboxamide **4a-4j** by the compounds **3a-3j**, which reacted with *tert*-butyl acetate in the presence of sulfuric acid and acetic acid. Then, the desired coumarinyl-pyrimidine carboxamide was carried out using both conventional and microwave irradiation yield 55% and 82 to 90% respectively. The compound **4f** had shown good antibacterial agent against *S. aureus* at MIC 2.5 µg/mL and comparison to standard drug as Ciprofloxacin. In SARs revealed that the electron donating substituents in coumarin ring had been reported as positive effect to antibacterial activity efficiency order of substitutions methyl > tri-butyl > methoxy, 5,6-benzo **4f** attached at C-6 coumarin had a good inhibitory action against *S. aureus* (Chavan et al., 2018).

2.72. Synthesis of triazole substituted coumarin derivatives

1,2,3-triazole derivatives were prepared by the Cu(I) ions catalysed [2 + 3] cycloaddition reaction between organic azides and terminal alkynes at an ambient temperature. In this synthesis, a series of triazolyl bearing coumarin derivatives **6a-6p** were performed *via.*, azide-alkyne cycloaddition reaction. The substituted 4-azidomethyl coumarins were synthesized by two step reaction, substituted 4-bromomethyl coumarin were prepared by Pechmann cyclisation of bromoethyl acetoacetate and the substituted phenol in acidifying agent. Then the obtained products were reacted with sodium azide in aqueous. Additionally, an intermediate compounds 4-ethynyl-1-substituted phenyl-1*H*-1,2,4-triazol-5(4*H*)-one **5a-5b** were prepared by the reaction of 1-substituted phenyl-1*H*-1,2,4-triazol-5(4*H*)-one with prop-1-yne using potassium carbonate in anhydrous acetone solution. This reaction was followed by azide-alkyne cycloaddition of ethynyl-1-substituted phenyl-1*H*-1,2,4-triazol-5(4*H*)-one and substituted 4-ethyl azido coumarin **4a-4h** in presence of copper ascorbate in THF/water 1:1 for yield of 4-(((1-(2-oxo-2*H*-chromen-4-yl)methyl)-1*H*-1,2,3-triazol-4-yl)methyl)-1-substituted phenyl-1*H*-1,2,4-triazol-5(4*H*)-one **6a-6p** (coumarinyl-1,2,3-triazolyl-1,2,4-triazolone) and recrystallized from suitable solvents. SARs of these derivatives indicated that the presence of electron donating groups in coumarin ring and phenyl attached 1,2,4-triazole compounds were enhanced the antimycobacterial action against *M. tuberculosis*. The compounds 4-(((1-(6-methyl-2-oxo-2*H*-chromen-4-yl)methyl)-1*H*-1,2,3-triazol-4-yl)methyl)-1-phenyl-1*H*-1,2,4-triazol-5(4*H*)-one **6e** was recorded as a good antimycobacterial agent against *M. tuberculosis* at MIC value 1.60 µg/ml individually in comparison to standard drug Pyrazinamide (Somagond et al., 2019).

2.73. Synthesis of coumarinyloxy derivatives

In this scheme, coumarinyloxy derivatives **2a-2d** were synthesized by undertaking O alkylation of 6-methyl-4-hydroxy coumarin **1** under phase transverse catalyse (PTC) reaction. PTC coumarin undergone alkylation by the treatment with alkyl halides (allyl bromide, benzyl chloride, 4-nitrobenzyl chloride, ethyl chloro acetate) using base medium as potassium carbonate and tertiary butyl ammonium chloride afforded corresponding oxygen alkylated at C-4 position through

nucleophilic displacement. Likewise, PTC reaction of coumarin derivatives with phenyl isothiocyanate in equimolar quantity through nucleophile addition on the carbon nitrogen double of phenyl isothiocyanate to produce 3-(*N*-phenyl) thiocarbamide coumarin. Similarly, coumarin was reacted with aromatic aldehyde like pipernal, anisaldehyde in equal amount in presence of piperidine as base, afforded corresponding 3-arylidene 6-methyl-4-hydroxy coumarin **3a-3b**. The compound **2d** was reported as a good antibacterial agent against *E. coli* at the MIC value 32 µg/mL (Regal et al., 2020).

2.74. Synthesis of isatin-triazolyl coumarin derivatives

A series of triazolyl linker isatin-coumarin hybrid molecules were synthesized. In this reaction, the substituted isatin **1a-1e** reacted with 1,2-dibromo alkanes using potassium carbonate as base DMF solvent then after resultant intermediate **2a-2e** was react with sodium azide in DMF produce 1-(4-azidoalkyl)-substituted isatin **3a-3e**. Another reactant 4-(prop-2ynyloxy)-coumarin **3g** was further reacted with various derivatives of 1-(4-azidoalkyl)-indolin-2,3-dione in presence of copper sulfate pentahydrate with sodium ascorbate in DMF solution to yield desired target candidates 1-(2-(4-(((2-oxo-2*H*-chromen-4-yl)oxy)methyl)-1*H*-1,2,3-triazol-1-yl)ethyl) substituted indoline-2,3-dione hybrids **4a-4u**. SARs of these desired derivatives indicate that electron density of the fifth position of isatin remarkable influence of antibacterial action and activity is directly proportional to increase the electronegativity on same position of isatin so order of potency substitution fluoro > chloro > bromo > iodo > nitro > methoxy > hydrogen and concerned for linker space carbon length $n = 1 > 2 > 3$. All the synthesized products were evaluated for their antibacterial potential against bacterial strains *E. coli*, *S. enteric*, *S. aureus*. The compound 1-(2-(4-(((2-oxo-2*H*-chromen-4-yl)oxy)methyl)-1*H*-1,2,3-triazol-1-yl)ethyl) indoline-2,3-dione **4a** and 1-(2-(4-(((2-oxo-2*H*-chromen-4-yl)oxy)methyl)-1*H*-1,2,3-triazol-1-yl)ethyl) 5-fluoro- indoline-2,3-dione **4b** had shown as good antibacterial action against *S. aureus* at the MIC value 30 and 312 µg/mL (Bhagat et al., 2019).

2.75. Synthesis of coumarin 3-carboxamide derivatives

A series of substituted coumarin 3-carboxamide derivatives **3a-3j** were synthesized by facile green synthetic methods. In this synthesis, the precursor 3-((4-bromo-2-fluorophenyl)amino)-3-oxopropanoic acid **1** was prepared by the condensation reaction of 4-bromo-2-fluoro aniline and diethyl malonate in presence ethanol and sodium carbonate. The obtained product **1** was mixed with substituted salicylaldehyde by addition of a few drops of piperidine and finally the reaction mixture was heated at 105 °C for 4 h. The obtained precipitate *N*-(4-bromo-2-fluorophenyl)-substituted-2-oxo-2*H*-chromene-3-carboxamide was recrystallize with acetone. The compounds *N*-(4-bromo-2-fluorophenyl)-6-nitro-2-oxo-2*H*-chromene-3-carboxamide **3c**, 6,8-dibromo-*N*-(4-bromo-2-fluorophenyl)-2-oxo-2*H*-chromene-3-carboxamide **3d**, 6,8-dichloro-*N*-(4-bromo-2-fluorophenyl)-2-oxo-2*H*-chromene-3-carboxamide **3e**, 6-chloro-*N*-(4-bromo-2-fluorophenyl)-8-nitro-2-oxo-2*H*-chromene-3-carboxamide **3f** and *N*-(4-bromo-2-fluorophenyl)-8-nitro-2-oxo-2*H*-chromene-3-carboxamide **3j** had been

shown growth inhibits against human pathogenic fungal species, *Candida albicans* at MIC 1000 µg/ml and comparison to standards Fluconazole and Ciprofloxacin (Khan et al., 2019).

2.76. Synthesis of 3-Aroyl substituted coumarin derivatives

Two series of Co(II), Cu(II), Zn(II) complexes of 3-(2-hydroxy benzoyl)-2H-chromen-2-one, 3-(3-hydroxy-2-naphthoyl)-2H-chromen-2-one were synthesized. These ligands were prepared by the reaction of equimolar alcoholic solution of 4-hydroxy coumarin and respective aldehyde such as, salicylaldehyde and 2-hydroxy naphthaldehyde in presence of triethylamine as catalyst quantity and refluxed for 1.5 h. The obtained desired ligands were recrystallized by methanol. Thereafter, these alcoholic solution ligands were reacted with corresponding metal chloride hydrated $MCl_2 \cdot H_2O$ in presence of ammonium hydroxide to produce corresponding complexes. ATB: Cefoxitin complex from ligand L1 was effective toward *S. aureus*, *E. coli*, *P. aeruginosa* at MIC value 30 µg/mL in comparison to the standard drug Cefoxitin (Belkhir-Talbi et al., 2019).

2.77. Synthesis of 4-triazolidin-thione coumarin derivatives

Two series of substituted coumarin containing 1,2,4-triazolidin-3-thione derivatives **3a-3j** were synthesised, by the formation of respective semithiocarbazonate in nucleophilic addition reaction of semithiocarbazonate to electron deficient carbon atom of carbonyl compound of substituted 4-formyl coumarin/benzaldehyde and followed by intramolecular nucleophilic attack of amine (NH_2) of thiosemicarbazonate to azomethine to liberate the desired target molecules substituted coumarin and phenyl triazolidin-2-thiones. SAR studies of these compounds revealed that substituted phenyl ring replaced by substituted coumarin triazolidin thiones enhanced the antitubercular activity, where as various mono substituted electron donating group like methoxy, methyl, attached either phenyl triazolidine thione or respective coumarin triazolidine thione derivatives had been reported as more potent than disubstituted system. Mono substitution of phenyl ring and coumarin bearing triazolidine thion had been shown a good antimycobacterial action, as the substituted coumarin triazolidin thione methyl, methoxy, 5,6-benzo and 7,8 benzo moderate increases the antibacterial action whereas in phenyl triazolothiones the substituted hydroxy, 5,6-benzo in the phenyl ring enhanced the significant antibacterial action. The compound 7-methoxy-4-(5-thioxo-1,2,4-triazolidin-3-yl)-2H-chromen-2-one **3d** was reported as good antibacterial agent against *B. subtilis* at MIC 0.8 µg/mL in comparison to the standard drug ciprofloxacin (Shaikh et al., 2019).

2.78. Synthesis of ruthenium complexes of 3-acetohydrazone coumarin

Coumarin hydrazone/hydrazone hybrids had been shown having potent anticancer and antibacterial actions. A new series Ruthenium(II)-DMSO complexes of substituted coumarin 3-acylhydrazone **4a-4d** were synthesised. In these complexes, the four novel substituted coumarin 3-acyl hydrazones derivatives ligands **3a-3d** had been synthesized by the reaction involv-

ing condensation the compound 6-diethyl amino-coumarin 3-carbohydrazone **1** with substituted benzaldehyde **2a-2d** in the presence of hydrochloric acid and ethanol. Thereafter these obtained ligands were further had been reacted with complexes $cis[RuCl_2(DMSO)_4]$ in presence of ethanol and the mixture was refluxed for 4 h and finally the obtained yellow precipitate complexes washed several times in cold ethanol. Among all the desired hydrazone ligand the compound, Ruthenium (II) (*E*)-*N'*-(4-bromo benzylidene)-6-(diethylamino)-2-oxo-2H-chromene-3-carbohydrazone **3c** in DMSO complex had been seen with notable antibacterial activity against *S. aureus* at MIC 40.5 µM (de Almeida et al., 2019).

2.79. Synthesis of bis triazole uracil based coumarin derivatives

Incorporation of 1,2,3-triazole ring in several drug designing strategies due to its better aromatic stabilization, good binding affinity, isostere of carboxylic group and resistant towards both oxidation and reduction in acidic and alkaline medium. 1,2,3-triazole and its derivatives had been shown with a wide range pharmacological actions. Thus, researchers had more attentions as 1,2,3-triazole ring is tethering agent in drug design. A series of compounds of bis coumarinyl alkyloxy 1,2,3-triazole linker with uracil hybrids **C1-28** were designed and synthesized. Antibacterial potentials of these obtained analogues were studied. These compounds were synthesized from 4-(2-azidoethoxy)-2H-chromen-2-one **1**. Initially the 4-hydroxy coumarin was dissolved DMSO and followed by addition of dibromoethane in presence of potassium carbonate to obtain alkylated coumarin, which further react with sodium azide in DMSO solution to give 4-(2-azidoethoxy)-2H-chromen-2-one **1**. Another reactant, 5-substituted-1,3-di(prop-2-yn-1-yl)pyrimidine-2,4(1*H*,3*H*)-dione **2a-2g** was prepared by the reaction of substituted uracil **1a-1g** with propargyl bromide in the presence of DMF solution as solvent and was used as potassium carbonate at a room temperature. Finally, 3-azidoalkylated coumarin were treated with propargylated uracil in the presence of copper sulfate and sodium ascorbate in DMF solution at room temperature to get the desired target analogues triazole tethered coumarin-uracil hybrids **C1-28**. SAR studies of these compounds indicated that the analogues containing substituted uracil were more potent with antibacterial actions than compounds without non-substituted uracil. Thus, compound poses electron withdrawing substituent had shown more inhibitory action, whereas potency decreased with increasing chain carbon length in between two nuclei. Among all the tested candidates, the compound bearing chloro uracil substituted triazolyl ethoxy coumarin **C-3** had reported as good antibacterial agent(s) against *E. faecalis*, *S. aureus*, *P. aeruginosa* and *E. coli* at MIC values, 7.23 µg/ml in comparison to the standard drug Levofloxacin (Sanduja et al., 2020).

2.80. Synthesis of schiffbase of 4-hydroxyl-3-acetyl coumarin derivatives

In this scheme, 4-hydroxy coumarin compound with the Schiff base enamine aminophenol **3** was synthesized by the reaction of 3-acetyl-4-hydroxy coumarin **1** with aminophenol **2** in methanol. The compound (*E*)-3-(1-((2-hydroxyphenyl)amino) ethylidene)chroman-2,4-dione **3** had been shown good antibacterial activities against *S. aureus*, *B. cereus*, *E. coli*, *K. pneumo-*

nia at MIC doses, 39, 78, 78, 78 mg dm⁻³, respectively in comparison to the standard drug Chloramphenicol (Avdović et al., 2019).

3. SARs of coumarin derivatives as antibacterial agents

The exploration of synthetic and semisynthetic coumarin derivatives against inhibitory actions of notorious Gram positive, negative and acid-fast mycobacteria are emphasised here. Evidently, more than twenty-five percent of developed molecules had been seen upstanding antibacterial action(s) and a few more had moderate to less efficacy. In the principle of medicinal chemistry synthetic strategies, molecular hybridization is an established etiquette for development of novel compounds. Indeed, the phytochemical coumarin is natural heterocyclic ring with various biological actions among all; the antibacterial action(s) is more predominant by the intermixing of various components (Fig. 3).

In this SAR study of coumarin had briefly emphasized on integument of active sites of the congener for properties of inhibitory actions. 2-(furan-2-ylmethyleneamino)-6-coumarinyl-4-substituted nicotinic nitriles **1** had been notable antibacterial inhibitory properties, due to the presence of electron donating substituents, -OCH₃, -CH₃ of phenyl ring and elec-

tron withdrawing NO₂, halogen groups respectively. Similarly, metal complexes bearing coumarinyl carbohydrazide with indole Schiffbase derivatives **2** had exhibited remarkable antibacterial activity due to the presence of withdrawing chloro substituents in complexes, which have better zone of inhibition than methylated ligand. Indeed, the increases antibacterial efficacy is directly proportionate to lipophilic character of metal chelate ions, which could favour to permeation by lipid layer of bacterial cell membrane. Moreover, substituted ribofuranosyl coumarinyl 1,2,3-triazole derivatives **3** had been reported as potent candidates against clinical isolates of MDR human pathogenic bacterial strains. The structure bearing ribosylfuranosyl 1,2,3-triazole nucleus connected to 4-methyl-7-hydroxycoumarin at C-7 position through oxy-methylene (-OCH₂) linker. Concomitantly, the coumarinyl linked pyrazole carboxamide derivatives **4** had been reported as good antibacterial bacterial agent as inhibitors of Topoisomerase II and Topoisomerase IV. On the *N*-(4-chloro phenyl) pyrazole 5-carboxamide **4** structure of coumarin at C-3 position, an attachment of diethyl amino or bromo may lead inhibitory effect on bacterial growth.

Furthermore, monocarbonylcurcumin-coumarin ring linker with 1,2,3-triazole nucleus through two carbon chain compounds due to the presence of 4-methoxy substitution at

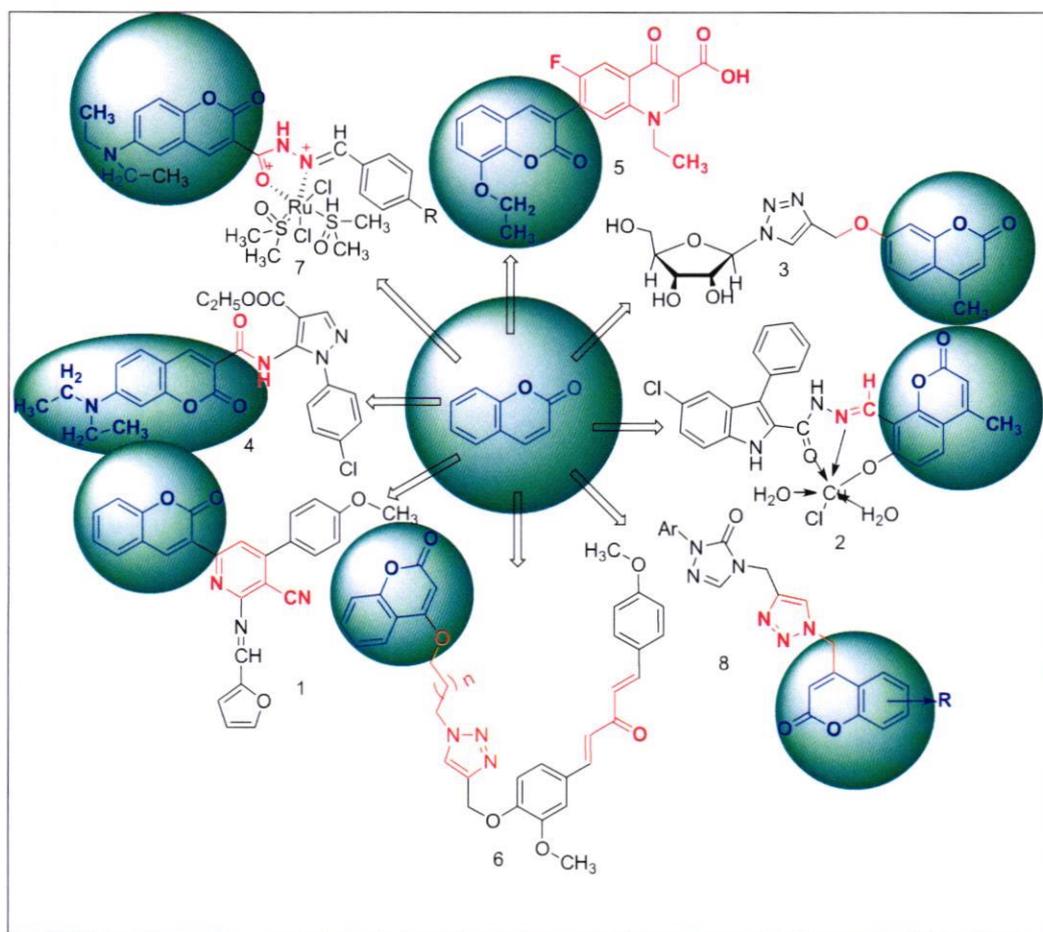


Fig. 3 Structural-activity-relationships of coumarin derivatives.

curcumin-coumarin hybrids **5** may have showed good antibacterial actions. Similarly, coumarin flouroquinolone hybrids **6** were reported as having good antimycobacterial actions due to flouro quinolone ring, which is essential for any antibacterial action. Thus, the developed molecules may have greater degrees of inhibitory actions on bacterial DNA gyrase or topoisomerase. Ruthenium(II)-DMSO complexes of substituted coumarin 3-acylhydrazone **7** had been reported from 7-diethylamino- coumarin hydrazide. These complexes had shown greater inhibitory action due to the presence of hydrazide group and metal ion Ruthenium(II) in structural frame. On the structure of compound **8**, where the presence of 1,2,3-triazolyl substituted coumarin ring which makes the molecules had exhibited significant antibacterial actions.

4. Conclusion

This phyto-compound coumarin, with its congeners would provide a frame for pharmacophore-based drug discovery against bacterial diseases. Herein, a comprehensive review of the various reaction strategies such as, Schiffbase, Azo-dye, Mannich-base, transitional metal complexes, Pechmann condensation and a few more synthetic principles for antibacterial activities are described. These are expected to be beneficial to control MDR bacterial pathogens in the rising demands of antibacterial candidates, from clinicians today. Indeed, these synthetic/semi-synthetic approaches of additions of newer phyto-based modified chemical entities with *in vitro* inhibitory actions against pathogenic microorganisms; particularly, against MRSA, mycobacteria and several other ghoulis infectious bacteria. The evolving of MDR bacterial strains have spiraled to unbridled notorious standards, due to the accumulation of multidrug resistance in them; surprisingly, one would hardly find a more vivid illustration of any commensal like, the Gram-positive *Staphylococcus aureus*, which is now the methicillin-resistant *S. aureus* (MRSA), transforming into a perilous MDR-MRSA with an armamentarium of multidrug resistance, Today 'MDR-MRSA' is regarded as the ghoulis superbug of the health domain! Thus, the necessity of some newer antibacterial agents to overcome the grievous resistance pattern of MRSA and other bacterial infective agent(s). Additionally, the SAR studies are the coveted corollary, as highlighted in detail. Further work is necessary to understand the various signalling unknown mechanisms with mode of administration and pharmacokinetics and dynamic properties in drug development cascades.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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